Regioselective combination of 1,3-dinucleophiles such as 2-mercaptobenzimidazole, 5-mercapto-3-phenyltriazole and potassium thiocyanate with 2,2-dicyanooxiranes

Mohammad Seifi and Hassan Sheibani*

Department of Chemistry, Shahid Bahonar University of Kerman, Kerman 76169, Iran E-mail: hsheibani@mail.uk.ac.ir

DOI: http://dx.doi.org/10.3998/ark.5550190.p008.002

Abstract

A group of heterocyclic compounds of condensed [1,3]thiazole with benzimidazole or triazole derivatives were obtained by the regioselective reactions of the 2,2-dicyanooxiranes with 1*H*-1,3-benzimidazol-2-thiol or 5-phenyl-4*H*-1,2,4-triazol-3-thiol respectively in good to excellent yields. Also, we performed a reaction between 2,2-oxiranedicarbonitrile reagents and potassium thiocyanate in acetic anhydride at room temperature and isolated 2-acetylimino-1,3-oxathiole derivatives.

Keywords: 2,2-Dicyanooxiranes, 1*H*-1,3-benzimidazol-2-thiols, 5-phenyl-4*H*-1,2,4-triazol-3-thiols, potassium thiocyanate

Introduction

The design, synthesis and property of novel thiazole derivatives as a class of the most important heterocyclic compounds have been paid extensive and continuous attention in the fields of industry, agriculture and pharmacy. Many synthetic small molecules from different groups of heterocycles with influence on carcinogenesis have been reported and several of them are currently in clinical trials.

Thiazolone nuclei fused to heterocyclic systems represent a very important class of compounds used mainly as intermediates for the preparation of bioactive compounds. For example, 2-arylidene-[1,3]thiazolo[3,2-a]benzimidazol-3(2H)-ones were found antibacterial agents,⁴ inhibitors of ubiquitin ligase⁵ and other enzymes, and they may be used for the treatment of viral and inflammatory diseases, neurological disorders and cancer. In addition, some isosteric thiazolo[3,4-a]benzimidazoles inhibit enteroviruses⁶ and tumors.⁷ The synthesis of such compounds has been the subject of several reviews which demonstrate the high importance of this class of compounds.⁸⁻¹³ Considering the importance of these compounds, many methods for

the synthesis of these derivatives have been reported successively. The conventional synthesis involves the nucleophilic addition of 1,3-dinucleophies with 2,2-dicyanooxirane, alkyl α -haloacetate or α -haloketone derivatives.¹⁴ 2,2-Dicyanooxiranes and their α -haloketone derivatives are known as very effective bielectrophile reagents in the synthesis of a large variety of heterocyclic compounds of pharmaceutical interest such as thiazoles, dithioles, imidazoles, and condensed imidazolo and thiazolo derivatives, because of their multifunctional structure. In general, nucleophilic reagents can react with these compounds regionelectively: they can attack the oxirane ring, cyano functional group to give either ring opened products, new functionalised oxiranes or different heterocycles. In our continuing interest in the synthesis of heterocycles containing nitrogen and sulfur atoms, $^{20-22}$ we decided to investigate the condensation reaction of 1,3-dinucleophies with 2,2-dicyanooxiranes, which is extended to the formation of 1,3-thiazolone derivatives.

So in this paper, we describe an investigation of the condensation of potassium thiocyanate with 2,2-dicyanooxiranes which released to 1,3-oxathiole derivatives. These compounds are establish in a variety of useful heterocyclic derivatives applied as biologically active compounds and pharmaceutical products.^{23, 24}

Results and Discussion

To study the regioselectivity of 1,3-dinucleophiles such as 1H-1,3-benzimidazol-2-thiol and 5-phenyl-4H-1,2,4-triazol-3-thiol with 2,2-dicyanooxiranes, we have investigated the nucleophilic reaction of 1H-1,3-benzimidazol-2-thiol **2** with 2,2-dicyanooxiranes **1a-d** to prepare of 2-arylidene [1,3]thiazolo[3,2,a]benzimidazol-3(2H)-one (**3a-d**) derivatives in mild reaction conditions in good to excellent yields (Scheme 1).

1a-d 2 1a, 3a:
$$R^1=R^2=R^3=H$$
1b, 3b: $R^1=R^3=H$, $R^2=CI$
1c, 3c: $R^1=R^3=OCH_3$, $R^2=H$

Scheme 1

Encouraged by these results, we have extended these reactions by nucleophilic addition of 5-substituted 5-phenyl-4*H*-1,2,4-triazol-3-thiol **4** on 2,2-dicyanooxiranes **1b** and **1d** to synthesis of 5-substituted thiazolo[3,2-*b*][1,2,4]triazol-6-ones **5a-b** in good yields (Scheme 2).

Scheme 2

In these protocols (Scheme 1 and 2) we have investigated reactions of 2,2-dicyanooxiranes 1, which have three electron-deficient centers, with 1,3-dinucleophile such as 1H-1,3benzimidazol-2-thiol 2 and 5-phenyl-4*H*-1,2,4-triazol-3-thiol 4 which arylidene[1,3]thiazolo[3,2,a]benzimidazol-3(2H)-one **3a-d** and 5-substituted thiazolo[3,2b][1,2,4]triazol-6-one **5a-b** derivatives, respectively. In these reactions 1,3-dinucleophiles act via regioselective on C_{β} and C_{α} of 2,2-dicyanooxiranes 1 which have three electron deficient centers. These processes are highly regioselective, and compounds 3 and 5 results from the initial attack of the nucleophilic sulfur atom of compounds 2 and 4 on C_β of 2,2-dicyanooxiranes 1 followed by ring-opening to form cyanhydrin intermediate and then cyanoformyl intermediate I, II was formed by losing of cyano group. The last step is along with intramolecular nucleophilic addition of nitrogen on carbonyl group of cyanoformyl intermediate I, II and followed by the elimination of HCN to form only one product (Scheme 3).

Scheme 3

Encouraged by these results, we performed the reaction of 2,2-dicyanooxiranes (1a, 1b and 1d) with potassium thiocyanate 6 in acetic anhydride 7 at room temperature and isolated 2-acetylimino-1,3-oxathiole derivatives 8a-c. The reaction presumably proceeds through thiocyanate attack on C_{β} of 2,2-dicyanooxiranes 1 followed by ring-opening to form cyanhydrin intermediate A. The intramolecular nucleophilic addition of oxygen on carbon of nitrile intermediate A leads to the formation of intermediate B which is trapped by acetic anhydride to yield intermediate C. The last step is along with elimination of HCN from intermediate (C) and then formation of 1,3-oxathioles 8a-c (Scheme 4).

Scheme 4

The structure of compounds **3a-d**, **5a, 5b** and **8a-c** were determined on the basis of their elemental analyses, mass spectrum, 1 H and 13 C NMR and IR spectroscopic data. Only one product was obtained in each case. The 1 H NMR spectrum of **3a** indicated two kinds of protons along with one signal at 6.40 ppm attributed to one proton of C_2 . The aryl protons appeared at δ 8.82-7.39 ppm (9H, m). The 1 H decoupled 13 C NMR spectrum of **3a** was measured in DMSO- d_6 and it is in accord with the proposed structure. The 13 C NMR of compound **3a** showed two signals at δ 153.84 and 144.75 ppm which are due to the carbonyl and imine carbons respectively. The ten aromatic carbons appeared at δ 142.92-117.76 ppm. These signals along with an upfield shift of aliphatic carbon at δ 50.10 ppm which was identified as aliphatic carbon (C_2). Also the structure of compound **8a** was established on the basis of spectral deta, which IR

spectrum of this compound showed the presence of CN at region 2225cm^{-1} and sharp bands at 1662 and 1601 cm⁻¹, which are due to carbonyl of amide (C=O) and imine (C=N) groups respectively. The ¹H NMR spectrum of **8a** indicated two kinds of proton signals with one at δ 2.66 (s, 3H, CH₃) ppm was identified as acetyl group. The five protons of phenyl group appeared at δ 7.77-7.64 ppm. The ¹³C NMR of compound **8a** showed two signals at δ 181.95 and 174.06 ppm which are due to the carbonyl of amide (C=O) and imine (C=N) groups respectively. The oxegen and cyano groups adjacent to double bond strongly polarize the double bond system. The α -carbon (C₅) and β -carbon (C₄) of these substituents appear at δ 138.41 and 114.67 ppm respectively. The aromatic carbons along with nitryl carbon appeared at δ 131.79, 129.82, 127.64, 126.34 and 110.39 (CN) ppm respectively. These signals along with the upfield shift of aliphatic carbon at δ 27.08 ppm which was identified as acetyl group.

Conclusions

We have developed a simple and efficient method for the synthesis of heterocyclic compounds which condensed [1,3]thiazole with benzimidazole or triazole derivatives via reaction of 2,2-dicyanooxiranes with 1,3-dinucleophiles. We have investigated reaction of potassium thiocyanate with 2,2-dicyanooxiranes in acetic anhydride to synthesis of oxathiole derivatives. The advantage of these procedures reported here are: high selectivity, high purity of products and easy workup.

Experimental Section

General. Melting points were measured on a Electrothermal Enginering LTD apparatus and are uncorrected. IR spectra were measured on a Mattson 1000 FT-IR spectrometer. The proton and carbon NMR spectra were recorded with a BRUKER DRX-400 AVANCE spectrometer at 400 and 100 MHz, respectively. Mass spectra were recorded on a MS-QP2000A Shimadzu mass spectrometer operating at an ionization potential of 70 eV. Elemental analyses were performed using a Heracus CHN-O-Rapid analyzer. 2,2-dicyanooxiranes **1a-d** was prepared according to a literature procedure.²⁵

Procedure for the preparation of 2-arylidene [1,3]thiazolo[3,2-a]benzimidazol-3(2H)-one (3a-d) and 5-substituted thiazolo[3,2-b][1,2,4]triazol-6-ones (5a-b). A mixture of 2,2-dicyanooxiranes **1a-d** (2 mmol) and 1*H*-benzo[*d*]imidazole-2-thiol **2** or 5-mercapto-3-phenyl-striazole **4** (2 mmol) in CH₃CN (20 mL) was stirred for 7 hour (the progress of the reaction being monitored by TLC and was used hexane/ethyl acetate as an eluent). When the reaction was completed as indicated by TLC, the crude product **3a-3d** and **5a, 5b** was precipitated from the reaction mixture, and the solid was filtered and recrystallized with hexane/ethyl acetate to get pure product.

- **2-Phenyl[1,3]thiazolo[3,2-a]benzimidazol-3(2***H***)-one (3a). White crystals; yield: 90%. mp 190-192 °C. IR (KBr, v_{max}/cm^{-1}): 1742 (C=O), 1603 (C=N). ¹H NMR (400 MHz, DMSO-d_6): 8.81 (d, 1H, {}^3J_{\text{H-H}} = 8 Hz, CH), 7.71 (d, 1H, {}^3J_{\text{H-H}} = 8 Hz, CH), 7.67-7.39 (m, 7H, Ar), 6.40 (s, 1H). {}^{13}C NMR (100 MHz, DMSO-d_6): 153.84 (C=O), 144.75 (C=N), 142.92, 133.21, 132.69, 132.38, 129.45, 129.04, 128.93, 124.63, 122.87, 117.76, 50.10 (C₂). MS (m/z): 266 (M⁺) (98), 237 (80), 205 (20), 150 (15), 121 (100), 90 (78), 77 (40). Anal. calcd. for C₁₅H₁₀N₂OS: C, 67.65; H, 3.78; N, 10.52%. Found: C, 67.42; H, 3.66; N, 10.18%.**
- **2-(4-Cholorophenyl)[1,3]thiazolo[3,2-a]benzimidazol-3(2***H***)-one (3b).** White crystals; yield: 93%. mp 136-139 °C. IR (KBr, $v_{\text{max}}/\text{cm}^{-1}$): 1726 (C=O), 1609 (C=N). ¹H NMR (400 MHz, DMSO- d_6): 7.87 (d, 1H, ${}^3J_{\text{H-H}} = 8$ Hz, CH), 7.66 (d, 1H, ${}^3J_{\text{H-H}} = 8$ Hz, CH), 7.62 (d, 2H, ${}^3J_{\text{H-H}} = 8$ Hz, CH), 7.50 (d, 2H, ${}^3J_{\text{H-H}} = 8$ Hz, CH), 7.42 (t, 1H, ${}^3J_{\text{H-H}} = 8$ Hz, CH), 7.35 (d, 1H, ${}^3J_{\text{H-H}} = 8$ Hz, CH), 6.32 (s, 1H). 13 C NMR (100 MHz, DMSO- d_6): 166.83 (C=O), 156.88 (C=N), 149.42, 133.61, 133.44, 131.17, 128.72, 128.02, 125.77, 123.74, 123.35, 118.70, 67.32 (C₂). MS (m/z): 300 (M⁺) (100), 271 (70), 237 (50), 156 (80), 124 (20), 89 (73), 63 (40). Anal. calcd. for C₁₅H₉ClN₂OS: C, 59.90; H, 3.02; N, 9.31%. Found: C, 59.83; H, 2.88; N, 9.01%.
- **2-(4-Nitrophenyl)[1,3]thiazolo[3,2-***a*]benzimidazol-3(2*H*)-one (3c). White crystals; yield: 94%. mp 193-196 °C. IR (KBr, $v_{\text{max}}/\text{cm}^{-1}$): 1750 (C=O), 1609 (C=N). ¹H NMR (400 MHz, DMSO-*d*₆): 8.49 (d, 2H, ${}^{3}J_{\text{H-H}} = 8$ Hz, CH), 7.72 (d, 1H, ${}^{3}J_{\text{H-H}} = 8$ Hz, CH), 7.50-7.26 (m, 5H, Ar), 5.79 (s, 1H). ${}^{13}\text{C}$ NMR (100 MHz, DMSO-*d*₆): 165.83 (C=O), 153.20 (C=N), 150.3, 145.4, 140.01, 138.5, 129.73, 126.56, 124.55, 124.49, 119.51, 128.9, 57.61 (C₂). MS (*m*/*z*): 311 (M⁺) (100), 282 (20), 265 (15), 236 (40), 167 (98), 151 (20), 145 (50), 121 (30), 109 (20), 90 (77), 77 (80), 63 (88). Anal. calcd. for C₁₅H₉N₃O₃S: C, 57.87; H, 2.91; N, 13.50%. Found: C, 57.69; H, 2.85; N, 13.21%.
- **2-(2,5-Dimethoxyphenyl)**[1,3]thiazolo[3,2-a]benzimidazol-3(2H)-one (3d). Yellow crystals; yield: 90%. mp 140-143 °C. IR (KBr, v_{max}/cm⁻¹): 1748 (C=O), 1611 (C=N). ¹H NMR (400 MHz, DMSO- d_6): 7.81 (d, 1H, ${}^3J_{\text{H-H}} = 8$ Hz, CH), 7.49 (d, 1H, ${}^3J_{\text{H-H}} = 8$ Hz, CH), 7.31-7.12 (m, 2H, CH), 7.05 (s, 1H, CH), 7.02 (d, 1H, ${}^3J_{\text{H-H}} = 8$ Hz, CH), 6.88 (d, 1H, ${}^3J_{\text{H-H}} = 8$ Hz, CH), 6.01 (s, 1H), 3.65, 3.48 (s, 6H, 2OCH₃). 13 C NMR (100 MHz, DMSO- d_6): 167.91 (C=O), 154.70 (C=N), 152.59, 150.56, 126.40, 124.75, 124.38, 123.38, 119.61, 117.79, 116.30, 56.88, 56.09 (2OCH₃) 57.61 (C₂). MS (m/z): 326 (M⁺) (100), 311 (85), 267 (20), 178 (15), 163 (50), 149 (60), 135 (20), 121 (40), 107 (12), 91 (30), 77 (43), 63 (35). Anal. calcd. for C₁₇H₁₄N₂O₃S: C, 62.56; H, 4.32; N, 8.58%. Found: C, 62.44; H, 4.28; N, 8.31%.
- **5-(4-Chlorophenyl)-2-phenylthiazolo[3,2-***b***][1,2,4]triazol-6(5***H***)-one (5a). White crystals; yield: 90%. mp 152-155 °C. IR (KBr, v_{max}/cm^{-1}): 1744 (C=O), 1692, 1611 (C=N). ¹H NMR (400 MHz, DMSO-***d***₆): 8.13-8.11 (m, 2H, Ar), 7.64 (d, 2H, {}^{3}J_{H-H} = 8 Hz, CH), 7.57-7.53 (m, 3H, Ar), 7.50 (d, 2H, {}^{3}J_{H-H} = 8 Hz, CH), 6.41 (s, 1H). {}^{13}C NMR (100 MHz, DMSO-***d***₆): 169.73 (C=O), 165.03, 163.96 (C=N), 134.00, 132.56, 131.35, 131.27, 129.15, 129.04, 128.97, 126.90, 58.17 (C₅). MS (m/z): 327 (M⁺) (80), 301 (12), 281 (12), 264 (55), 177 (20), 152 (60), 125 (40), 103 (100), 89 (85), 77 (30). Anal. calcd. for C₁₆H₁₀ClN₃OS: C, 58.63; H, 3.07; N, 12.28%. Found: C, 58.44; H, 2.98; N, 11.97%.**

5-(2,5-Dimethoxyphenyl)-2-phenylthiazolo[3,2-*b***][1,2,4]triazol-6(5***H***)-one (5***b***). Orange crystals; yield: 94%. mp: 204-206 °C. IR (KBr, v_{\text{max}}/\text{cm}^{-1}): 1754 (C=O), 1606 (C=N). ¹H NMR (400 MHz, DMSO-***d***₆): 8.13 (d, 2H, ³***J***_{H-H}=8Hz, CH), 7.70-7.66 (m, 3H, Ar), 7.20 (s, 1H, CH), 7.05-7.00 (m, 2H, CH), 6.31 (s, 1H), 3.76, 3.67 (s, 6H, 2OCH₃). ¹³C NMR (100 MHz, DMSO-***d***₆): 169.09 (C=O), 164.73, 164.32 (C=N), 153.08, 151.17, 131.19, 129.10, 129.05, 126.88, 122.58, 117.28, 115.78, 56.45, 55.84 (2OCH₃), 55.67 (C₅). MS (***m***/***z***): 353 (M⁺) (12), 294 (7), 178 (30), 163 (95), 147 (20), 135 (23), 121 (15), 103 (100), 92 (18), 77 (28). Anal. calcd. for C_{18}H_{15}N_3O_3S: C, 61.18; H, 4.28; N, 11.89%. Found: C, 60.95; H, 4.15; N, 11.60%.**

Procedure for preparation of 2-acetylimino-1,3-oxathiole derivatives 8a-c. A mixture of 2,2-dicyanooxiranes **1a-d** (2 mmol) and KSCN (2 mmol) was dissolved in 10 ml of acetic anhydride and stirred at room temperature for 3 h. The product precipitated from the reaction mixture and was collected by filtration. It was recrystallized from EtOH.

(5*E*)-*N*-(5-Cyano-4-phenyl)-1,3-oxathiol-2-ylidene)acetamide (8a). Brown crystals; yield: 90%. mp 97-99 °C. IR (KBr, v_{max}/cm^{-1}): 2225 (CN), 1662 (C=O), 1601 (C=N). ¹H NMR (400 MHz, DMSO- d_6): 7.77-7.74 (m, 3H, Ar), 7.64 (d, 2H, $^3J_{H-H}$ = 8 Hz, CH), 2.66 (s, 3H, CH₃). ¹³C NMR (100 MHz, DMSO- d_6): 181.95 (C=O), 174.06 (C=N), 138.41 (C₅), 131.79, 129.82, 127.64, 126.34, 114.67 (C₄), 110.39 (CN), 27.08 (CH₃). MS (m/z): 244 (M⁺) (7), 193 (8), 168 (15), 135(18), 121 (100), 107 (10), 89 (45), 77 (85). Anal. calcd. for C₁₂H₈N₂O₂S: C, 59.00; H, 3.30; N, 11.47%. Found: C, 58.79; H, 3.21; N, 11.12%.

(5*E*)-*N*-(4-(4-Chlorophenyl)-5-cyano-1,3-oxathiol-2-ylidene)acetamide (8b). Yellow crystals; yield: 95%. mp117-120 °C. IR (KBr, $v_{\text{max}}/\text{cm}^{-1}$): 2229 (CN), 1666 (C=O), 1599 (C=N). ¹H NMR (400 MHz, DMSO- d_6): 7.77 (d, 2H, $^3J_{\text{H-H}}$ = 8 Hz, CH), 7.72 (d, 2H, $^3J_{\text{H-H}}$ = 8 Hz, CH), 2.36 (s, 3H, CH₃). ¹³C NMR (100 MHz, DMSO- d_6): 182.66 (C=O), 163.37 (C=N), 149.82 (C₅), 136.97, 136.46, 129.90, 128.63, 124.29 (C₄), 110.22 (CN), 26.70 (CH₃). MS (m/z): 278 (M⁺) (90), 237 (8), 209 (8), 180(10), 155 (100), 114 (80), 93 (40), 70 (65). Anal. calcd. for C₁₂H₇ClN₂O₂S: C, 51.71; H, 2.53; N, 10.05%. Found: C, 51.48; H, 2.27; N, 9.71%.

(5*E*)-*N*-(5-Cyano-4-(2,5-dimethoxyphenyl)-1,3-oxathiol-2-ylidene)acetamide (8c). Yellow crystals; yield: 92%. mp 186-189 °C. IR (KBr, v_{max}/cm^{-1}): 2229 (CN), 1656 (C=O), 1609 (C=N).

¹H NMR (400 MHz, DMSO- d_6): 7.26-7.20 (m, 3H, Ar), 3.86, 3.77 (s, 6H, 2OCH₃), 2 (s, 3H, CH₃).

¹³C NMR (100 MHz, DMSO- d_6): 181.95 (C=O), 174.36 (C=N), 153.14 (C₅), 152.77, 150.62, 118.68, 115.42, 114.52, 113.85, 113.40 (C₄), 110.323 (CN), 56.34, 55.70 (2OCH₃), 26.89 (CH₃). MS (m/z): 304 (M⁺) (10), 262 (30), 235 (25), 181 (100), 149 (90), 133 (40), 121 (70), 107 (18), 95 (20), 77 (8). Anal. calcd. for C₁₄H₁₂N₂O₄S: C, 55.52; H, 3.97; N, 9.21%. Found: C, 55.39; H, 3.89; N, 8.94%.

Acknowledgements

The authors express appreciation to the Shahid Bahonar University of Kerman Faculty Research Committee for its support of this investigation.

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