## **Supplementary Material**

# Synthesis of bis-oxathiaaza[3.3.3]propellanes *via* nucleophilic addition of (1,ω-alkanediyl)bis(N'-organylthioureas) on dicyanomethylene-1,3-indanedione

Alaa A. Hassan,<sup>a</sup>\* Kamal M. A. El-Shaieb,<sup>a</sup> Amal S. Abd El-Aal,<sup>a</sup> Stefan Bräse,<sup>b</sup> and Martin Nieger<sup>c</sup>

<sup>a</sup> Chemistry Department, Faculty of Science, Minia University, 61519 El-Minia, Egypt. <sup>b</sup> Institute of Organic Chemistry, Karlsruhe Institute of Technology, Fritz-Haber-Weg 6, 76131 Karlsruhe, Germany.

<sup>c</sup> Laboratory of Inorganic Chemistry, Department of Chemistry, University of Helsinki P.O Box 55 (A. I. Virtasen aukio 1), 00014 Helsinki, Finland E-mail: alaahassan2001@mu.edu.eg

Single crystal X-ray diffraction study was carried out on an Agilent Super Nova diffractometer at 173 K with EOS-detector and MoK $\alpha$  radiation ( $\lambda = 0.71073$  Å). Direct Methods (SHELXS-97 [Sheldrick, G.M., *Acta Crystallogr.*, **2008**. *A64*, 112-122.]) were used for structure solution and refinement was carried out using SHELXL-2013 [Sheldrick, G.M., *Acta Crystallogr.*, **2008**. *A64*, 112-122.] (full-matrix least-squares on F<sup>2</sup>). Hydrogen atoms were localized by difference Fourier Synthesis map and refined using a riding model [H(N) free]. A semi-empirical absorption correction and an extinction correction were applied. Compound **7a**: C<sub>40</sub>H<sub>26</sub>N<sub>8</sub>O<sub>4</sub>S<sub>2</sub> · 2 (C<sub>3</sub>H<sub>7</sub>NO), Mr = 893.00 gmol<sup>-1</sup>, colorless plates, crystal size 0.30 × 0.02 × 0.10 mm, triclinic, P-1 (no. 2), a = 9.7515 (5) Å, b = 9.8024 (8) Å, c = 13.4038 (8) Å,  $\alpha = 90.279$  (6)°,  $\beta = 102.516$  (5)°,  $\gamma = 118.268$  (7)°, V = 1092.98(14) Å<sup>3</sup>, Z = 1, D<sub>calcd</sub> = 1.357 Mg m<sup>-3</sup>, F(000)= 466,  $\mu = 0.184$  mm<sup>-1</sup>, T = 173 K, 7638 measured reflections ( $2\theta_{max} = 55^{\circ}$ ), 4954 independent reflections (R<sub>int.</sub> = 0.016) 298 parameters, 2 restraints, R1 ( for 3998 I > 2 $\sigma$ (I)) = 0.044, wR<sup>2</sup> (for all data ) = 0.107, S = 1.03, largest diff. peak and hole = 0.35 eA<sup>-3</sup>, -0.300 eA<sup>-3</sup>.

#### Supplementary data

Crystallographic data for the structure reported in this work have been deposited with Cambridge Crystallographic Data Center on supplementary publication no CCDC-1417181 Copies of the data

can be obtained free of charge on publication to the Director, CCDC, 12 Union Road, Cambridge CB2 IEZ,UK (fax:+44(1223)336033:e-mail: <u>deposit@ccdc.cam.ac.uk.</u>



**Figure 1.** Molecular structure of **7a** in the crystal (displacement parameters are drawn at 50% probability level). The crystallographic numbering does not reflect the systematic IUPAC numbering (SB651\_HY).



Figure 2. Crystal packing of 7a (in the voids are DMF).

	Z = 1
$C_{40}H_{26}N_8O_4S_2{\cdot}2(C_3H_7NO)$	
$M_r = 893.00$	F(000) = 466
Triclinic, P-1 (no.2)	$D_{\rm x} = 1.357 {\rm ~Mg} {\rm m}^{-3}$
<i>a</i> = 9.7515 (5) Å	Mo $K\alpha$ radiation, $\lambda$ = 0.71073 Å
b = 9.8024 (8) Å	Cell parameters from 2581 reflections
<i>c</i> = 13.4038 (8) Å	$\theta = 2.4 - 29.6^{\circ}$
$\alpha = 90.279 \ (6)^{\circ}$	$\mu = 0.18 \text{ mm}^{-1}$
$\beta = 102.516 \ (5)^{\circ}$	<i>T</i> = 173 K
$\gamma = 118.268 \ (7)^{\circ}$	Plates, colourless
$V = 1092.98 (14) \text{ Å}^3$	0.30 × 0.20 × 0.10 mm

## **Table 2.** Data collection for **7a**

Agilent, SuperNova, Single source at offset, Eos diffractometer	4954 independent reflections
Radiation source: SuperNova (Mo) X-ray Source	3998 reflections with $I > 2\sigma(I)$
Detector resolution: 32.0214 pixels mm <sup>-1</sup>	$R_{\rm int} = 0.016$
rotation in ω, 1° scans	$\theta_{\text{max}} = 27.5^{\circ}, \ \theta_{\text{min}} = 2.4^{\circ}$
Absorption correction: multi-scan <i>CrysAlis PRO</i> , Agilent Technologies. Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.	$h = -12 \rightarrow 12$
$T_{\min} = 0.858, T_{\max} = 1.000$	$k = -12 \rightarrow 12$
7638 measured reflections	$l = -13 \rightarrow 17$

#### Table 3. Refinement data for 7a

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.044$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.107$	$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0373P)^{2} + 0.4217P] \text{ where } P$ = $(F_{o}^{2} + 2F_{c}^{2})/3$

<i>S</i> = 1.03	$(\Delta/\sigma)_{\rm max} = 0.001$
4954 reflections	$\Delta \rangle_{\text{max}} = 0.31 \text{ e} \text{ Å}^{-3}$
298 parameters	$\Delta$ <sub>min</sub> = -0.30 e Å <sup>-3</sup>
2 restraints	Extinction correction: <i>SHELXL</i> , Fc <sup>*</sup> =kFc[1+0.001xFc <sup>2</sup> $\lambda^3$ /sin(2 $\theta$ )] <sup>-1/4</sup>
Primary atom site location: structure-invariant direct methods	Extinction coefficient: 0.0088 (15)

**Table 4.** Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(\text{\AA}^2)$  for **7a** 

	x	у	z	$U_{\rm iso}$ */ $U_{\rm eq}$
<b>S</b> 1	0.50133 (5)	0.68657 (5)	0.32202 (3)	0.03351 (14)
C2	0.66191 (18)	0.79297 (18)	0.43225 (12)	0.0266 (3)
N3	0.80959 (15)	0.84120 (15)	0.41067 (10)	0.0259 (3)
C4	0.81228 (18)	0.83774 (18)	0.30497 (12)	0.0242 (3)
C5	0.9020 (2)	0.7625 (2)	0.27211 (13)	0.0288 (4)
C6	1.0661 (2)	0.8146 (2)	0.29775 (16)	0.0419 (5)
H6	1.1392	0.9092	0.3417	0.050*
C7	1.1193 (3)	0.7243 (3)	0.2572 (2)	0.0590 (6)
H7	1.2311	0.7573	0.2745	0.071*
C8	1.0146 (3)	0.5867 (3)	0.1920 (2)	0.0611 (6)
H8	1.0552	0.5265	0.1663	0.073*
C9	0.8521 (3)	0.5367 (2)	0.16434 (17)	0.0482 (5)
H9	0.7798	0.4435	0.1188	0.058*
C10	0.7968 (2)	0.6265 (2)	0.20494 (14)	0.0331 (4)
C11	0.6311 (2)	0.6013 (2)	0.18278 (14)	0.0333 (4)
011	0.51103 (17)	0.49477 (16)	0.12920 (12)	0.0537 (4)

C12	0.63788 (18)	0.74375 (19)	0.23738 (12)	0.0267 (3)
C13	0.62873 (19)	0.85742 (19)	0.16584 (12)	0.0280 (4)
C14	0.76816 (19)	0.99725 (19)	0.19299 (12)	0.0262 (3)
N14	0.81327 (19)	1.13119 (18)	0.15477 (12)	0.0359 (4)
H14A	0.914 (2)	1.207 (2)	0.1778 (17)	0.054*
H14B	0.747 (2)	1.137 (3)	0.1019 (14)	0.054*
015	0.87662 (12)	0.99695 (12)	0.27543 (8)	0.0269 (3)
C16	0.4923 (2)	0.8296 (2)	0.08880 (14)	0.0346 (4)
N17	0.3808 (2)	0.8088 (2)	0.02660 (14)	0.0505 (5)
N18	0.64954 (17)	0.82089 (17)	0.52141 (11)	0.0337 (3)
C19	0.4960 (2)	0.7639 (2)	0.54239 (14)	0.0350 (4)
C20	0.4663 (3)	0.8708 (3)	0.58898 (16)	0.0484 (5)
H20	0.5455	0.9780	0.6030	0.058*
C21	0.3219 (3)	0.8216 (3)	0.61498 (19)	0.0618 (7)
H21	0.3014	0.8956	0.6454	0.074*
C22	0.2078 (3)	0.6667 (3)	0.59722 (19)	0.0601 (6)
H22	0.1098	0.6332	0.6169	0.072*
C23	0.2358 (3)	0.5605 (3)	0.5510 (2)	0.0570 (6)
H23	0.1562	0.4535	0.5377	0.068*
C24	0.3794 (2)	0.6080 (2)	0.52361 (17)	0.0455 (5)
H24	0.3978	0.5335	0.4918	0.055*
C25	0.95697 (18)	0.91077 (19)	0.49401 (12)	0.0275 (4)
H25A	1.0305	0.8742	0.4799	0.033*
H25B	0.9299	0.8748	0.5595	0.033*
O1D	0.14507	0.33006	0.22949	0.0611 (4)

	(17)	(18)	(14)	
C1D	0.2159 (2)	0.2568 (3)	0.22229 (18)	0.0496 (5)
H1D	0.2730	0.2426	0.2847	0.060*
N1D	0.2202 (2)	0.1958 (2)	0.13559 (15)	0.0516 (5)
C2D	0.3156 (4)	0.1194 (3)	0.1349 (2)	0.0793 (8)
H2D1	0.3617	0.1117	0.2059	0.119*
H2D2	0.4025	0.1803	0.1016	0.119*
H2D3	0.2474	0.0145	0.0968	0.119*
C3D	0.1552 (3)	0.2305 (4)	0.0375 (2)	0.0821 (9)
H3D1	0.0988	0.2875	0.0482	0.123*
H3D2	0.0798	0.1329	-0.0079	0.123*
H3D3	0.2429	0.2947	0.0057	0.123*

Table 7. Atomic displacement parameters  $Å^2$  for 7a.

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
<b>S</b> 1	0.0190 (2)	0.0430 (3)	0.0300 (2)	0.01033 (19)	0.00122 (16)	0.00262 (18)
C2	0.0230 (8)	0.0271 (8)	0.0272 (8)	0.0118 (7)	0.0022 (6)	0.0038 (6)
N3	0.0182 (6)	0.0319 (7)	0.0219 (7)	0.0096 (6)	0.0006 (5)	0.0017 (5)
C4	0.0201 (7)	0.0251 (8)	0.0237 (8)	0.0103 (6)	0.0002 (6)	0.0029 (6)
C5	0.0310 (8)	0.0331 (9)	0.0275 (8)	0.0199 (7)	0.0066 (7)	0.0076 (7)
C6	0.0321 (9)	0.0472 (11)	0.0512 (12)	0.0242 (9)	0.0080 (8)	0.0053 (9)
C7	0.0450 (12)	0.0678 (15)	0.0812 (18)	0.0408 (12)	0.0159 (12)	0.0055 (13)
C8	0.0687 (15)	0.0613 (15)	0.0809 (18)	0.0498 (14)	0.0273 (13)	0.0080 (13)
C9	0.0618 (13)	0.0385 (11)	0.0534 (13)	0.0307 (11)	0.0161 (11)	0.0026 (9)
C10	0.0400 (10)	0.0297 (9)	0.0337 (9)	0.0201 (8)	0.0096 (8)	0.0061 (7)
C11	0.0353	0.0285	0.0311 (9)	0.0123 (8)	0.0062 (7)	0.0025

Г

	(9)	(9)				(7)
011	0.0417 (8)	0.0395 (8)	0.0584 (10)	0.0085 (7)	-0.0005 (7)	-0.0164 (7)
C12	0.0212 (7)	0.0292 (8)	0.0251 (8)	0.0109 (7)	0.0002 (6)	0.0003 (6)
C13	0.0248 (8)	0.0325 (9)	0.0240 (8)	0.0144 (7)	-0.0006 (6)	0.0026 (6)
C14	0.0257 (8)	0.0308 (8)	0.0232 (8)	0.0168 (7)	0.0011 (6)	0.0012 (6)
N14	0.0314 (8)	0.0329 (8)	0.0364 (9)	0.0147 (7)	-0.0029 (7)	0.0091 (7)
015	0.0224 (5)	0.0265 (6)	0.0262 (6)	0.0106 (5)	-0.0016 (4)	0.0045 (4)
C16	0.0310 (9)	0.0349 (9)	0.0306 (9)	0.0137 (8)	-0.0009 (7)	0.0035 (7)
N17	0.0396 (9)	0.0489 (10)	0.0458 (10)	0.0178 (8)	-0.0130 (8)	0.0069 (8)
N18	0.0291 (7)	0.0423 (9)	0.0285 (8)	0.0165 (7)	0.0073 (6)	0.0047 (6)
C19	0.0315 (9)	0.0501 (11)	0.0285 (9)	0.0230 (9)	0.0092 (7)	0.0120 (8)
C20	0.0482 (12)	0.0570 (13)	0.0451 (12)	0.0270 (11)	0.0178 (10)	0.0010 (10)
C21	0.0644 (15)	0.0829 (18)	0.0587 (15)	0.0461 (15)	0.0293 (12)	0.0060 (13)
C22	0.0475 (13)	0.0885 (19)	0.0621 (15)	0.0397 (14)	0.0311 (11)	0.0280 (13)
C23	0.0453 (12)	0.0561 (14)	0.0752 (17)	0.0237 (11)	0.0277 (12)	0.0289 (12)
C24	0.0425 (11)	0.0471 (12)	0.0579 (13)	0.0265 (10)	0.0216 (10)	0.0195 (10)
C25	0.0211 (7)	0.0334 (9)	0.0237 (8)	0.0126 (7)	-0.0012 (6)	0.0034 (6)
01D	0.0389 (8)	0.0463 (9)	0.0858 (13)	0.0160 (7)	0.0035 (8)	-0.0015 (8)
C1D	0.0381 (11)	0.0503 (12)	0.0526 (13)	0.0182 (10)	0.0041 (9)	0.0114 (10)
N1D	0.0502 (10)	0.0527 (11)	0.0491 (11)	0.0213 (9)	0.0157 (9)	0.0153 (9)
C2D	0.092 (2)	0.0783 (19)	0.089 (2)	0.0499 (17)	0.0427 (18)	0.0208 (16)
C3D	0.0672 (17)	0.106 (2)	0.0490 (15)	0.0279 (17)	0.0022 (13)	0.0184 (15)

S1—C2	1.7771 (16)	N14—H14B	0.871 (15)
S1—C12	1.8245 (17)	C16—N17	1.146 (2)
C2—N18	1.268 (2)	N18—C19	1.423 (2)
C2—N3	1.385 (2)	C19—C24	1.385 (3)
N3—C4	1.423 (2)	C19—C20	1.390 (3)
N3—C25	1.4652 (18)	C20—C21	1.381 (3)
C4—O15	1.4739 (18)	С20—Н20	0.9500
C4—C5	1.509 (2)	C21—C22	1.372 (4)
C4—C12	1.550 (2)	C21—H21	0.9500
C5—C6	1.388 (2)	C22—C23	1.370 (3)
C5—C10	1.390 (2)	С22—Н22	0.9500
C6—C7	1.378 (3)	C23—C24	1.385 (3)
С6—Н6	0.9500	С23—Н23	0.9500
С7—С8	1.387 (3)	C24—H24	0.9500
С7—Н7	0.9500	C25—C25 <sup>i</sup>	1.529 (3)
C8—C9	1.378 (3)	С25—Н25А	0.9900
C8—H8	0.9500	С25—Н25В	0.9900
C9—C10	1.390 (3)	O1D—C1D	1.225 (3)
С9—Н9	0.9500	C1D—N1D	1.322 (3)
C10—C11	1.474 (2)	C1D—H1D	0.9500
C11—O11	1.203 (2)	N1D—C2D	1.445 (3)
C11—C12	1.540 (2)	N1D—C3D	1.447 (3)
C12—C13	1.495 (2)	C2D—H2D1	0.9800
C13—C14	1.369 (2)	C2D—H2D2	0.9800
C13—C16	1.403 (2)	C2D—H2D3	0.9800
C14—N14	1.318 (2)	C3D—H3D1	0.9800
C14—O15	1.3551 (18)	C3D—H3D2	0.9800
N14—H14A	0.888 (15)	C3D—H3D3	0.9800
C2—S1—C12	92.11 (7)	C14—N14— H14B	118.1 (14)
N18—C2—N3	122.07	H14A—N14—	123 (2)

Geometric parameters (Å, °) for  ${\bf 7a}$ 

	(14)	H14B	
N18—C2—S1	126.61 (13)	C14—O15—C4	108.30 (11)
N3—C2—S1	111.30 (12)	N17—C16—C13	178.9 (2)
C2—N3—C4	117.03 (12)	C2—N18—C19	120.80 (15)
C2—N3—C25	120.23 (13)	C24—C19—C20	118.90 (18)
C4—N3—C25	122.28 (13)	C24—C19—N18	123.59 (17)
N3—C4—O15	110.09 (12)	C20—C19—N18	117.43 (17)
N3—C4—C5	116.58 (13)	C21—C20—C19	120.2 (2)
O15—C4—C5	108.85 (13)	C21—C20—H20	119.9
N3—C4—C12	109.35 (13)	C19—C20—H20	119.9
015—C4— C12	106.09 (12)	C22—C21—C20	120.5 (2)
C5—C4—C12	105.29 (13)	C22—C21—H21	119.7
C6—C5—C10	120.52 (17)	C20—C21—H21	119.7
C6—C5—C4	128.68 (16)	C23—C22—C21	119.7 (2)
C10—C5—C4	110.76 (14)	C23—C22—H22	120.2
C7—C6—C5	117.71 (19)	C21—C22—H22	120.2
С7—С6—Н6	121.1	C22—C23—C24	120.5 (2)
С5—С6—Н6	121.1	С22—С23—Н23	119.7
C6—C7—C8	122.0 (2)	C24—C23—H23	119.7
С6—С7—Н7	119.0	C19—C24—C23	120.2 (2)
С8—С7—Н7	119.0	C19—C24—H24	119.9
C9—C8—C7	120.4 (2)	C23—C24—H24	119.9
C9—C8—H8	119.8	N3—C25—C25 <sup>i</sup>	112.39 (16)
С7—С8—Н8	119.8	N3—C25— H25A	109.1
C8—C9—C10	118.1 (2)	C25 <sup>i</sup> —C25—	109.1

		H25A	
С8—С9—Н9	120.9	N3—C25— H25B	109.1
С10—С9—Н9	120.9	C25 <sup>i</sup> —C25— H25B	109.1
C5—C10—C9	121.17 (17)	H25A—C25— H25B	107.9
C5—C10—	110.85	O1D—C1D—	126.0 (2)
C11	(15)	N1D	
C9—C10—	127.92	O1D—C1D—	117.0
C11	(17)	H1D	
O11—C11—	128.27	N1D—C1D—	117.0
C10	(18)	H1D	
O11—C11—	124.49	C1D—N1D—	121.7 (2)
C12	(17)	C2D	
C10—C11—	107.17	C1D—N1D—	120.2 (2)
C12	(14)	C3D	
C13—C12—	113.09	C2D—N1D—	117.0 (2)
C11	(14)	C3D	
C13—C12—	102.01	N1D—C2D—	109.5
C4	(12)	H2D1	
C11—C12—	104.96	N1D—C2D—	109.5
C4	(13)	H2D2	
C13—C12—S1	115.58 (12)	H2D1—C2D— H2D2	109.5
C11—C12—S1	111.88 (12)	N1D—C2D— H2D3	109.5
C4—C12—S1	108.13 (11)	H2D1—C2D— H2D3	109.5
C14—C13—	125.02	H2D2—C2D—	109.5
C16	(16)	H2D3	
C14—C13—	109.73	N1D—C3D—	109.5
C12	(13)	H3D1	
C16—C13—	125.00	N1D—C3D—	109.5
C12	(15)	H3D2	
N14—C14—	115.07	H3D1—C3D—	109.5
O15	(14)	H3D2	
N14—C14—	131.58	N1D—C3D—	109.5
C13	(15)	H3D3	
O15—C14—	113.34	H3D1—C3D—	109.5
C13	(14)	H3D3	
C14—N14— H14A	118.6 (15)	H3D2—C3D— H3D3	109.5

C12—S1—	170.23	C5—C4—C12—	-108.46
C2—N18	(16)	C13	(14)
C12—S1—	-11.52	N3—C4—C12—	-116.30
C2—N3	(12)	C11	(14)
N18—C2—	-165.20	O15—C4—	125.00
N3—C4	(16)	C12—C11	(13)
S1—C2—N3— C4	16.45 (17)	C5—C4—C12— C11	9.69 (16)
N18—C2— N3—C25	7.3 (2)	N3—C4—C12— S1	3.26 (15)
S1—C2—N3—	-171.07	015—C4—	-115.44
C25	(11)	C12—S1	(12)
C2—N3—	103.49	C5—C4—C12—	129.26
C4—O15	(15)	S1	(12)
C25—N3—	-68.83	C2—S1—C12—	-109.09
C4—O15	(18)	C13	(12)
C2—N3—	-131.93	C2—S1—C12—	119.56
C4—C5	(15)	C11	(12)
C25—N3— C4—C5	55.8 (2)	C2—S1—C12— C4	4.45 (12)
C2—N3—	-12.70	C11—C12—	-116.71
C4—C12	(19)	C13—C14	(16)
C25—N3—	174.99	C4—C12—	-4.52
C4—C12	(13)	C13—C14	(18)
N3—C4—	-68.6 (2)	S1—C12—	112.51
C5—C6		C13—C14	(14)
015—C4— C5—C6	56.6 (2)	C11—C12— C13—C16	68.8 (2)
C12—C4—	169.97	C4—C12—	-178.97
C5—C6	(18)	C13—C16	(17)
N3—C4—	113.55	S1—C12—	-61.9 (2)
C5—C10	(16)	C13—C16	
O15—C4—	-121.23	C16—C13—	-3.7 (3)
C5—C10	(14)	C14—N14	
C12—C4—	-7.84 (18)	C12—C13—	-178.10
C5—C10		C14—N14	(18)
C10—C5—	-2.2 (3)	C16—C13—	174.72
C6—C7		C14—O15	(16)
C4—C5—	-179.79	C12—C13—	0.3 (2)
C6—C7	(19)	C14—O15	
C5—C6—	0.7 (4)	N14—C14—	-176.88
C7—C8		O15—C4	(15)
C6—C7—	0.9 (4)	C13—C14—	4.47 (18)

C8—C9		O15—C4	
C7—C8—	-1.1 (4)	N3—C4—O15—	-125.32
C9—C10		C14	(13)
C6—C5—	2.0 (3)	C5—C4—O15—	105.76
C10—C9		C14	(14)
C4—C5—	-179.94	C12—C4—	-7.11
C10—C9	(16)	O15—C14	(16)
C6—C5—	-175.48	N3—C2—N18—	-177.20
C10—C11	(16)	C19	(15)
C4—C5— C10—C11	2.5 (2)	S1—C2—N18— C19	0.9 (2)
C8—C9— C10—C5	-0.4 (3)	C2—N18— C19—C24	54.9 (2)
C8—C9—	176.7 (2)	C2—N18—	-128.4
C10—C11		C19—C20	(2)
C5—C10—	-178.95	C24—C19—	-0.6 (3)
C11—O11	(19)	C20—C21	
C9—C10—	3.7 (3)	N18—C19—	-177.45
C11—O11		C20—C21	(19)
C5—C10— C11—C12	3.9 (2)	C19—C20— C21—C22	1.4 (4)
C9—C10—	-173.37	C20—C21—	-1.6 (4)
C11—C12	(18)	C22—C23	
011—C11— C12—C13	-75.3 (2)	C21—C22— C23—C24	1.0 (4)
C10—C11—	101.90	C20—C19—	0.0 (3)
C12—C13	(16)	C24—C23	
011—C11—	174.29	N18—C19—	176.68
C12—C4	(18)	C24—C23	(19)
C10—C11— C12—C4	-8.47 (18)	C22—C23— C24—C19	-0.2 (3)
011—C11— C12—S1	57.3 (2)	C2—N3—C25— C25 <sup>i</sup>	-97.0 (2)
C10—C11—	-125.50	C4—N3—C25—	75.1 (2)
C12—S1	(13)	C25 <sup>i</sup>	
N3—C4—	125.55	01D—C1D—	176.1 (2)
C12—C13	(14)	N1D—C2D	
015—C4— C12—C13	6.84 (16)	O1D—C1D— N1D—C3D	8.5 (3)

Symmetry code: (i) -*x*+2, -*y*+2, -*z*+1.

Hydrogen-bond geometry (Å, °) for 7a.

D—H···A	<i>D</i> —Н	Н…А	$D \cdots A$	<i>D</i> — H…A
N14— H14 <i>A</i> …O1 <i>D</i> <sup>ii</sup>	0.89 (2)	1.94 (2)	2.806 (2)	166 (2)
N14— H14 <i>B</i> …N17 <sup>iii</sup>	0.87 (2)	2.12 (2)	2.985 (2)	170 (2)
C25— H25 <i>B</i> ····O15 <sup>i</sup>	0.99	2.44	3.0677 (19)	121
C3 <i>D</i> — H3 <i>D</i> 3…O11	0.98	2.58	3.141 (3)	116

Symmetry codes: (i) -*x*+2, -*y*+2, -*z*+1; (ii) *x*+1, *y*+1, *z*; (iii) -*x*+1, -*y*+2, -*z*.

#### <sup>1</sup>H-NMR of (7a):-



# <sup>13</sup>C-NMR of (7a):-



# <sup>1</sup>H-NMR of (7b):-



# <sup>13</sup>C-NMR of (7b):-



# 1 <sup>1</sup>H-NMR of (7c):-



# 2

## 3 $\frac{^{13}\text{C-NMR of (7c):-}}{^{13}\text{C-NMR of (7c):-}}$



# 5 <u><sup>1</sup>H-NMR of (7d):-</u>





# 7 <sup>13</sup>C-NMR of (7d):-



8 9

## 10 11 **<u><sup>1</sup>H-NMR of (7e):-</u>**



# 13 <sup>13</sup>C-NMR of (7e):-

12



14

15