

The Free Internet Journal for Organic Chemistry

Paper

Archive for Organic Chemistry Arkivoc **2021**, part iv, 0-0 to be inserted by editorial office

Synthesis of a pyruvylated *N*-acetyl-β-D-mannosamine containing disaccharide repeating unit of a cell wall glycopolymer from *Paenibacillus alvei*

Simon Krauter, Christina Schäffer, and Paul Kosma

Department of Chemistry^a and Department of Nanobiotechnology^b, University of Natural Resources and Life Sciences, Muthgass 18, A-1190 Vienna, Austria Email: paul.kosma@boku.ac.at

Dedicated to Horst Kunz on the occasion of his 80th birthday

Received mm-dd-yyyy

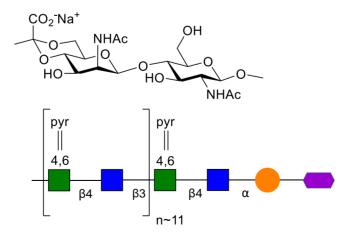
Accepted mm-dd-yyyy

Published on line mm-dd-yyyy

Dates to be inserted by editorial office

Abstract

A disaccharide implicated in anchoring of bacterial Surface-layer proteins to secondary cell wall glycopolymers has been prepared by glycosylation of a protected N-acetyl glucosamine acceptor with a glucopyranosyl donor to generate the β -(1 \rightarrow 4)-linkage. Subsequent inversion of the configuration and azide introduction at position 2 with triflic anhydride, however, led to formation of a tetrazole derivative. Alternatively, displacement of a 2-O-mesylate by sodium azide, reduction and N-acetylation enabled the conversion into the distal N-acetyl- β -D-mannosamine residue. Pyruvylation of the latter unit and global deprotection afforded the disaccharide repeating unit from Paenibacillus alvei as a ligand for crystallographic and binding studies.



Keywords: Secondary cell wall polymer, pyruvate, glycosylation, surface layer protein, tetrazole

DOI: https://doi.org/10.24820/ark.5550190.p011.358 Page 1 [©]AUTHOR(S)

Introduction

Pyruvic acid ketal substitutents attached to diol systems bridging positions 2,3; 3,4 as well as 4,6 on various glycans have increasingly been detected in the past few years. Due to their negative charge, pyruvyl groups may exert a number of biologically relevant interactions such as contributing to antigenic properties in bacterial polysaccharides such as Streptococcus pneumoniae serotype 4, Klebsiella pneumoniae, Acinetobacter baumannii and Bacteroides fragilis as examples.²⁻⁷ In addition, in a few Gram-positive bacteria, pyruvylation of so-called non-classical secondary cell wall polymers (SCWP) has been implicated to exert non-covalent interactions with surface (S-) layer proteins. S-layer proteins self-assemble into two-dimensional crystalline lattices that completely cover the bacterial surface. A terminal 4,6-O-Pyr-β-D-ManNAc residue in the SCWP of Bacillus anthracis has been described to serve as the main mechanism to anchor the S-layer via N-terminal homology (SLH) domain. 10-12 A similar role may be relevant for SCWP-SLH interactions in Bacillus cereus. 13 We have been interested in studying the underlying molecular details of SCWP-SLH binding using the SCWP from Paenibacillus alvei as model system. This SCWP contains \rightarrow 3)-4,6-*O*-Pyr-β-D-ManNAc-(1→4)-β-D-GlcNAc-(1→ repeating units covalently attached to N-acetylmuramic residues of peptidoglycan. ¹⁴ A similar backbone structure has been found in an SCWP from Lysinibacillus sphaericus CCM 2177, albeit with only every second β-D-ManNAc unit being substituted by a pyruvyl ketal, probably due to hydrolysis occurring upon isolation from the peptidoglycan using hydrofluoric acid. 15 Recently we could show by using X-ray crystallography of a truncated variant of the P. alvei S-layer protein SpaA_{SLH} - representing the three consecutive SLH domains liganded to the monosaccharide 4,6-O-Pyr- β -D-ManNAcOMe and the disaccharide β -D-GlcNAc- $(1\rightarrow 3)$ -4,6-O-Pyr-β-D-ManNAcOMe, that the 4,6(S)-O-pyruvylated N-acetyl-mannosamine residue was the main contributor to binding. ¹⁶ Main binding features included a π-stacking interaction of ManNAc with tryptophan 151 and ionic interactions of the pyruvyl carboxylic group with arginine 61. It should be noted that the synthetic mono- and disaccharide ligands were bound in two mutually exclusive different binding grooves of the protein allowing for dynamic properties during cell growth and division. Isothermal microcalorimetry data, however, indicated that the 4,6-O-Pyr- β -D-ManNAcOMe ligand was efficiently bound in solution ($K_D \sim 29$ nM), while the disaccharide fragment did not show any binding. The crystal structure of the complex with the latter ligand with a terminal N-acetylglucosamine residue also revealed disorder of the N-acetylglucosamine moiety and absence of any hydrogen bonding to the protein surface. In order to provide more insight into these interactions and to clarify the role of an "internal" N-acetyl glucosamine unit we have, thus, set out to prepare the alternate disaccharide mimicking the native situation in the SCWP repeat for crystallographic and binding studies. Herein we describe the synthesis of that disaccharide ligand and report on an unusual side reaction encountered during azide introduction of a triflate intermediate leading to formation of a tetrazole derivative.

Results and Discussion

The synthesis of the alternately connected repeating unit of the SCWP from P.~alvei required the challenging 1,2-cis-connection of the ManNAc residue to position 4 of a suitable GlcNAc acceptor derivative. Previously, a 4,6-O-pyruvylated 2-azido-2-deoxy-mannopyranosyl phenylthioglycoside donor had been used for the assembly of a terminal trisaccharide fragment of the SCWP from B.~anthracis, which, however, also led to formation of the unwanted α -anomer. Reactions of 4,6-O-benzylidene protected 2-azido-2-deoxy-mannopyranosyl diphenyl phosphate as well as related phenylthioglycoside donors, however, were reported to proceed in high β -selectivity and in good yields. Thus, we opted to use a 4,6-O-benzylidene group as a

Page 2 [©]AUTHOR(S)

temporary protecting group for a late stage introduction of the 4,6-O-pyruvyl ketal and follow the established route for β -mannosides by inverting the configuration of a *gluco*-precursor into the 2-deoxy-2-azido-*manno* derivative. ¹⁹⁻²²

The synthesis of the *gluco*-configured imidate donor - developed by R. Schmidt - commenced with the selective de-O-acetylation of the known (Ref 23) 3-O-benzyl substituted glucopyranose **1** to give the hemiacetal **2** in 76% yield (Scheme **1**). Compound **2** was then converted into the N-phenyltrifluoroacetimidate donor **3** in 82% yield by reaction with N-phenyl trifluoroacetimidoyl chloride/ K_2CO_3 in dichloromethane. As glycosyl acceptor, 3,6-di-O-benzyl methyl glycoside **4** was prepared according to literature. ^{24,25} The coupling step of **4** with donor **3** promoted by TMSO-triflate at room temperature led to complete consumption of the donor within **1** h and formation of a main product, which, however, was first identified as the orthoester intermediate. The structural assignment of the orthoester was based on the homonuclear coupling constant $J_{1',2'}$ (5 Hz), the upfield shifts of proton H-2' (4.46 ppm) and the orthoester methyl group (1.7 ppm), in agreement with literature data. ²⁶ In order to allow for orthoester rearrangement into the glycoside, the reaction time was then increased to 3 days, when TLC controls revealed complete disappearance of the orthoester and formation of disaccharide product **5**, isolated in a 1:1 mixture with acceptor **4** in 43% yield (based on the respective ¹H NMR integration values of **4** and **5**). Separation of the residual glycosyl acceptor **4** almost comigrating with disaccharide **5** - was attempted by de-O-acetylation with triethylamine.

Scheme 1. Synthesis of disaccharide **9** and attempted azide introduction via triflate activation

After 40 h reaction time, the disaccharide product was obtained as 3:1 mixture of the 2,4,6-triol 6 and diol 7 containing a 2-O-acetyl group. The latter ester group was surprisingly stable and would have needed forcing

Page 3 [©]AUTHOR(S)

conditions for complete removal. Hence, the mixture was carried out through the ensuing introduction of the 4,6-O-benzylidene group using benzaldehyde dimethylacetal and FeCl₃ as catalyst²⁷ to give a mixture of **8** and **9** followed by a subsequent Zemplén transesterification - again with prolonged reaction time - to eventually provide the alcohol **9** in an overall isolated yield of 65% (for 3 steps).

The introduction of the 2-azido group with inversion of configuration was first attempted via nucleophilic triflic anhydride in pyridine resulted in the formation of a highly polar spot as observed on a TLC plate. After work-up of the reaction mixture the crude residue was dissolved in DMF and treated with sodium azide at 70 °C for 2 h. Chromatography of the reaction mixture afforded a major applar product 10 (33%) and a poor yield of the expected azido derivative 11 (10 %). The NMR spectra of 10 showed a significant downfield proton shift of one methyl group (2.44 ppm), connected to an upfield shifted carbon at 8.8 ppm as observed in an HSQC experiment.²⁸ In addition, the N-H signal was absent, and a quaternary carbon was seen at 153.7 ppm, in a range expected for an N-C=N mojety. The upfield shifted methyl group showed an HMBC correlation to this quaternary carbon and a NOESY interaction with H-2 of the glucosamine residue. This data thus indicated the presence of a 1,2,3,4-tetrazole unit at position 2 of the glucosamine unit. The structure of 10 was eventually confirmed by HR-ESI-TOF MS data (m/z = 806.349) being consistent with the formula $C_{43}H_{47}N_7O_9$. Formation of the tetrazole ring may be rationalized by reaction of pyridine with trifluoromethanesulfonic anhydride to produce a cationic pyridinium ion intermediate A or a direct reaction of the acetamido group with triflic anhydride to produce an iminium triflate that is in equilibrium with the nitrilium ion B.29-31 A subsequent formal 3+2 cycloaddition with the azide anion would then generate the 2'-tetrazolyl product 10. The occurrence of a planar keteniminium intermediate C was considered to be less likely, since the glucoconfiguration of the reducing unit in product 10 remained unchanged throughout the reaction. A non-charged triflate imidate was also ruled out, since a highly polar product was observed by TLC upon activation of 9 with triflic anhydride and pyridine. In order to gain more insight into the amide activation, a model reaction was carried out in an in-situ NMR experiment using the per-O-acetylated N-acetyl-glucosamine 12 in a 2:1 mixture of deuterated dichloromethane / pyridine with 2 equivalents of triflic anhydride at 27 °C. The NMR spectra of the resulting product 13 formed within one hour only showed broad lines of non-deuterated pyridine signals at 8.65, 7.61 and 7.21 ppm for H-2/6, H-4 and H-3/5, respectively, indicating the presence of pyridiniumhydrotriflate.²⁹ In addition, ¹³C NMR signals showed only pyridine solvent signals at 149.11, 135.40 and 123.11 ppm for C-2/6, C-4 and C-3/5. Signals of pyridinium species were absent (see supporting information). Notably, a downfield ¹H NMR shift of one methyl group (2.88 ppm) connected to a carbon signal at 15.51 ppm and an HMBC correlation to a quaternary carbon signal at 158.66 ppm was observed. In addition, H-2 of the glucosamine unit also showed an HMBC correlation to the latter quaternary signal and an HSQC connectivity to an unusually downfield-shifted ¹³C NMR signal at 64.23 ppm, in glucosamine residue present in a 2acetimidoyl linkage had previously been observed. 32 Thus, the in-situ NMR experiment indicated the presence of the triflyl-imidate 13 as the reactive intermediate. Upon aqueous extraction - as used for the ensuing azidation reaction - the NMR spectrum of the crude mixture, however, indicated the presence of cationic pyridinium species as seen from characteristic ¹H and ¹³C NMR shifts of H-4/C-4 (8.66/148.82 ppm) and C-3/C-5 (8.15/128.01 ppm), respectively.

To the best of our knowledge a similar tetrazole carbohydrate byproduct has rarely been described in the literature, but this side reaction could be a reason for reduced yields of triflation/azidation reactions when involving sterically hindered alcohols in the presence of an acetamido sugar.^{33,34} On the other hand, this facile amide activation would merit to be explored in the future for versatile modifications of acetamido sugars.

Next, replacement of pyridine by less nucleophilic bases (*sym*-collidine, triethylamine) was tried but was not successful. Also, introduction of the 2-amino moiety via oxidation of the secondary alcohol, oxime formation

and reduction were explored, but proved to be inefficient. Eventually, reactivity of the leaving group was modified by replacing the triflyl group by a mesyl group in order to prevent amide activation. Reaction of **9** with methanesulfonyl chloride for 2 days at room temperature afforded the crude mesylate **14** which was directly subjected to reaction with sodium azide in DMF (Scheme 2). Reaction conditions had to be optimized to prevent degradation but still to enable progress of the reaction. Heating to 120 °C and a prolonged reaction time of 5 days was necessary to eventually give the azido-derivative **11** in 70% yield. The structure of the resulting 2-azido-2-deoxy-mannosyl fragment was evident from the small value of the homonuclear coupling constant $J_{1',2'}$ being characteristic of a *manno*-glycoside (1.4 Hz) and by the value of the heteronuclear coupling constant ($J_{C1',H1'}$ = 160 Hz).

Scheme 2. Synthesis of target disaccharide 19: Azide introduction, pyruvylation and global deprotection

Reduction of the azido group with ensuing N-acetylation was accomplished via a Staudinger reaction of 11 using polymer-bound triphenylphosphine as described previously. 16 The resin had to be extracted thoroughly with aqueous MeOH to recover the intermediate free amine, which was subsequently N-acetylated under standard conditions to furnish the acetamido derivative 15 in 69% yield. For the introduction of the pyruvyl unit, the benzylidene group was cleaved by the action of trifluoroacetic acid to give diol 16 in near theoretical yield. Installation of the pyruvyl moiety on the β -D-ManNAc residue was achieved by reaction of **16** with methyl pyruvate - promoted by TMSO-triflate in acetonitrile - to generate the (S)-isomer 17 in a good yield of 75%. 35,36 The stereochemical assignment of the pyruvyl ketal was based on literature-known chemical shift features.³⁷ The ¹³C NMR chemical shift of the methyl carbon in S-configured 4,6-O-pyruvyl ketals in hexopyranoses was observed significantly downfield (17: 25.55 ppm) compared to the R-configured counterparts.³⁸ In addition, diagnostic ROESY correlations were observed between the geminal H-6 protons of the ManNAc unit and the methyl ester group. Deprotection of 17 was carried out by hydrogenolysis of the benzyl protecting groups with 10% Pd-carbon in methanol to afford 18, followed by saponification of the methyl ester with aqueous NaOH to furnish disaccharide 19 as the sodium salt. NMR data of 19 were in full agreement with the structural assignments and the NMR features of the native glycan. 14 Binding and crystallographic data with the SpaA_{SLH} protein from P. alvei in the presence of disaccharide 19 will be published in due course.

Experimental Section

General. All purchased chemicals were used without further purification unless stated otherwise. Solvents were dried over activated 4 Å (CH₂Cl₂, DMF) and 3 Å (CH₃CN) molecular sieves. Cation exchange resin DOWEX

Page 5 ©AUTHOR(S)

50 H⁺ was regenerated by consecutive washing with HCl (3 M), water and dry MeOH. Aqueous solutions of salts were saturated unless stated otherwise. Concentration of organic solutions was performed under reduced pressure < 40 °C. Optical rotations were measured with an Anton Paar MCP100 Polarimeter. Thin layer chromatography was performed on Merck precoated plates: generally, on 5 x 10 cm, layer thickness 0.25 mm, Silica Gel 60F₂₅₄; alternatively, on HPTLC plates with 2.5 cm concentration zone (Merck). Spots were detected by staining with a dipping reagent (anisaldehyde-H₂SO₄) and heating. For column chromatography silica gel (0.040 – 0.063 mm) was used. HP-column chromatography was performed on pre-packed columns (YMC-Pack SIL-06, 0.005 mm, 25 x 1 cm and 25 x 2 cm). NMR spectra were recorded with a Bruker Avance III 600 instrument (600.22 MHz for ¹H, 150.93 MHz for ¹³C, 564.77 MHz for ¹⁹F) using standard Bruker NMR software. ¹H spectra were referenced to 7.26 (CDCl₃), 3.34 (MeOD) and 0.00 (D₂O, external calibration to 2,2-dimethyl-2-silapentane-5-sulfonic acid) ppm unless stated otherwise. ¹³C spectra were referenced to 77.16 (CDCl₃), 53.52 (CD₂Cl₂), 49.00 (CD₃OD) and 67.40 (D₂O, external calibration to 1,4-dioxane) ppm. Assignments were based on COSY, HSQC, HMBC and TOCSY data. ESI-MS data were obtained on a Micromass Q-TOF Ultima Global instrument.

2.4.6-Tri-O-acetyl-3-O-benzyl-p-glucopyranose (2). A solution of 1 (3.8 g, 8.67 mmol) and NH₄OAc (2.67 g, 34.7 mmol) in dry DMF (35 mL) was stirred for 5 days at rt under Ar when TLC showed complete conversion. The solution was concentrated in vacuo and the residue purified by chromatography on silica gel (toluene-EtOAc 4:1 \rightarrow 1:1) to give 2 (2.62 g, 76%) as colorless syrup; R_f = 0.46 (toluene-EtOAc 4:1); ¹H NMR (600 MHz, CDCl₃): $\delta = 7.34-7.22$ (m, 5 H, Ar-H), 5.45 (dd, 1 H, ${}^{3}J_{1.0H} = 3.7$ Hz, ${}^{3}J_{1.2} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.5} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.5} = 3.7$ Hz, H-1_{\alpha}), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.5} = 3.7$ 9.7 Hz, H-4_B), 5.09 (t, 1 H, ${}^{3}J_{4.5} = {}^{3}J_{4.3} = 9.7$ Hz, H-4_{\alpha}), 4.87 (dd, 1 H, ${}^{3}J_{2.3} = 9.2$, ${}^{3}J_{2.1} = 7.9$ Hz, H-2_B), 4.87 (dd, 1 H, $^{3}J_{2,3} = 9.7$, $^{3}J_{2,1} = 3.9$ Hz, H-2 $_{\alpha}$), 4.71 (d, 1 H, $^{2}J = 12.1$ Hz, CH₂Ar $_{\alpha}$), 4.66 (d, 1 H, $^{2}J = 11.6$ Hz, CH₂Ar $_{\beta}$), 4.62 (d, 1 H, $^{2}J = 11.9 \text{ Hz}$, $CH_{2}Ar_{\alpha}$), 4.61 (d, 1 H, $^{2}J = 11.8 \text{ Hz}$, $CH_{2}Ar_{\beta}$), 4.61 (d, 1 H, $^{3}J_{1,2} = 8.2 \text{ Hz}$, H-1_{\beta}), 4.19 (dd, 1 H, $^{2}J_{6a,6b} = 1.00 \text{ Hz}$ 12.4, ${}^{3}J_{6a,5} = 5.2$ Hz, H-6a_β), 4.18 -4.08 (m, 4 H, H-6b_β, H-6a_α, H-6b_α, H-5a), 4.04 (t, 1 H, ${}^{3}J_{3,4} = {}^{3}J_{3,2} = 9.6$ Hz, H-3a), 3.79 (d, 1 H, ${}^{3}J_{OH,1}$ = 9.6 Hz, OH_B), 3.72 (t, 1 H, ${}^{3}J_{3,4}$ = ${}^{3}J_{3,2}$ = 9.3 Hz, H-3_B), 3.63 (ddd, 1 H, ${}^{3}J_{5,4}$ = 10.1, ${}^{3}J_{5,6a}$ = 5.2, $^{3}J_{5.6b} = 2.4 \text{ Hz}, \text{ H-5}_{\beta}$), 3.33 (d, 1 H, $^{3}J_{OH.1} = 3.3 \text{ Hz}, \text{ OH}_{\alpha}$), 2.08 (s, 3 H, CH₃CO), 2.06 (s, 3 H, CH₃CO_{α}), 2.05 (s, 3 H, CH₃CO₆), 1.97 (s, 3 H, CH₃CO₆), 1.95 (s, 3 H, CH₃CO_{α}); ¹³C NMR for α -anomer (150 MHz, CDCl₃): δ = 170.90, 170.10 and 169.48 (C=O), 138.12 (Cq, Ar-C), 128.39 (2 C, Ar-C), 127.71 (Ar-C), 127.54 (2 C, Ar-C), 90.31 (C-1), 76.86 (C-3), 74.88 (CH₂Ar), 73.41 (C-2), 69.79 (C-4), 67.73 (C-5), 62.27 (C-6), 20.82, 20.74 and 20.70 (CH₃CO); ¹³C NMR for β-anomer (150 MHz, CDCl₃): δ = 171.19, 170.85 and 169.39 (C=O), 137.68 (Cq, Ar-C), 128.46 (2 C, Ar-C), 127.90 (Ar-C), 127.67 (2 C, Ar-C), 95.84 (C-1), 79.59 (C-3), 75.39 (C-2), 74.46 (CH₂Ar), 72.30 (C-5), 69.66 (C-4), , 62.27 (C-6), 20.82, 20.74 and 20.70 (CH₃CO) ppm. NMR data were in agreement with the literature.³⁹ 3-O-Benzyl-2,4,6-tri-O-acetyl-D-glucopyranosyl N-phenyl-trifluoroacetimidate (3). K₂CO₃ (660 mg, 4.773 mmol) was added to a solution of 2 (860 mg, 2.170 mmol) in dry DCM (30 mL) under Ar followed by dropwise addition of N-phenyltrifluoroacetimidoyl chloride (NPTFI-Cl, 690 µL, 4.34 mmol) at room temperature. The suspension was stirred for 7 d, then filtered over Celite® and rinsed with DCM. The organic phase was concentrated in vacuo to give a yellowish, waxy product. Purification of the residue by column chromatography (toluene-EtOAc 4:1 containing 0.2% TEA) afforded 1.008 g (82%) of 3 as off-white waxy solid; $R_f = 0.27$ (toluene-EtOAc 4:1); $[\alpha]_D^{20} + 29$ (c 1.3 CHCl₃); ¹H NMR: (600 MHz, CDCl₃) $\delta = 7.36-7.32$ (m, 3 H, Ar-H), 7.32-7.21 (m, 2 H, Ar-H), 7.25-7.23 (m, 2 H, Ar-H), 7.14-7.11 (m, 1 H, Ar-H), 6.86-6.80 (m, 2 H, Ar-H), 5.78-5.60 (br s, 1 H, H-1), 5.30 (t, 1 H, ${}^{3}J_{4,3} = {}^{3}J_{4,5} = 8.3$ Hz, H-4), 5.19 (t, 1 H, ${}^{3}J_{2,3} = {}^{3}J_{2,1} = 9.3$ Hz, H-2), 4.64 (d, 1 H, ${}^{2}J = 11.7$ Hz, CH₂Ar), 4.61 (d, 1 H, ${}^{2}J$ = 11.8 Hz, CH₂Ar), 4.21 (dd, 1 H, ${}^{3}J_{6a,5}$ = 5.3, ${}^{2}J_{6a,6b}$ = 12.6 Hz, H-6a), 4.16-4.09 (m, 1 H, H-6b), 3.77-3.58 (m, 1 H, H-5), 3.75 (dd, 1 H, ${}^{3}J_{3,4}$ = 8.0 Hz, ${}^{3}J_{3,2}$ = 9.2 Hz, H-3), 2.06, 2.03 and 1.98 (3 x s, each 3 H, CH₃CO); ¹³C NMR: (150 MHz, CDCl₃): δ = 128.82, 128.49, 127.97, 127.83, 124.57 and 119.24 (Ar-C), 79.55 (C-

Page 6 ©AUTHOR(S)

3), 73.89 (CH_2Ar), 73.05 (C-5), 71.21 (C-4), 68.99 (C-2), 61.84 (C-6), 20.68 and 20.59 (CH_3CO) ppm. HRMS ($^+ESI-TOF$) m/z [M+Na] $^+$ calcd for $C_{27}H_{28}F_3NNaO_9$, found 590.1598.

2,4,6-tri-O-acetyl-3-O-benzyl-β-D-glucopyranosyl-(1→4)-2-acetamido-3,6-di-O-benzyl-2-deoxy-β-Dglucopyranoside (6). A suspension of 4 (490 mg, 1.18 mmol), donor 3 (880 mg, 1.55 mmol) and acid-washed molecular sieves 4 Å (1.37 g) in DCM (30 mL) was stirred at rt under Ar for 30 min, then cooled with an external NaCl/ice-bath for further 40 min. TMSOTf (21 µL, 0.12 mmol) was added to the mixture stirred overnight at rt. TLC showed complete consumption of the donor, intermediate orthoester ($R_f = 0.63$, EtOAc-DCM 4:1) and unreacted acceptor. The reaction mixture was stirred for additional 24 h, followed by quenching with TEA (150 µL) and filtration over Celite®. The filtrate was concentrated in vacuo and the residue subjected to column chromatography (toluene-EtOAc 3:2 \rightarrow 2:3 \rightarrow 0:1) to give a 1:1 mixture of 620 mg of the desired disaccharide 5 (43 %) as well as recovered acceptor 4 (43 %); R_f = 0.39 (EtOAc-DCM 4:1). An analytical aliquot of the mixture was purified by HILIC-HPLC using a gradient H₂O \rightarrow MeCN; $[\alpha]_D^{20}$ -19 (c 0.2, CHCl₃); ¹H NMR: (600 MHz, CDCl₃): $\delta = 7.37-7.20$ (m, 15 H, Ar-H), 5.91 (d, 1 H, ${}^{3}J_{NH,2} = 8.3$ Hz, NH), 5.08 (t, 1 H, ${}^{3}J_{4',3'} = {}^{3}J_{4',5'} = 9.7$ Hz, H-4'), 5.00 (dd, 1 H, ${}^{3}J_{2',3'}$ = 9.6, ${}^{3}J_{2',1'}$ = 8.2 Hz, H-2'), 4.72 (d, 1 H, ${}^{2}J$ = 11.9 Hz, CH₂Ar), 4.65 (d, 1 H, ${}^{2}J$ = 12.4 Hz, CH₂Ar), 4.63 (d, 1 H, 2J = 12.2 Hz, CH₂Ar), 4.59 (d, 1 H, 2J = 12.2 Hz, CH₂Ar), 4.57 (d, 1 H, $^3J_{1,2}$ = 8.0 Hz, H-1), 4.57 (d, 1 H, 2J = 12.5 Hz, CH₂Ar), 4.49 (d, 1 H, 2J = 12.0 Hz, CH₂Ar), 4.44 (d, 1 H, $^3J_{1',2'}$ = 8.0 Hz, H-1'), 4.18 (dd, 1 H, $^{3}J_{6a',5'} = 4.8$, $^{2}J_{6a',6b'} = 12.3$ Hz, H-6a'), 3.99 (dd, 1 H, $^{3}J_{6b',5'} = 2.7$, $^{2}J_{6b',6a'} = 12.6$ Hz, H-6b'), 3.97 (dd, 1 H, H-4), 3.84 $(dd, 1 H, {}^{3}J_{6a,5} = 5.2, {}^{2}J_{6a,6b} = 10.2 Hz, H-6a), 3.81 (dd, 1 H, H-3), 3.79 (dd, 1 H, {}^{3}J_{6b,5} = 4.7, {}^{2}J_{6b,6a} = 10.4 Hz, H-6b),$ 3.64 (ddd, 1 H, ${}^{3}J_{5,6b} = 4.7$, ${}^{3}J_{5,4} = 10.0$, ${}^{3}J_{5,6a} = 5.2$ Hz, H-5), 3.57 (t, 1 H, ${}^{3}J_{3',2'} = {}^{3}J_{3',4'} = 9.5$ Hz, H-3'), 3.43 (s, 3 H, OCH₃), 3.40 (ddd, 1 H, ${}^{3}J_{5',6b'}$ = 2.5, ${}^{3}J_{5',6a'}$ = 4.9, ${}^{3}J_{5',4'}$ = 10.1 Hz, H-5'), 1.99 (s, 3 H, CH₃CO), 1.98 (s, 3 H, CH₃CO), 1.97 (s, 3 H, CH₃CO), 1.92 (s, 3 H, CH₃CO); ¹³C NMR: (150 MHz, CDCl₃): δ = 170.71 (C=O), 170.25 (C=O), 169.62 (C=O), 169.28 (C=O), 138.45 (Cq, Ar-C), 138.08 (Cq, Ar-C), 137.72 (Cq, Ar-C), 129.57 (Ar-C), 128.47 (2 C, Ar-C), 128.45 (2 C, Ar-C), 128.29 (2 C, Ar-C), 127.90 (2 C, Ar-C), 127.88 (2 C, Ar-C), 127.84 (2 C, Ar-C), 127.66 (2 C, Ar-C) C), 127.56 (1 C, Ar-C), 101.30 (C-1), 99.67 (C-1'), 79.86 (C-3'), 76.74 (C-3), 74.97 (C-4), 74.36 (C-5), 74.04 (CH₂Ar), 73.55 (CH₂Ar), 72.94 (C-2'), 72.89 (CH₂Ar), 72.04 (C-5'), 69.53 (C-4'), 69.12 (C-6), 61.98 (C-6'), 56.50 (OCH₃), 52.05 (C-2), 23.28 (HNCOCH₃), 20.88 (CH₃CO), 20.72 (CH₃CO), 20.68 (CH₃CO) ppm. HRMS (*ESI-TOF) m/z [M+H]⁺ calcd for C₄₂H₅₁NO₁₄ 794.3382; found 794.3384.

Methyl 3-O-benzyl-4,6-O-benzylidene-β-D-glucopyranosyl-(1→4)-2-acetamido-3,6-di-O-benzyl-2-deoxy-β-Dglucopyranoside (8). The mixture of 4 and 5 was dissolved in MeOH (8 mL), water (1 mL) and treated with triethylamine (1 mL) for 40 h at rt. The solution was then concentrated, and the residue was subjected to chromatography on silica gel (EtOAc) to remove monosaccharide acceptor 4 and to give a 3:1 mixture of 6 and the 2-O-acetyl derivative **7** (245 mg). Data for **6**: $R_f = 0.23$ (EtOAc); ¹H NMR (600 MHz, CDCl₃): $\delta = 7.37-7.27$ (m, 15 H, Ar-H), 5.64 (d, 1 H, ${}^{3}J_{NH,2}$ = 7.8 Hz, N-H), 4.95 (d, 1 H, ${}^{2}J$ = 11.6 Hz, CH₂Ar), 4.87 (d, 1 H, ${}^{2}J$ = 11.6 Hz, CH₂Ar), 4.74 (d, 1 H, ${}^{3}J_{1,2} = 7.8$ Hz, H-1), 4.73 (d, 1 H, ${}^{2}J = 11.6$ Hz, CH₂Ar), 4.70 (d, 1 H, ${}^{2}J = 12.0$ Hz, CH₂Ar), 4.62 (d, 1 H, $^{2}J = 11.9 \text{ Hz}$, CH₂Ar), 4.56 (d, 1 H, $^{2}J = 12.0 \text{ Hz}$, CH₂Ar), 4.51 (d, 1 H, $^{3}J_{1',2'} = 7.8 \text{ Hz}$, H-1'), 4.12 (dd, 1 H, $^{3}J_{3,4} = 1.0 \text{ Hz}$ 8.9, ${}^{3}J_{3,2} = 8.9 \text{ Hz}$, H-3), 3.97 (dd, 1 H, ${}^{3}J_{4,3} = 8.9$, ${}^{3}J_{4,5} = 8.9 \text{ Hz}$, H-4), 3.92 (dd, 1 H, ${}^{3}J_{6a,5} = 3.6$, ${}^{2}J_{6a,6b} = 11.2 \text{ Hz}$, H-6a), 3.80 (dd, 1 H, ${}^{3}J_{6b,5} = 2.9$, ${}^{2}J_{6b,6a} = 11.4$ Hz, H-6b), 3.67 (dd, 1 H, ${}^{3}J_{6a',5'} = 3.5$, ${}^{2}J_{6a',6b'} = 11.9$ Hz, H-6a'), 3.59 $(ddd, 1 H, {}^{3}J_{5,6b} = 2.9, {}^{3}J_{5,6a} = 3.4, {}^{3}J_{5,4} = 9.1 Hz, H-5), 3.50 (dd, 1 H, {}^{3}J_{6b',5'} = 5.3, {}^{2}J_{6b',6a'} = 11.9 Hz, H-6b'), 3.47 (dd, 1 H, 1 H)$ 1 H, ${}^{3}J_{4',3'}$ = 9.3, ${}^{3}J_{4',5'}$ = 9.3 Hz, H-4'), 3.47 (s, 3 H, OMe), 3.44 (dd, 1 H, ${}^{3}J_{2',3'}$ = 9.3, ${}^{3}J_{2',1'}$ = 7.7 Hz, H-2'), 3.37 (ddd, 1 H, ${}^{3}J_{2,3} = 8.6$, ${}^{3}J_{2,1} = 7.7$, ${}^{3}J_{2,N-H} = 7.7$ Hz, H-2), 3.26 (dd, 1 H, ${}^{3}J_{3',2'} = {}^{3}J_{3',4'} = 9.1$ Hz, H-3'), 3.11 (ddd, 1 H, ${}^{3}J_{5',6a'} =$ 3.5, ${}^{3}J_{5',6b'} = 5.4$, ${}^{3}J_{5',4'} = 9.6$ Hz, H-5') and 1.86 (s, 3H, NAc); ${}^{13}C$ NMR: (150 MHz, CDCl₃): $\delta = 170.60$ (C=O), 138.60 (2 C, Cq, Ar-C), 137.60 (Cq, Ar-C), 128.62 (2 C, Ar-C), 128.47 (2 C, Ar-C), 128.09 (2 C, Ar-C), 127.93 (3 C, Ar-C), 127.70 (1 C, Ar-C), 127.37 (2 C, Ar-C), 102.56 (C-1'), 100.79 (C-1), 83.82 (C-3'), 78.71 (C-3), 77.53 (C-4), 75.28 (C-5'), 75.13 (C-2'), 74.68 (CH₂Ar), 74.46 (C-5), 74.20 (CH₂Ar), 73.69 (CH₂Ar), 70.24 (C-4'), 68.74 (C-6), 62.42 (C-

Page 7 [©]AUTHOR(S)

6'), 56.98 (C-2), 56.68 (OCH₃) and 23.57 (NHCOCH₃) ppm. ESI-QTOFMS: m/z calcd for C₃₆H₄₅NO₁₁: [M+H⁺]⁺ 668.3065; found 668.3066.

The mixture was dispersed in MeCN (5 mL) under Ar. Benzylidene dimethylacetal (0.079 mL, 0.526 mmol) and FeCl₃ (11 mg, 0.07 mmol) were then added and the dispersion was stirred for 3 h at rt. The mixture was concentrated, and the residue was dissolved in EtOAc (100 mL), washed with satd ag NaHCO₃ and water. The organic phase was dried and concentrated to afford a ~4:1 mixture of 8 and 9 as syrup (275 mg). A solution of the residue in dry MeOH (9 mL) was stirred with solid NaOMe (13 mg, 0.241 mmol) for 18 days at rt. The reaction was quenched by addition of Dowex 50 H⁺ resin and filtered. The filtrate was concentrated and flashchromatographed with dichloromethane to give 9 as colorless fluffy solid. Yield: 250 mg (0.33 mmol, 65% for 3 steps). $R_f = 0.45$ (EtOAc); $[\alpha]_D^{20} + 5.6$ (c 0.95, CHCl₃); ¹H NMR (600 MHz, CDCl₃): $\delta = 7.42 - 7.26$ (m, 20 H, ArH), 5.56 (d, 1 H, ${}^{3}J_{NH,2}$ = 7.7 Hz, NH), 5.47 (s, 1 H, ArCH), 4.93 (d, 1 H, ${}^{2}J$ = 11.5 Hz, CH₂Ar), 4.88 (d, 1 H, ${}^{2}J$ = 11.5 Hz, CH₂Ar), 4.78 (d, 1 H, ${}^{3}J_{1,2}$ = 7.8 Hz, H-1), 4.76 (d, 1 H, ${}^{2}J$ = 11.7 Hz, CH₂Ar), 4.69 (d, 1 H, ${}^{2}J$ = 11.9 Hz, CH₂Ar), 4.59 (d, 1 H, 2J = 11.9 Hz, CH₂Ar), 4.59 (d, 1 H, $^3J_{1',2'}$ = 7.0 Hz, H-1'), 4.54 (d, 1 H, 2J = 12.2 Hz, CH₂Ar), 4.12 (t, 1 H, $^3J_{3,4}$ = ${}^{3}J_{3.2}$ = 9.0 Hz, H-3), 4.08 (dd, 1 H, ${}^{3}J_{6a'.5'}$ = 5.0, ${}^{2}J_{6a'.6b'}$ = 10.4 Hz, H-6a'), 4.00 (t, 1 H, ${}^{3}J_{4,3}$ = ${}^{3}J_{4,5}$ = 8.8 Hz, H-4), 3.98 (dd, 1 H, ${}^{3}J_{6a,5}$ = 3.3, ${}^{2}J_{6a,6b}$ = 11.0 Hz, H-6a), 3.80 (dd, 1 H, ${}^{2}J_{6b,6a}$ = 11.2, ${}^{3}J_{6b,5}$ = 2.3 Hz, H-6b), 3.58 (dd, 1 $H_{1}^{3}J_{4',3'} = 8.9$, $^{3}J_{4',5'} = 8.9$ Hz, H-4'), 3.58 (m, 1 H, H-5), 3.53 (dd, 1 $H_{1}^{2}J_{6b',6a'} = 10.4$, $^{3}J_{6b',5'} = 10.4$ Hz, H-6b'), 3.49 (t, 1 H, ${}^{3}J_{3',2'} = {}^{3}J_{3',4'} = 8.4$ Hz, H-3'), 3.47 (dd, 1 H, ${}^{3}J_{2',3'} = 7.7$, ${}^{3}J_{2',1'} = 7.3$ Hz, H-2'), 3.47 (s, 3 H, OCH₃), 3.31 (ddd, 1 H, $^{3}J_{2,3} = 9.4$, $^{3}J_{2,1} = 7.8$, $^{3}J_{2,NH} = 7.8$ Hz, H-2), 2.87 (dt, 1 H, $^{3}J_{5',60'} = 5.0$, $^{3}J_{5',6b'} = ^{3}J_{5',4'} = 9.7$ Hz, H-6b'), 1.86 (s, 3 H, NHCOCH₃); ¹³C NMR: (150 MHz, CDCl₃): 170.41 (C=O), 138.79 (Cq, Ar-C), 138.44 (Cq, Ar-C), 137.81 (Cq, Ar-C), 137.31 (Cq, Ar-C), 128.96 (Ar-C), 128.42 (2 C, Ar-C), 128.37 (2 C, Ar-C), 128.29 (2 C, Ar-C), 128.22 (2 C- Ar-C), 128.02 (2 C, Ar-C), 127.97 (2 C, Ar-C), 127.86 (Ar-C), 127.76 (Ar-C), 127.60 (Ar-C), 127.44 (2 C, Ar-C), 126.03 (2 C, Ar-C), 103.21 (C-1), 101.20 (ArCH), 100.77 (C-1), 81.31 (C-4'), 80.33 (C-3'), 78.33 (C-3), 77.91 (C-4), 75.03 (C-2'), 74.52 (CH₂Ar), 74.42 (C-5), 74.24 (CH₂Ar), 73.53 (CH₂Ar), 68.64 (C-6'), 68.49 (C-6), 66.28 (C-5'), 57.09 (C-2), 56.69 (OCH₃), 23.60 (CH₃CO). ESI-QTOFMS: m/z calcd for C₄₃H₄₉NO₁₁: [M+H⁺]⁺ 756.3378; found 756.3381.

Methyl 2-azido-3-O-benzyl-4,6-O-benzylidene-2-deoxy-β-D-mannopyranosyl-(1→4)-3,6-di-O-benzyl-2-deoxy-2-(1H-5-methyl-1,2,3,4-tetrazol-1-yl)-β-D-glucopyranoside (10). Compound 9 (20 mg, 0.026 mmol) was dried on a high vacuum line and coevaporated with toluene twice. DCM (0.2 mL) and pyridine (0.1 mL) were added under Ar through a septum and the mixture was cooled with an external ice/NaCl-mixture. A 1 M Tf₂O-solution in DCM (63 µL) was added and the colour changed to dark violet/blue. The solution was stirred for 30 min, whereupon additional Tf₂O-solution (20 μL) was added and the reaction warmed up to room temperature and kept stirring for 70 min. The mixture was diluted with DCM (10 mL) and quenched with saturated NaHCO₃solution (10 mL). The aqueous phase was washed with DCM (5 mL), the combined organic phases were dried over MgSO₄ and filtered. Concentration of the filtrate afforded the crude triflate (25 mg) as brown oil. The activated triflate was dissolved in dry DMF (0.3 mL) under Ar-atmosphere and NaN₃ (8.5 mg, 0.13 mmol) was added and the mixture was stirred for a6 h at rt. Thereafter a condenser was attached to the flask and the mixture was heated to 70 °C for 2 h, when TLC showed complete consumption of the starting material. The reaction mixture was diluted with EtOAc (10 mL) and washed with H₂O (10 mL) and brine (10 mL). The organic phase was dried with MgSO₄ and concentrated in vacuo. The residue was purified by column chromatography (toluene-EtOAc 5:1) giving **10** (7 mg, 33 %) and **11** (2 mg, 10 %) as syrup. Data for **10**: $R_f = 0.88$ (toluene-EtOAc 5:1); $[\alpha]_D^{20}$ -29.8 (c 0.6, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ = 7.48-7.46 (m, 2 H, Ar-H), 7.41-7.31 (m, 13 H, Ar-H) H), 7.25-7.23 (m, 3 H, Ar-H), 6.94-6.92 (m, 2 H, Ar-H), 5.51 (s, 1 H, CHAr), 4.92 (d, 1 H, 2J = 10.6 Hz, CH₂Ar), 4.82 (d, 1 H, ${}^{3}J_{1,2}$ = 8.2 Hz, H-1), 4.81 (d, 1 H, ${}^{2}J$ = 12.3 Hz, CH₂Ar), 4.75 (d, 1 H, ${}^{2}J$ = 11.7 Hz, CH₂Ar), 4.66 (d, 1 H, ${}^{2}J$ = 12.4 Hz, CH₂Ar), 4.63 (d, 1 H, ${}^{3}J_{1',2'}$ = 1.3 Hz, H-1'), 4.44 (d, 1 H, ${}^{2}J$ = 12.0 Hz, CH₂Ar), 4.28 (dd, 1 H, ${}^{3}J_{3,4}$ = 9.2, ${}^{3}J_{3,2}$ = 10.8 Hz, H-3), 4.16 (dd, 1 H, ${}^{3}J_{4,3}$ = 8.9, ${}^{3}J_{4,5}$ = 9.9 Hz, H-4), 4.11 (dd, 1 H, ${}^{3}J_{6a',5'}$ = 5.0, ${}^{2}J_{6a',6b'}$ = 10.4 Hz, H-6a'),

Page 8 ©AUTHOR(S)

4.10 (d, 1 H, 2J = 10.7 Hz, CH₂Ar), 4.05 (dd, 1 H, ${}^3J_{2,1}$ = 8.1, ${}^3J_{2,3}$ = 10.4 Hz, H-2), 3.91 (t, 1 H, ${}^3J_{4',3'}$ = ${}^3J_{4',5'}$ = 9.5 Hz, H-4'), 3.80 (dd, 1 H, ${}^3J_{60,5}$ = 2.3, ${}^2J_{60,6b}$ = 11.2 Hz, H-6a), 3.80 (dd, 1 H, ${}^3J_{2',1'}$ = 1.2, ${}^3J_{2',3'}$ = 3.8 Hz, H-2'), 3.78 (dd, 1 H, ${}^3J_{6b,5}$ = 2.8, ${}^2J_{6b,6a}$ = 11.4 Hz, H-6b), 3.65 (ddd, 1 H, ${}^3J_{5,6a}$ = 2.5, ${}^3J_{5,6b}$ = 2.5, ${}^3J_{5,4}$ = 9.8 Hz, H-5), 3.56 (dd, 1 H, ${}^3J_{6b',5'}$ = 10.1, ${}^2J_{6b',6a'}$ = 10.5 Hz, H-6b'), 3.47 (dd, 1 H, ${}^3J_{3',4'}$ = 9.6, ${}^3J_{3',2'}$ = 3.9 Hz, H-3'), 3.37 (s, 3 H, OCH₃), 3.07 (dt, 1 H, ${}^3J_{5',6a'}$ = 4.8, ${}^3J_{5',6b'}$ = ${}^3J_{5',4'}$ = 9.8 Hz, H-5'), 2.44 (s, 3 H, CH₃C); 13 C NMR: (150 MHz, CDCl₃): δ = 153.69 (N-C=N), 137.91 (Cq, Ar-C), 137.52 (Cq, Ar-C), 137.35 (Cq, Ar-C), 137.25 (Cq, Ar-C), 129.04 (1 C, Ar-C), 128.70 (2 C, Ar-C), 128.50 (2 C, Ar-C), 128.35 (1 C, Ar-C), 128.27 (2 C, Ar-C), 128.25 (2 C, Ar-C), 128.12 (2 C, Ar-C), 127.88 (1 C, Ar-C), 127.84 (1 C, Ar-C), 127.79 (2 C, Ar-C), 127.49 (2 C, Ar-C), 126.03 (2 C, Ar-C), 101.72 (C-1), 101.57 (CHAr), 99.89 (C-1'), 80.03 (C-3), 78.41 (C-4'), 77.58 (C-4), 76.53 (C-3'), 75.39 (CH₂Ar), 74.40 (CH₂Ar), 73.84 (CH₂Ar), 72.84 (C-6'), 68.33 (C-6), 68.06 (C-5'), 67.29 (C-2'), 63.63 (C-2), 57.42 (OCH₃), 8.82 (CH₃C) ppm; HRMS (+ESI-TOF) m/z [M+H]+ calcd for C₄₃H₄₇N₇O₁₁, found 806.3490.

Reaction of 2-acetamido-1,3,4,6-tetra-O-acetyl-β-D-glucopyranose (12) with triflic anhydride and pyridine. Compound 12 (8 mg, 0.021 mmol) was dried in high vacuum overnight and dissolved in CD₂Cl₂ (0.4 mL) and pyridine-_{d5} (0.2 mL) under Ar-atmosphere in an NMR tube. Tf₂O (0.045 mL of a 1 M solution in DCM) was added slowly at room temperature and NMR spectra recorded over several hours. Data for 13: ¹H NMR (600 MHz, CDCl₃): δ = 8.56 (br signal, 0.6 H, H-2,6-pyr), 7.61 (br signal, 0.53 H, H-4-pyr), 7.21 (br signal, H-3,5-pyr), 6.02 (dd, 1 H, ³J_{1,2} = 8.1, J = 0.8 Hz, H-1), 5.57 (t, 1 H, ³J_{3,2} = ³J_{3,4} = 9.5 Hz, H-3), 5.28 (ddd, 1 H, ³J_{4,3} = 9.5, ³J_{4,5} = 10.2 Hz, H-4), 4.35 (dd, 1 H, ³J_{60,5} = 4.5, ²J_{60,6b} = 12.5 Hz, H-6a), 4.23 (ddd, 1 H, J = 0.7, ³J_{2,1} = 8.1, ³J_{2,3} = 9.4 Hz, H-2), 4.14 (dd, 1 H, ³J_{6b,5} = 2.3, ²J_{6b,5} = 12.5 Hz, H-6b), 4.06 (ddd, 1 H, ³J_{5,6b} = 2.4, ³J_{5,6a} = 4.4, ³J_{5,4} = 10.2 Hz, H-5), 2.88, 2.01, 2.00, 1.99 and 1.97 (5 s, each 3 H, CH₃CO); ¹³C NMR: (150 MHz, CDCl₃): δ = 170.12 (C=O), 169.99 (C=O), 169.39 (C=O), 168.70 (C=O), 158.66 (C=N), 149.12 (2 C, J = 27.4 Hz, C-2,6-pyr), 135.40 (J = 24.7 Hz, C-4-pyr), 123.11 (2 C, J = 24.8 Hz, C-3,5-pyr), 92.71 (C-1), 73.12 (C-3), 72.86 (C-5), 67.58 (C-4), 64.23 (C-2), 61.54 (C-6), 20.43 (CH₃CO), 20.31 (CH₃CO), 20.28 (CH₃CO), 20.22 (CH₃CO), 15.51 (CH₃C=N); ¹⁹F NMR (564.7 MHz, CDCl₃): δ -78.41 ppm.

Methyl 2-azido-3-O-benzyl-4,6-O-benzylidene-2-deoxy-β-D-mannopyranosyl-(1→4)-2-acetamido-3,6-di-Obenzyl-2-deoxy-β-D-glucopyranoside (11). A solution of 9 (160 mg, 0.212 mmol) in dry pyridine (2.5 mL) and DCM (5 mL) was cooled with an ice bath under Ar. Mesyl chloride (49 µL, 0.635 mmol), was then added through a septum with a syringe and the solution was stirred for 2 d at rt. The mixture was diluted with DCM (20 mL), washed with water (20 mL) and NaHCO₃-solution (20 mL). The aqueous phase was reextracted with DCM (3 x 10 mL). The combined organic phases were dried (MgSO₄) and concentrated in vacuo to obtain 170 mg (96 %) of **14** as a brownish amorphous solid; $R_f = 0.58$ (EtOAc); $[\alpha]_D^{20}$ -27.4 (c 0.23, CHCl₃); ¹H NMR (600 MHz, CDCl₃): $\delta = 7.40-7.27$ (m, 20 H, Ar-H), 5.94 (d, 1 H, ${}^{3}J_{NH,2} = 8.3$ Hz, NH), 5.51 (s, 1 H, ArCH), 4.99 (d, 1 H, ${}^{2}J =$ 11.2 Hz, CH₂Ar), 4.70 (s, 2 H, CH₂Ar), 4.68 (d, 1 H, ${}^{3}J_{1,2}$ = 6.5 Hz, H-1), 4.65 (d, 1 H, ${}^{2}J$ = 11.8 Hz, CH₂Ar), 4.64 (d, 1 H, ${}^{2}J = 11.1$ Hz, CH₂Ar), 4.57 (d, 1 H, ${}^{3}J_{1',2'} = 7.9$ Hz, H-1'), 4.50 (d, ${}^{2}J = 11.5$ Hz, CH₂Ar), 4.37 (dd, 1 H, ${}^{3}J_{2',1'} = 8.4$ Hz, ${}^{3}J_{2',3'} = 8.4$ Hz, H-2'), 4.24 (dd, 1 H, ${}^{3}J_{6a',5'} = 4.9$ Hz, ${}^{2}J_{6a',6b'} = 10.5$ Hz, 1 H, H-6a'), 4.10 (dd, 1 H, ${}^{3}J_{3,4} = 5.7$, ${}^{3}J_{3,2} = 5.7$ = 5.7 Hz, H-3), 3.91 (dd, 1 H, ${}^{3}J_{6a,5}$ = 2.9, ${}^{2}J_{6a,6b}$ = 9.2 Hz, H-6a), 3.84-3.74 (m, 4 H, H-2, H-5, H-6b, H-5) 3.68 (t, 1 $H_{3}^{3}J_{3',4'} = {}^{3}J_{3',2'} = 8.4 Hz, H-3'), 3.66 (t, 1 H, {}^{3}J_{4',3'} = {}^{3}J_{4',5'} = 9.1 Hz, H-4'), 3.60 (dd, 1 H, {}^{3}J_{6b',5'} = 10.5, {}^{2}J_{6b',6a'} = 10.5$ Hz, H-6b'), 3.46 (s, 3 H, OMe), 3.16 (ddd, 1 H, ${}^{3}J_{5',6b'}$ = 4.6, ${}^{3}J_{5',6a'}$ = 9.4, ${}^{3}J_{5',4'}$ = 9.4 Hz, H-5'), 2.84 (s, 3 H, CH₃SO₂), 1.90 (s, 3 H, NAc). HRMS ($^{+}$ ESI-TOF) m/z [M+H] $^{+}$ calcd for C₄₄H₅₁NO₁₃S, found 834.3154.

A solution of **14** (36 mg, 0.043 mmol) and NaN₃ (44 mg, 0.68 mmol) in dry DMF (1 mL) was stirred under Ar at 140 °C for 16 d. The mixture was diluted with EtOAc and washed with H₂O (10 mL) and brine (10 mL). The organic phase was dried (MgSO₄), filtered and concentrated *in vacuo*. The product was purified by column chromatography (toluene-EtOAc 1:1) to give 23 mg (70 %) of **11** as colorless amorphous solid; $R_f = 0.6$ (EtOAc); $[\alpha]_D^{20}$ -28.2 (*c* 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃): $\delta = 7.47$ -7.27 (m, 20 H, Ar-H), 5.66 (d, 1 H, ³ $J_{NH,2} = 7.8$ Hz,

Page 9 ©AUTHOR(S)

N-H), 4.88 (d, 1 H, 2J = 11.6 Hz, CH₂Ar), 4.78 (d, 1 H, 2J = 12.1 Hz, CH₂Ar), 4.75 (d, 1 H, ${}^3J_{1,2}$ = 7.1 Hz, H-1), 4.66 (d, 1 H, 2J = 11.8 Hz, CH₂Ar), 4.65 (d, 1 H, 2J = 12.4 Hz, CH₂Ar), 4.63 (d, 1 H, 2J = 11.3 Hz, CH₂Ar), 4.62 (d, 1 H, ${}^3J_{1,2}$ = 1.4 Hz, H-1'), 4.45 (d, 1 H, 2J = 11.9 Hz, CH₂Ar), 4.09 (dd, 1 H, ${}^3J_{6a',5'}$ = 5.0, ${}^2J_{6a',6b'}$ = 10.6 Hz, H-6a'), 4.01 (t, 1 H, ${}^3J_{3,2}$ = ${}^3J_{3,4}$ = 8.0 Hz, H-3), 3.95 (t, 1 H, ${}^3J_{4,3}$ = ${}^3J_{4,5}$ = 7.9 Hz, H-4), 3.91 (t, ${}^3J_{6a',5'}$ = 9.5 Hz, 1 H, H-4'), 3.84 (dd, 1 H, ${}^3J_{2',1'}$ = 1.1 Hz, ${}^3J_{2',3'}$ = 3.7 Hz, H-2'), 3.79 (dd, 1 H, ${}^3J_{6a,5}$ = 3.7, ${}^3J_{6a,6b}$ = 10.7 Hz, H-6a), 3.75 (dd, 1 H, ${}^3J_{6b,5}$ = 3.6, ${}^2J_{6b,6a}$ = 10.8 Hz, H-6b), 3.62 (ddd, 1 H, ${}^3J_{5,6a}$ = 3.8, ${}^3J_{5,4}$ = 7.8, ${}^3J_{5,6b}$ = 3.8 Hz, H-5), 3.59 (t, 1 H, ${}^3J_{6b',5'}$ = 10.1, ${}^2J_{6a',6b'}$ = 10.1 Hz, H-6b'), 3.52 (dd, 1 H, ${}^3J_{3',2'}$ = 3.8, ${}^3J_{3',4'}$ = 9.6 Hz, H-3'), 3.48 (ddd, 1 H, ${}^3J_{2,1}$ = 7.7, ${}^3J_{2,3}$ = 8.0, ${}^3J_{2,NH}$ = 8.0 Hz, H-2), 3.47 (s, 3 H, OCH₃), 3.07 (ddd, 1 H, ${}^3J_{5',6a'}$ = 4.9, ${}^3J_{5',6b'}$ = 9.8, ${}^3J_{5',4'}$ = 9.8 Hz, H-5'), 1.89 (s, 3 H, NAc); 13 C NMR: (150 MHz, CDCl₃): δ = 170.24 (C=O), 138.69 (Cq, Ar-C), 137.83 (Cq, Ar-C), 128.22 (2 C, Ar-C), 128.01 (1 C, Ar-C), 127.92 (4 C, Ar-C), 127.86 (Ar-C), 127.65 (Ar-C), 127.54 (2 C, Ar-C), 128.02 (2 C, Ar-C), 101.55 (ArCH), 101.00 (C-1), 99.95 (C-1'), 78.44 (C-4'), 77.94 (C-3), 77.77 (C-4), 76.68 (C-3'), 74.41 (C-5), 74.03 (CH₂Ar), 73.64 (CH₂Ar), 72.85 (CH₂Ar), 69.05 (C-6), 68.34 (C-6'), 67.21 (C-5'), 63.63 (C-2'), 56.67 (OCH₃), 55.59 (C-2), 23.51 (CH₃CO). HRMS (*ESI-TOF) m/z [M+H]* calcd for C₄3H₄₈N₄O₁₀ 781.3447; found 781.3443.

Methyl 2-acetamido-3-O-benzyl-4,6-O-benzylidene-2-deoxy-β-D-mannopyranosyl-(1→4)-2-acetamido-3,6-di-O-benzyl-2-deoxy-β-D-glucopyranoside (15). A suspension of 14 (50 mg, 0.064 mmol) in DCM (4 mL) was stirred with polymer-bound PPh₃ (475 mg, 1.425 mmol) for 2 d at rt under Ar. H₂O (3 mL) was added followed by 5 minutes of vigorous stirring. Then, the polymer was filtered off with Celite® followed by washing with MeOH (10 mL) and H₂O (10 mL). Concentration of the filtrate afforded the amine intermediate (33 mg, 69%). The residue (33 mg, 0.044 mmol) and a catalytic amount of DMAP were dissolved in pyridine (0.5 mL) under Ar. Ac₂O (25 μL, 0.264 mmol) was added at ice-bath temperature and the solution warmed to rt and stirred for 1.5 h. The reaction was quenched by addition of MeOH (0.1 mL), concentrated and coevaporated with toluene three times in vacuo which gave 15 (29 mg, 83 %) as colorless amorphous solid. $R_f = 0.4$ (EtOAc); $[\alpha]_D^{20}$ -0.3 (c 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ = 7.49-7.27 (m, 20 H, Ar-H), 5.67 (d, 1 H, ${}^{3}J_{NH,2}$ = 8.2 Hz, NH), 5.56 (d, 1 H, ${}^{3}J_{NH',2'} = 9.1$ Hz, NH'), 5.48 (d, 1 H, ArCH), 4.82 (d, 1 H, ${}^{2}J = 11.8$ Hz, CH₂Ar), 4.78 (d, 1 H, ${}^{3}J_{1,2} = 7.7$ Hz, H-1), 4.71-4.67 (m, 2 H, H-2', H-1'), 4.70 (d, 1 H, 2J = 12.2 Hz, CH₂Ar), 4.70 (d, 1 H, 2J = 11.9 Hz, CH₂Ar), 4.63 (d, 1 H, 2J = 11.6 Hz, CH₂Ar), 4.52 (d, 1 H, 2J = 12.1 Hz, CH₂Ar), 4.47 (d, 1 H, 2J = 12.1 Hz, CH₂Ar), 4.15 (dd, 1 H, $^3J_{6a',5'}$ = 4.8, $^{2}J_{6a',6b'} = 10.4 \text{ Hz}, \text{ H-6a'}), 4.08 \text{ (t, 1 H, } ^{3}J_{3,2} = 8.5, ^{3}J_{3,4} = 8.5 \text{ Hz}, \text{ H-3)}, 4.04 \text{ (dd, 1 H, } ^{3}J_{4,3} = 8.3, ^{3}J_{4,5} = 8.3 \text{ Hz}, \text{ H-4)},$ 3.80 (dd, 1 H, ${}^{3}J_{6a.5} = 3.3$, ${}^{2}J_{6a.6b} = 11.1$ Hz, H-6a), 3.73 (dd, 1 H, ${}^{3}J_{6b.5} = 2.8$, ${}^{2}J_{6b.5} = 10.9$ Hz, H-6b), 3.61 (t, 1 $H_{3}^{3}J_{6b',5'} = 10.2, \, ^{2}J_{6b',6a'} = 10.2 \, Hz, \, H-6b'), \, 3.57 \, (t, \, 1 \, H, \, ^{3}J_{4',3'} = 9.3, \, ^{3}J_{4',5'} = 9.3 \, Hz, \, H-4'), \, 3.55 \, (ddd, \, 1 \, H, \, ^{3}J_{5,6b} = 2.7, \, H-6b')$ $^{3}J_{5,6a} = 3.2, ^{3}J_{5,4} = 8.4 \text{ Hz}, \text{ H-5}), 3.50-3.45 (m, 1 H, H-3'), 3.48 (s, 3 H, OMe), 3.36 (ddd, 1 H, <math>^{3}J_{2,1} = 8.1, ^{3}J_{2,3} = 8.1, ^{3}J_{2,3} = 8.1, ^{3}J_{2,4} = 8.1, ^{3}J_$ ${}^{3}J_{2,NH} = 8.1 \text{ Hz}, H-2), 3.15 \text{ (ddd, 1 H, } {}^{3}J_{5',6a'} = 5.0, {}^{3}J_{5',6b'} = 9.7, {}^{3}J_{5',a'} = 9.7 \text{ Hz}, H-5'), 1.92 \text{ (s, 3 H, NAc), 1.90 (s, 3 H, NAc)}$ NAc); ¹³C NMR: (150 MHz, CDCl₃): δ = 170.51 (2 C, C=O), 137.89 (Cq, Ar-C), 129.03 (Ar-C), 128.56 (2 C, Ar-C), 128.43 (2 C, Ar-C), 128.39 (2 C, Ar-C), 128.23 (2 C, Ar-C), 128.00 (1 C, Ar-C), 127.90 (2 C, Ar-C), 127.72 (2 C, Ar-C) C), 127.7 (2 C, Ar-C), 126.08 (2 C, Ar-C), 101.68 (ArCH), 100.77 (C-1), 99.61 (C-1'), 78.69 (C-4'), 78.58 (C-3), 76.70 (C-4)*, 75.64 (C-3'), 74.21 (C-5), 73.97 (CH_2Ar), 73.55 (CH_2Ar), 71.44 (CH_2Ar), 68.63 (C-6), 68.63 (C-6'), 67.03 (C-5'), 56.79 (2 C, C-2, OCH₃), 50.42 (C-2'), 23.56 (CH₃CO), 23.39 (CH₃CO) ppm. HRMS ($^{+}$ ESI-TOF) m/z $[M+H]^+$ calcd for $C_{45}H_{52}N_2O_{11}$ 797.364; found 797.3645.

Methyl 2-acetamido-3-*O*-benzyl-2-deoxy-β-D-mannopyranosyl-(1 \rightarrow 4)-2-acetamido-3,6-di-*O*-benzyl-2-deoxy-β-D-glucopyranoside (16). Trifluoroacetic acid (290 μL) was added to a solution of 15 (26 mg, 0.033 mmol) in DCM (1150 μL) externally cooled with an ice/NaCl-bath below 0 °C under Ar and stirred for 25 min. The solution was diluted with DCM (5 mL), followed by addition of DOWEX anion-exchange resin (HCO₃-form, 10 g). The resin was filtered off and washed with DCM (20 mL). The filtrate was concentrated and the residue was purified by column chromatography (EtOAc-MeOH 95:5) to give 16 (23 mg, 99%) as colorless crystals, m.p.

113-115°C; $R_f = 0.25$ (EtOAc-MeCN 4:1), $[\alpha]_D^{20}$ -45.8 (c 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃): $\delta = 7.35-7.26$ (m, 15 H, Ar-H), 6.03 (d, 1 H, ${}^{3}J_{NH',2'}$ = 9.7 Hz, NH'), 5.66 (d, 1 H, ${}^{3}J_{NH,2}$ = 8.3 Hz, NH), 4.83 (d, 1 H, ${}^{2}J$ = 11.8 Hz, CH₂Ar), 4.76 (d, 1 H, 2J = 11.2 Hz, CH₂Ar), 4.65 (ddd, 1 H, $^3J_{2',1'}$ = 1.4, $^3J_{2',NH'}$ = 9.8, $^3J_{2',3'}$ = 4.5 Hz, H-2'), 4.65 (d, 1 H, 2J = 11.9 Hz, CH₂Ar), 4.61 (d, 1 H, 2J = 11.9 Hz, CH₂Ar), 4.60 (d, 1 H, $^3J_{1',2'}$ = 1.4 Hz, H-1'), 4.60 (d, 1 H, $^3J_{1,2}$ = 7.0 Hz, H-1), 4.44 (d, 1 H, ${}^{2}J$ = 12.0 Hz, CH₂Ar), 4.29 (d, 1 H, ${}^{2}J$ = 11.4 Hz, CH₂Ar), 4.04 (t, 1 H, ${}^{3}J_{3,4}$ = 8.4, ${}^{3}J_{3,2}$ = 8.4 Hz, H-3), 3.90 (t, 1 H, ${}^{3}J_{4,5} = 8.6$, ${}^{3}J_{4,5} = 8.6$ Hz, H-4), 3.72 (dd, 1 H, ${}^{3}J_{6a',5'} = 2.8$, ${}^{2}J_{6a',6b'} = 12.0$ Hz, H-6a'), 3.71 (d, 1 H, ${}^{3}J_{6,5} = 3.8 \text{ Hz}, \text{ H-6a}, 3.71 \text{ (app d, 1 H, } {}^{3}J_{6,5} = 3.8 \text{ Hz}, \text{ H-6b}, 3.61 \text{ (dd, 1 H, } {}^{3}J_{6b',5'} = 4.8, {}^{2}J_{6b',6a'} = 12.1 \text{ Hz}, \text{ H-6b'},$ 3.57 (ddd, 1 H, ${}^{3}J_{2,1} = 7.5$, ${}^{3}J_{2,3} = 8.4$, ${}^{3}J_{2,NH} = 8.4$ Hz, H-2), 3.54 (t, 1 H, ${}^{3}J_{4',3'} = 9.5$, ${}^{3}J_{4',5'} = 9.5$, H-4'), 3.53 (ddd, 1 H, $^{3}J_{5.6a} = 3.3$, $^{3}J_{5.4} = 8.8$, $^{3}J_{5.6b} = 3.3$ Hz, H-5), 3.45 (s, 3 H, OCH₃), 3.21 (dd, 1 H, $^{3}J_{3',2'} = 4.2$, $^{3}J_{3',4'} = 9.4$ Hz, H-3'), 3.10 (ddd, 1 H, ${}^{3}J_{5',6g'}$ = 3.1, ${}^{3}J_{5',6b'}$ = 4.7, ${}^{3}J_{5',4'}$ = 9.8 Hz, H-5'), 1.93 (s, 3 H, CH₃CO), 1.82 (s, 3 H, CH₃CO); ${}^{13}C$ NMR: (150 MHz, CDCl₃): 170.68 (C=O), 170.51 (C=O), 138.50 (Cq, Ar), 137.78 (Cq, Ar), 137.53 (Cq, Ar-C), 128.52 (6 C, Ar-C), 128.22 (2 C, Ar-C), 127.97 (1 C, Ar-C), 127.95 (2 C, Ar-C), 127.87 (1 C, Ar-C), 127.82 (1 C, Ar-C), 127.62 (2 C, Ar-C) C), 101.10 (C-1), 98.80 (C-1'), 79.52 (C-3'), 78.71 (C-4), 76.35 (C-3), 76.16 (C-5'), 74.13 (C-5), 74.02 (CH_2Ar), 73.49 (CH_2Ar), 70.68 (CH_2Ar), 68.76 (C-6), 66.74 (C-4), 61.83 (C-6), 56.58 (OCH_3), 55.69 (C-2), 49.11 (C-2), 23.48 (CH₃CO), 23.26 (CH₃CO) ppm. HRMS ($^{+}$ ESI-TOF) m/z [M+H] $^{+}$ calcd for C₃₈H₄₈N₂O₁₁ 709.3331; found 709.3332.

2-acetamido-3-O-benzyl-2-deoxy-4,6-O-[1-(methoxycarbonyl)ethylidene]-B-D-mannopyranosyl-Methyl (1->4)-2-acetamido-3,6-di-O-benzyl-2-deoxy-β-D-glucopyranoside (17). A solution of methyl pyruvate (100 μL of a stock solution containing 100 µL in 2.4 mL MeCN) was added to an ice-cold solution of 16 in MeCN (1150 μL) under Ar and stirred for 10 min. TMSOTf (19 μL, 0.105 mmol) was then added through a septum and the solution was stirred at ice-bath temperature for 4.5 h, when TLC showed consumption of the starting material. The solution was diluted with EtOAc and washed with satd ag NaHCO₃-solution. The organic phase was dried (MgSO₄), filtered and concentrated in vacuo. The product was purified by column chromatography (toluene-EtOAc 1:1) to give 10 mg (75 %) of **17** as syrup; $R_f = 0.43$ (EtOAc-MeCN 4:1); $[\alpha]_D^{20}$ -16 (c 0.15, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ = 7.32-7.20 (m, 15 H, Ar-H), 5.67 (br s, 1 H, NH), 5.50 (br s, 1 H, NH), 4.68 (d, 1 H, 2J = 11.6 Hz, CH₂Ar), 4.65 (d, 1 H, ${}^{3}J_{1,2}$ = 7.4 Hz, H-1), 4.62 (d, 1 H, ${}^{2}J$ = 12.3 Hz, CH₂Ar), 4.58 (d, 1 H, ${}^{2}J$ = 12.1 Hz, CH₂Ar), 4.57 (d, 1 H, ${}^{2}J$ = 11.8 Hz, CH₂Ar), 4.553 (d, 1 H, ${}^{3}J_{1',2'}$ = 1.3 Hz, H-1'), 4.547 (ddd, 1 H, ${}^{3}J_{2',1'}$ = 1.4, ${}^{3}J_{2',3'}$ = 5.0, $^{3}J_{2',NH'}$ = 9.9 Hz, H-2'), 4.51 (d, 1 H, ^{2}J = 11.6 Hz, CH₂Ar), 4.36 (d, 1 H, ^{2}J = 11.8 Hz, CH₂Ar), 3.93-3.88 (m, 2 H, H-3, H-4'), 3.81 (dd, 1 H, ${}^{3}J_{6a.5} = 5.0$, ${}^{2}J_{6a.6b} = 10.6$ Hz, H-6a), 3.75 (s, 3 H, CO₂CH₃), 3.73 (dd, 1 H, ${}^{3}J_{6a'.5'} = 3.4$, ${}^{2}J_{6a'.6b'} = 3.4$ 10.9 Hz, H-6a'), 3.63 (dd, 1 H, ${}^{3}J_{6b',5'} = 3.1$, ${}^{2}J_{6b',6a'} = 11.0$ Hz, H-6b'), 3.49 (dd, 1 H, ${}^{3}J_{6b,5} = 10.3$, ${}^{2}J_{6b,6a} = 10.7$ Hz, H-6b), 3.47-3.40 (m, 2 H, H-5', H-4), 3.39 (s, 3 H, OCH₃), 3.36-3.30 (m, 2 H, H-2, H-3'), 2.97 (ddd,1 H, ${}^{3}J_{5,6a}$ = 4.8, $^{3}J_{5,6b}$ = 10.3, $^{3}J_{5,4}$ = 10.3 Hz, H-5), 1.81 (s, 6 H, 2 x CH₃CO), 1.48 (s, 3 H, CH₃); 13 C NMR: (150 MHz, CDCl₃): δ = 170.69 (2 x C=O), 170.10 (C=O, pyr), 138.63 (Cq, Ar-C), 138.21 (Cq, Ar-C 137.93 (Cq, Ar-C), 128.50 (2 C, Ar-C), 128.38 (2 C, Ar-C), 128.28 (2 C, Ar-C), 127.94 (1 C, Ar-C), 127.87 (2 C, Ar-C), 127.65 (1 C, Ar-C), 127.53 (3 C, Ar-C) C), 127.32 (2 C, Ar-C), 100.86 (C-1), 99.51 (C-1'), 99.11 (Cq, CCO₂CH₃), 78.67 (C-3), 76.47 (C-4'), 75.25 (C-4), 75.10 (C-3'), 74.26 (C-5'), 73.85 (CH₂Ar), 73.52 (CH₂Ar), 71.3 (CH₂Ar), 68.72 (C-6'), 66.43 (C-5), 65.18 (C-6), 56.72 (OCH₃), 56.24 (C-2), 52.76 (CO₂CH₃), 50.34 (C-2'), 25.54 ($CH_3CCO_2CH_3$), 23.36 (2 x CH₃CO) ppm. HRMS (*ESI-TOF) m/z [M+H]* calcd for C₄₂H₅₂N₂O₁₃, 793.3542; found 793.3541.

Methyl 2-acetamido-2-deoxy-4,6-O-[1-(methoxycarbonyl)ethylidene]-β-D-mannopyranosyl-(1 \rightarrow 4)-2-acetamido-2-deoxy-β-D-glucopyranoside (18). A suspension of 17 (5 mg, 0.007 mmol) in dry MeOH (1.5 mL) containing 10% Pd-carbon catalyst was evacuated and flushed with Ar four times. The atmosphere was switched to H₂ and the mixture was stirred for 5 h at rt, when TLC showed complete consumption of the starting material. The catalyst was filtered off over Celite® and washed with MeOH several times (10 mL in total). The filtrate was concentrated *in vacuo* to give 18 (3.2 mg, 89 %) as amorphous solid; $R_f = 0.75$ (CHCl₃-

MeOH-H₂O 3:2.5:0.5); ¹H NMR (600 MHz, MeOD): δ = 4.85 (d, 1 H, ${}^{3}J_{1',2'}$ = 1.7 Hz, H-1'), 4.61 (dd, 1 H, ${}^{3}J_{2',1'}$ = 1.6, ${}^{3}J_{2',3'}$ = 4.6 Hz, H-2'), 4.30 (d, 1 H, ${}^{3}J_{1,2}$ = 8.5 Hz, H-1), 4.00 (dd, 1 H, ${}^{3}J_{6\alpha',5'}$ = 5.0, ${}^{2}J_{6\alpha',6b'}$ = 10.5 Hz, H-6a'), 3.85 (dd, 1 H, ${}^{3}J_{3',2'}$ = 4.7, ${}^{3}J_{3',4'}$ = 9.8 Hz, H-3'), 3.82 (s, 3 H, CO₂CH₃), 3.81 (dd, 1 H, ${}^{3}J_{6a,5}$ = 2.1, ${}^{2}J_{6a,6b}$ = 12.3 Hz, H-6a), 3.77 (dd, 1 H, ${}^{3}J_{6b',5'}$ = 10.5, ${}^{3}J_{6b',6a'}$ = 10.5 Hz, H-6b'), 3.72 (dd, 1 H, ${}^{3}J_{6b,5}$ = 4.0, ${}^{2}J_{6b,6a}$ = 12.2 Hz, H-6b), 3.66 (dd, 1 H, ${}^{3}J_{2,3}$ = 10.3, ${}^{3}J_{2,1}$ = 8.5 Hz, H-2), 3.64 (dd, 1 H, ${}^{3}J_{4,5}$ = 9.2, ${}^{3}J_{4,3}$ = 9.2 Hz, H-4), 3.57 (dd, 1 H, ${}^{3}J_{4',5'}$ = 10.0, ${}^{3}J_{4',3'}$ = 10.0 Hz, H-4'), 3.53 (dd, 1 H, ${}^{3}J_{3,4}$ = 8.9, ${}^{3}J_{3,2}$ = 10.4 Hz, H-3), 3.44 (s, 3 H, OCH₃), 3.37 (dt, 1 H, ${}^{3}J_{5',6a'}$ = 5.0, ${}^{3}J_{5',6a'}$ = 5.0, ${}^{3}J_{5',6a'}$ = 10.0, ${}^{3}J_{5',4'}$ = 10.0 Hz, H-5'), 3.28 (ddd, 1 H, ${}^{3}J_{5,6a}$ = 2.0, ${}^{3}J_{5,6b}$ = 3.9, ${}^{3}J_{5,4}$ = 9.7 Hz, H-5), 2.01 (s, 3 H, CH₃CO), 1.96 (s, 3 H, CH₃CO), 1.48 (s, 3 H, CH₃); 13 C NMR: (150 MHz, MeOD): 174.72 (NHC=O), 173.63 (NHC=O), 171.96 (OC=O), 103.47 (C-1), 101.72 (C-1'), 100.72 (Cq, CCO₂CH₃), 80.64 (C-4), 76.34 (C-5), 75.96 (C-4'), 74.08 (C-3), 70.96 (C-3'), 68.38 (C-5'), 65.65 (C-6'),61.63 (C-6), 57.00 (OCH₃), 56.97 (C-2), 54.83 (C-2'), 53.17 (CO₂CH₃), 25.86 (CH₃), 22.89 (CH₃CO), 22.66 (CH₃CO) ppm. HRMS (*ESI-TOF) m/z [M+H]⁺ calcd for C₂₁H₃₄N₂O₁₃, 523.2134; found 523.2134.

2-acetamido-2-deoxy-4,6-O-(1-carboxyethylidene)- β -D-mannopyranosyl-(1 \rightarrow 4)-2-acetamido-2-Methyl deoxy-β-D-glucopyranoside sodium salt (19). A solution of 18 (3.2 mg, 0.007 mmol) in MeOH (1.5 mL) and 0.2 M NaOH (530 μl) was stirred for 2 h at rt when TLC indicated complete consumption of the starting material. The pH-value was adjusted to neutral with Dowex cation exchange resin (H⁺ form). The resin was filtered off and washed with water (10 mL). The solution was frozen and lyophilized to afford 2.5 mg (77 %) of product. Final purification was done by filtration on BioGel P-2 (5% EtOH in water) to give 2.2 mg (68 %) of 19 as colorless solid ; $[\alpha]_D^{20}$ -38.9 (c 0.1, H₂O); ¹H NMR (600 MHz, D₂O): δ = 4.94 (d, 1 H, ³ $J_{1',2'}$ = 1.8 Hz, H-1'), 4.61 $(dd, 1 H, {}^{3}J_{2',1'} = 1.7, {}^{3}J_{2',3'} = 4.7 Hz, H-2'), 4.43 (d, 1 H, {}^{3}J_{1,2} = 8.0 Hz, H-1), 4.04 (dd, 1 H, {}^{3}J_{6a',5'} = 5.3, {}^{2}J_{6a',6b'} = 10.7)$ Hz, H-6a'), 3.99 (dd, 1 H, ${}^{3}J_{3',2'} = 4.7$, ${}^{3}J_{3',4'} = 10.0$ Hz, H-3'), 3.87 (dd, 1 H, ${}^{3}J_{6a,5} = 2.2$, ${}^{2}J_{6a,6b} = 12.3$ Hz, H-6a), 3.75 (t, 1 H, ${}^{3}J_{6b',5'} = 10.9$, ${}^{2}J_{6b',6a'} = 10.9$ Hz, H-6b'), 3.73 (dd, 1 H, ${}^{3}J_{6b,5} = 5.2$, ${}^{2}J_{6b,6a} = 12.2$ Hz, H-6b), 3.70 (dd, 1 H, $^{3}J_{2,1} = 8.1$, $^{3}J_{2,3} = 10.2$ Hz, H-2), 3.69 (t, 1 H, $^{3}J_{4,3} = 9.1$, $^{3}J_{4,5} = 9.1$ Hz, H-4), 3.65 (dd, 1 H, $^{3}J_{3,4} = 9.0$, $^{3}J_{3,2} = 10.1$ Hz, H-3), 3.63 (t, 1 H, ${}^{3}J_{4',5'}$ = 10.1, ${}^{3}J_{4',3'}$ = 10.1 Hz, H-4'), 3.50 (s, 3H, OCH₃), 3.49 (ddd, 1 H, ${}^{3}J_{5,6a}$ = 2.4, ${}^{3}J_{5,6b}$ = 5.0, ${}^{3}J_{5,4} = 9.2 \text{ Hz}, \text{H-4}, 3.45 \text{ (ddd, 1 H, } {}^{3}J_{5',66'} = 5.0, {}^{3}J_{5',6b'} = 10.1, {}^{3}J_{5',4'} = 10.1 \text{ Hz}, \text{H-5'}, 2.08 \text{ (s, 3 H, CH}_{3}CO), 2.04 \text{ (s, 3 H, CH}_{3}CO)}$ H, CH₃CO), 1.47 (s, 3 H, CH₃); ¹³C NMR: (150 MHz, D₂O): 176.30 (C=O), 176.25 (C=O), 175.50 (C=O), 102.69 (C-O) 1), 102.67 (Cq, CCO₂), 100.65 (C-1'), 79.65 (C-4), 75.31 (C-5), 74.71 (C-4'), 73.11 (C-3), 70.16 (C-3'), 67.65 (C-5'), 64.87 (C-6'), 60.95 (C-6), 57.90 (OCH₃), 56.08 (C-2), 54.31 (C-2'), 25.49 (CH₃CCO₂Na), 22.97 (CH₃CO), 22.84 (CH₃CO) ppm. HRMS ($^{+}$ ESI-TOF) m/z [M+Na] $^{+}$ calcd for C₂₀H₃₂N₂NaO₁₃ 531.1797, found 531.1798.

Acknowledgements

The authors a grateful to Dr. Markus Blaukopf for support with NMR measurements and HPLC separations as well as Marlene Neulinger and Elise Loppinet for technical support. Financial support by the Austrian Science Fund FWF to C. Schäffer and P. Kosma (project P27374-B22) is gratefully acknowledged.

Supplementary Material

¹H and ¹³C NMR spectra of novel compounds can be found in the supporting information.

References

Page 12 [©]AUTHOR(S)

1. Hager, F. F.; Sützl, L.; Stefanovic, C.; Blaukopf, M; Schäffer, C. *Int. J. Mol. Sci.* **2019**, *20*, E4929. https://doi.org/10.3390/ijms20194929

- 2. Pereira, C. L.; Geissner, A.; Anish, C.; Seeberger, P. H. *Angew. Chem. Int. Ed.* **2015**, *54*, 10016-10019. https://doi.org/10.1002/anie.201504847
- 3. Geissner, A.; Pereira, C. L.; Leddermann, M.; Anish, C.; Seeberger, P. H. *ACS Chem. Biol.* **2016**, *11*, 335-344. https://doi.org/10.1021/acschembio.5b00768
- Yang, F. L.; Liao, P. C.; Chou, J. C.; Tsai, K. C.; Yang, A. S.; Sheu, F.; Lin, T. L.; Hsieh, P. F.; Wang, J. T.; Hua, K. F.; Wu, S. H. J. Biol. Chem. 2011, 286, 21041-21051. https://doi.org/10.1074/jbc.M111.222091
- 5. Kenyon, J. J.; Marzaioli, A. M.; Hall, R. M.; De Castro, C. *Glycobiology* **2014**, *24*, 554-563. https://doi.org/10.1093/glycob/cwu024
- Kenyon, J. J.; Speciale, I.; Hall, R. M.; De Castro, C. *Carbohydr. Res.* 2016, 434, 12-17. https://doi.org/10.1016/j.carres.2016.07.016
- 7. Baumann, H.; Tzianabos, A. O.; Mallory, B. C.; Carey, V. J.; Kasper, D. L.; Jennings, H. J. *Biochemistry*, **1992**, *31*, 4081-4089.
 - https://doi.org/10.1021/bi00131a026
- 8. Schäffer, C.; Messner, P. *Microbiol.* **2005**, *151*, 643-651. https://doi.org/10.1099/mic.0.27749-0
- 9. Sleytr, U. B.; Schuster, B.; Egelseer, E. M.; Pum, D. *FEMS Microbiol. Rev.* **2014**, *38*, 823-864. https://doi.org/10.1111/1574-6976.12063
- 10. Sychanta, D.; Chapman, R. N.; Bamford, N. C.; Boons, G.-J.; Howell, P.L.; Clarke, A. J. *Biochemistry* **2018**, *57*, 1949-1953.
 - https://doi.org/10.1021/acs.biochem.8b00060
- 11. Chapman, R. N.; Liu, L.; Boons, G.-J. *J. Am. Chem. Soc.* **2018**, *140*, 17079-17085. https://doi.org/10.1021/jacs.8b08857
- 12. Mesnage, S.; Fontaine, T.; Mignot, T.; Delepierre, M.; Mock, M.; Fouet, A. *EMBO J.* **2000**, 19, 4473-4484. https://doi.org/10.1093/emboj/19.17.4473
- 13. Forsberg, L. S.; Choudhury, B.; Leoff, C.; Marston, C. K.; Hoffmaster, A. R.; Saile, E.; Quinn; C. P.; Kannenberg, E. L.; Carlson, R. W. *Glycobiology*, **2011**, *21*, 934-948. https://doi.org/10.1093/glycob/cwr026
- 14. Schäffer, C.; Müller, N.; Mandal, P. K.; Christian, R.; Zayni, S.; Messner, P. *Glycoconj. J.* **2000**, *17*, 681-690. https://doi.org/10.1023/A:1011062302889
- 15. Ilk, N.; Kosma, P.; Puchberger, M.; Egelseer, E. M.; Mayer, H. F.; Sleytr, U. B.; Sára, M. *J. Bacteriol.* **1999**, *181*, 7643-7646.
 - https://doi.org/10.1128/JB.181.24.7643-7646.1999
- 16. Blackler, R.; López-Guzmán, A., Hager, F. F.; Janesch, B.; Martinz, G.; Gagnon, S. M. L.; Haji-Ghassemi, O.; Kosma, P.; Messner, P.; Schäffer, C.; Evans, S. V. *Nat. Commun.* **2018**, *9*, 3120. https://doi.org/10.1038/s41467-018-05471-3
- 17. Arihara, R., Kakita, K., Yamada, K., Nakamura, S., Hashimoto, S. *J. Org. Chem.* **2015**, *80*, 4278-4288. https://doi.org/10.1021/acs.joc.5b00139
- 18. Gagarinov, I. A.; Fang, T.; Liu, L.; Srivastava, A. D.; Boons, G.-J. *Org. Lett.* **2015**, *17*, 928-931. https://doi.org/10.1021/acs.orglett.5b00031
- 19. Classon, B.; Garegg, P. J.; Oscarson, S.; Tiden, A. K.; Carbohydr. Res. 1992, 216, 187-196.

https://doi.org/10.1016/0008-6215(92)84161-K

20. Gridley, J. J.; Osborn, H. M. I. J. Chem. Soc. Perkin Trans 1, 2000, 1471-1491.

https://doi.org/10.1039/a909165c

21. Kunz, H.; Günther, W. Angew. Chem. 1988, 100, 1118-1119.

https://doi.org/10.1002/ange.19881000822

22. David, S.; Malleron, A.; Dini, C. Carbohydr. Res. 1989, 188, 193-200.

https://doi.org/10.1016/0008-6215(89)84070-4

- 22. Van Dorst, J. A. L. M.; Voskamp, A. F.; Kamerling, J. P.; Vliegenthart, J. F. G. *Liebigs Ann. Chem.* **1997**, 1227-1233.
- 23. Ogawa, S.; Hirai, K.; Odagiri, T.; Matsunaga, N.; Yamazaki, T.; Nakajima, A. *Eur. J. Org. Chem.* **1998**, 1099-1109.
 - https://doi.org/10.1002/(SICI)1099-0690(199806)1998:6<1099::AID-EJOC1099>3.0.CO;2-R
- 24. Cromer, R.; Spohr, U.; Khare, D. P.; LePendu, J.; Lemieux, R. U. *Can. J. Chem.* **1992**, *70*, 1511-1522. https://doi.org/10.1139/v92-187
- 25. Yang, Z.; Lin, W.; Yu, B. *Carbohydr. Res.* **2000**, *329*, 879-884. https://doi.org/10.1016/S0008-6215(00)00242-1
- 26. Basu, N.; Maity, S. K.; Roy, S.; Singha, S.; Ghosh, R. *Carbohydr. Res.* **2011**, *346*, 534-539. https://doi.org/10.1016/j.carres.2011.01.003
- 27. Sveshnikov, N. N.; Nelson, J. H. *Magn. Res. Chem.* **1997**, *35*, 209-212. https://doi.org/10.1002/(SICI)1097-458X(199703)35:3<209::AID-OMR40>3.0.CO;2-6
- 28. Charette, A. B.; Grenon, M. *Can. J. Chem.* **2001**, *79*, 1694-1703. https://doi.org/10.1139/v01-150
- 29. Tona, V.; Maryasin, B.; de la Torre, A.; Sprachmann, J.; González, L.; Maulide, N. *Org. Lett.* **2017**, *19*, 2662-2665.
 - https://doi.org/10.1021/acs.orglett.7b01004
- 30. Kaiser, D.; Maulide, N. J. Org. Chem. **2016**, *81*, 4421-4428.

https://doi.org/10.1021/acs.joc.6b00675

31. Trattnig, N.; Blaukopf, M.; Bruxelle, J.-F.; Pantophlet, R.; Kosma, P. *J. Am. Chem. Soc.* **2019**, *141*, 7946-7954.

https://doi.org/10.1021/jacs.9b02872

33. Cai, Y.; Ling, C.-C.; Bundle, D. R. J. Org. Chem. **2009**, 74, 580-589.

https://doi.org/10.1021/jo801927k

34. Lehnhoff, S.; Ugi, I. Heterocycles **1995**, 40, 801-808.

https://doi.org/10.3987/COM-94-S69

- 32. Ziegler, T.; Eckhardt, E.; Herold, G. *Tetrahedron Lett.* **1992**, *33*, 4413-4416. https://doi.org/10.1016/S0040-4039(00)60097-7
- 33. Schüle, G.; Ziegler, T. *Liebigs Ann.* **1996**, 1599-1607.
 - https://doi.org/10.1002/jlac.199619961016
- 34. Garegg, P. J.; Jansson, P. E.; Lindberg, B.; Lindh, F.; Lönngren, J.; Kvarnström, I.; Nimmich, W. *Carbohydr. Res.* **1980**, *78*, 127.
 - https://doi.org/10.1016/S0008-6215(00)83666-6
- 35. Jansson, P. E.; Lindberg, J.; Widmalm, G. *Acta Chem. Scand.* **1993**, *47*, 711-715. https://doi.org/10.3891/acta.chem.scand.47-0711
- 39. Takano, T.; Harada, Y.; Nakatsubo, F.; Koji, M. Cell. Chem. Technol. 1990, 24, 333-341.

https://doi.org/10.1016/0008-6215(90)80036-3

This paper is an open access article distributed under the terms of the Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/)

Page 15 ©AUTHOR(S)