# Organocatalytic asymmetric conjugate addition of cyclic 

## 1,3-dicarbonyl compounds to $\beta, \gamma$-unsaturated $\alpha$-ketoesters

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#### Abstract

The conjugate addition of cyclic 1,3-dicarbonyl compounds to $\beta, \gamma$-unsaturated $\alpha$-keto-esters was studied using a series of chiral bifunctional organocatalysts. Takemoto's catalyst was found to be most efficient for this transformation. Excellent yields and good enantioselectivities were achieved for a variety of $\beta, \gamma$-unsaturated $\alpha$-keto-esters and cyclic 1,3-dicarbonyl compounds. A bifunctional catalytic mechanism is proposed. The method provides a new asymmetric synthetic route for chiral courmarin derivatives.


Keywords: Asymmetric conjugate addition, organocatalyst, cyclic 1,3-dicarbonyl compound, $\beta, \gamma$-unsaturated $\alpha$-keto-ester

## Introduction

Asymmetric conjugate addition of 1,3-dicarbonyl compounds to various Michael acceptors is an important method for the preparation of chiral compounds. ${ }^{1}$ Although chiral metal catalysts have been applied successfully for a number of transformations, great efforts are continuing to develop more efficient, cheaper catalysts and to expand the substrate scope. In recent years asymmetric organocatalysis has emerged as a powerful tool for the synthesis of chiral compounds. ${ }^{2}$ Organocatalytic asymmetric conjugate additions of nucleophiles to Michael acceptors have been studied extensively. ${ }^{3}$ Excellent enantioselectivities have been obtained for the organocatalytic conjugate addition of 1,3 -dicarbonyl compounds to $\alpha, \beta$-unsaturated aldehydes, ketones and nitro compounds. Although $\beta, \gamma$-unsaturated $\alpha$-keto-esters are highly reactive Michael acceptors, their reactions with 1,3-dicarbonyl compounds have been seldom
studied. To the best of our knowledge, only a few papers concerning this reaction have appeared. Jørgensen et al. found that chiral bisoxazoline-copper catalysts are efficient for the asymmetric conjugate addition of cyclic 1,3-dicarbonyl compounds to $\beta, \gamma$-unsaturated $\alpha$-keto-esters. ${ }^{4}$ Calter and Wang reported that pyrimidinyl-cinchona alkaloid derivatives are suitable organocatalysts for the reaction, however the substrates were limited to cyclohexane-1,3-dione and 5,5-dimethylcyclohexane-1,3-dione. ${ }^{5}$ Feng, Liu and co-workers found that chiral $N, N^{\prime}$-dioxide-yttrium(III) complexes catalyze the conjugate addition of malonates to $\beta, \gamma$-unsaturated $\alpha$-keto-esters with excellent yields and enantioselectivities. ${ }^{6}$ Very recently Xu and co-workers reported that chiral squaramides are highly enantioselective catalysts for the conjugate addition of 4-hydroxycoumarins and 4-hydroxypyrone to $\beta, \gamma$-unsaturated $\alpha$-keto-esters. ${ }^{7}$ In recent years, chiral bifunctional organocatalysts combining hydrogen-bond donors (such as thioureas, ureas, sulfonamides) and amine groups have been proved to be extremely efficient catalysts for many reactions. ${ }^{8}$ We initiated a series of studies on the design and application of chiral bifunctional organocatalysts for asymmetric conjugate additions. ${ }^{9}$ Herein we report the asymmetric conjugate addition of cyclic 1,3-dicarbonyl compounds to $\beta, \gamma$-unsaturated $\alpha$-keto-esters. Chiral bifunctional thioureas were found to be efficient catalysts for the reaction.

## Results and Discussion

The conjugate addition of 4-hydroxycoumarin 1a to $(E)$-methyl 2-oxo-4-phenylbut-3-enoate 2a was examined with chiral bifunctional organocatalysts 4-9 (Scheme 1). The experimental results are summarized in Table 1. Quinine-derived thiourea 4 provided the expected product 3a in excellent yield and with moderate enantioselectivity (Table 1, entry 1). The product 3a was found to be in a rapid equilibrium with cyclic hemiketal 3a'. The equilibrium is sufficiently rapid that only two peaks from enantiomers were observed during chiral HPLC analysis. The same phenomenon was observed by Jørgensen and his co-workers. ${ }^{4}$ Cinchonine-derived thiourea 5 also gave 3a in excellent yield, but with lower enantioselectivity (Table 1, entry 2). Quinine-derived sulfonamide 6 also catalyzed the reaction with low enantioselectivity (Table 1, entry 3). Further study showed that Takemoto's catalyst 7 is a more efficient catalyst and $\mathbf{3 a}$ could be obtained with better enantioselectivity and excellent yield (Table 1, entry 4). ${ }^{10}$ Good yields were also achieved with chiral thiourea-primary amines $\mathbf{8}$ and 9 , however the enantioselectivities were low (Table 1, entries 5, 6). A screening of reaction solvents with catalyst 7 indicated that THF is optimal in terms of best enantioselectivity (Table 1, entries 4, 7-9). Decrease of reaction temperature resulted in the loss of enantioselectivity (Table 1, entries 10, 11). Using $5 \mathrm{~mol} \%$ catalyst 7 provided the product $\mathbf{3 a}$ in lower yield, but the enantioselectivity was almost unchanged (Table 1, entry 12). On the other hand, higher catalyst loading (20 mol\% 7) also did not provide substantial improvement (Table 1, entry 13).


Scheme 1

Table 1. Catalyst screening and optimization of reaction conditions ${ }^{\text {a }}$

| Entry | Catalyst | Solvent | Time (h) | Yield(\%) $^{\mathrm{b}}$ | $\mathrm{Ee}(\%)^{\mathrm{c}}$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 1 | $\mathbf{4}$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 | 99 | $59(\mathrm{R})$ |
| 2 | $\mathbf{5}$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 | 99 | $-32(\mathrm{~S})$ |
| 3 | $\mathbf{6}$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 | 99 | $30(\mathrm{R})$ |
| 4 | $\mathbf{7}$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 | 99 | $63(\mathrm{R})$ |
| 5 | $\mathbf{8}$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 3 | 89 | $-23(\mathrm{~S})$ |
| 6 | $\mathbf{9}$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 3 | 97 | $20(\mathrm{R})$ |
| 7 | $\mathbf{7}$ | $\mathrm{THF}^{2}$ | 3 | 99 | $76(\mathrm{R})$ |
| 8 | $\mathbf{7}$ | CHCl | 1 | 99 | $64(\mathrm{R})$ |
| 9 | $\mathbf{7}$ | toluene | 1 | 99 | $50(\mathrm{R})$ |
| $10^{[\mathrm{dd]}}$ | $\mathbf{7}$ | THF | 12 | 97 | $74(\mathrm{R})$ |
| $11^{[\mathrm{e}]}$ | $\mathbf{7}$ | THF | 48 | 94 | $53(\mathrm{R})$ |
| $12^{[\mathrm{ff}]}$ | $\mathbf{7}$ | THF | 6 | 71 | $75(\mathrm{R})$ |
| $13^{[\mathrm{g}]}$ | $\mathbf{7}$ | THF | 2.5 | 99 | $77(\mathrm{R})$ |

${ }^{\mathrm{a}}$ Unless otherwise noted, reactions were carried out with 1a ( 0.1 mmol ), 2a ( 0.1 mmol ) and catalyst ( 0.01 mmol ) in solvent $(1.0 \mathrm{~mL})$ at room temperature;
${ }^{\mathrm{b}}$ Isolated yields;
${ }^{\text {c }}$ Determined by chiral HPLC on a chiralpak AD-H column. The absolute configurations of 3a were assigned by comparing the optical rotations with the reported data (ref. 8);
${ }^{\mathrm{d}}$ The reaction was conducted at $0^{\circ} \mathrm{C}$;
${ }^{e}$ The reaction was conducted at $-30^{\circ} \mathrm{C}$;
${ }^{\mathrm{f}} 5 \mathrm{~mol} \% 7$ was used.
$\mathrm{g}_{20} \mathrm{~mol} \% 7$ was used.
The scope of $\beta, \gamma$-unsaturated $\alpha$-keto-esters was examined and the results are summarized in Table 2. The $\beta, \gamma$-unsaturated $\alpha$-keto-esters $\mathbf{2 a - 2 k}$ bearing various $\gamma$-aryl and heteroaryl substitutions afforded the desired adduct products $\mathbf{3 a - 3 k}$ in excellent yields and with good enantioselectivities ( $74-79 \%$ ee). The electronic property of the substituents on the phenyl ring exerted negligible effects on either the yields or enantioselectivities (Table 2, entries 1-8). Bulky ester groups of $\beta, \gamma$-unsaturated $\alpha$-keto-esters were also well tolerated (Table 2, entries 9-11). Furthermore $\gamma$-alkyl unsaturated keto-ester $\mathbf{2 l}$ is also compatible with this reaction. The product 31 was obtained in good yield and enantioselectivity after a prolonged reaction time (Table 2, entry 12). It should be noted that many optically active coumarin derivatives, such as 3a-31, possess various interesting biological activities. ${ }^{11}$

Table 2. Asymmetric conjugate addition of $\mathbf{1 a}$ to $\mathbf{2 a - 2 k}$ catalyzed by $\mathbf{7}^{\text {a }}$


| Entry | 2 | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | Time (h) | Product | Yield (\%) ${ }^{\text {b }}$ | $\mathrm{Ee}(\%)^{\text {c }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 2a | Ph | Me | 3 | 3a/3a' | 99 | 76 |
| 2 | 2b | 4-MeC66 $\mathrm{H}_{4}$ | Me | 3 | 3b/3b' | 98 | 75 |
| 3 | 2c | $4-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | Me | 3 | $3 \mathrm{c} / 3 \mathrm{c}^{\prime}$ | 99 | 75 |
| 4 | 2d | 4-FC6 $\mathrm{H}_{4}$ | Me | 1 | 3d/3d' | 99 | 75 |
| 5 | 2e | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | Me | 1 | $3 \mathrm{e} / 3 \mathrm{e}^{\prime}$ | 99 | 79 |
| 6 | $2 f$ | $3-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | Me | 1 | 3f/3f ${ }^{\prime}$ | 99 | 75 |
| 7 | 2 g | $4-\mathrm{BrC}_{6} \mathrm{H}_{4}$ | Me | 1 | $3 \mathrm{~g} / 3 \mathrm{~g}^{\prime}$ | 98 | 76 |
| 8 | 2h | 2-thienyl | Me | 3 | 3h/3h' | 95 | 77 |
| 9 | 2 i | Ph | Et | 3 | $3 \mathrm{i} / 3 \mathrm{i}^{\prime}$ | 96 | 76 |
| 10 | 2j | Ph | $i$-Pr | 3 | 3j/3j' | 95 | 74 |
| 11 | 2k | Ph | Allyl | 3 | 3k/3k' | 99 | 77 |
| 12 | 21 | $i-\operatorname{Pr}$ | Et | 9 | 31/31' | 74 | 80 |

${ }^{\text {a }}$ Reactions were carried out with $\mathbf{1 a}(0.1 \mathrm{mmol}), \mathbf{2 a - 2 k}(0.1 \mathrm{mmol})$, and $7(0.01 \mathrm{mmol})$ in THF $(1.0 \mathrm{~mL})$ at room temperature;
${ }^{\mathrm{b}}$ Isolated yields;
${ }^{\text {c }}$ Determined by chiral HPLC with a chiralpak AD-H column.

Furthermore cyclic 1,3-dicarbonyl compounds 1a-1f were examined as the nucleophiles in the reaction with $\mathbf{2 a}$. The results are summarized in Table 3. 6-Methyl-4-hydroxycoumarin $\mathbf{1 b}$ and 4-hydroxy-6-methyl-2H-pyranone $\mathbf{1 c}$ provided similar yields and enantioselectivities with $\mathbf{1 a}$ (Table 3, entries 1-3). 4-Hydroxy-6-methylpyrone 1d and cyclohexane-1,3-dione 1e were also suitable substrates. Excellent yields and enantioselectivities were achieved (Table 3, entries 4-5). 5,5-Dimethylcyclohexane-1,3-dione $\mathbf{1 f}$ afforded better enantioselectivity (Table 3, entry 6). The
results imply that the present catalytic method is generally applicable for various cyclic 1,3-dicarbonyl compounds.

Table 3. Asymmetric conjugate addition of 1a-f to $\mathbf{2 a}{ }^{a}$



1c

1d

$1 e$


| Entry | Substrate | Time (h) | Product | ${\text { Yield }(\%)^{b}}^{\text {b }}$ | Ee $(\%)^{\text {c }}$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 1 | $\mathbf{1 a}$ | 3 | $\mathbf{3 a} / \mathbf{3 a}^{\mathbf{\prime}}$ | 99 | 76 |
| 2 | $\mathbf{1 b}$ | 3 | $\mathbf{1 0 b} / \mathbf{1 0 b}$ | 99 | 79 |
| 3 | $\mathbf{1 c}$ | 3 | $\mathbf{1 0 c} / \mathbf{1 0 \mathbf { c } ^ { \prime }}$ | 99 | 74 |
| 4 | $\mathbf{1 d}$ | 3 | $\mathbf{1 0 d} / \mathbf{1 0 d ^ { \prime }}$ | 92 | 74 |
| 5 | $\mathbf{1 e}$ | 3 | $\mathbf{1 0 e} / \mathbf{1 0 \mathbf { e } ^ { \prime }}$ | 99 | 79 |
| 6 | $\mathbf{1 f}$ | 3 | $\mathbf{1 0 f / 1 0 f ^ { \prime }}$ | 97 | 85 |

${ }^{\text {a }}$ Reactions were carried out with $\mathbf{1 a - 1 f}(0.1 \mathrm{mmol})$, 2a $(0.1 \mathrm{mmol})$, and $7(0.01 \mathrm{mmol})$ in THF $(1.0 \mathrm{~mL})$ at room temperature;
${ }^{\mathrm{b}}$ Isolated yields;
${ }^{\mathrm{c}}$ Determined by HPLC with a chiralpak AD-H column.
The reaction is proposed to proceed via a bifunctional catalytic mechanism (Scheme 2). ${ }^{8}$ $\beta, \gamma$-unsaturated $\alpha$-keto-ester $\mathbf{2 a}$ is activated to nucleophilic attack through double hydrogen-bonding interactions with the thiourea group of catalyst 7. On the other hand, the tertiary amine group of 7 removes one proton of 4-hydroxycoumarin 2a. The resulting amino cation forms another hydrogen bond with the oxygen ion. The consequent nucleophilic attack gives $(R)-\mathbf{3 a}$ as the major product.


## Scheme 2

## Conclusions

We have developed an efficient asymmetric conjugate addition of cyclic 1,3-dicarbonyl compounds to $\beta, \gamma$-unsaturated $\alpha$-keto-esters. Bifunctional thiourea-tertiary amines were found to be suitable organocatalysts. The reaction provided the products in excellent yields and with good enantioselectivities for a variety of $\beta, \gamma$-unsaturated $\alpha$-keto-esters and cyclic 1,3-dicarbonyl compounds. Further attempts to improve the enantioselectivity and to apply this method for the preparation of valuable chiral products are currently underway.

## Experimental Section

General. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were recorded on Bruker AVANCE 400 spectrometer. Chemical shifts of protons are reported in parts per million downfield from tetramethylsilane ( $\delta$ $=0)$. Chemical shifts of carbon are referenced to the carbon resonances of the solvent $\left(\mathrm{CHCl}_{3}: \delta\right.$ $=77.0$ ). Peaks are labeled as singlet ( s ), doublet ( d ), triplet $(\mathrm{t})$, quartet $(\mathrm{q})$ and multiplet $(\mathrm{m})$. Optical rotations were measured on a Perkin-Elmer 341 digital polarimeter. Melting points were measured on a WRS-2A melting point apparatus and are uncorrected. The high resolution mass spectroscopic data were obtained at Shimadazu LCMS-IT-TOF spectrometer. Infrared (IR) spectra were recorded on a Bruker Tensor 37 spectrophotometer. Data are represented as follows: frequency of absorption ( $\mathrm{cm}^{-1}$ ), intensity of absorption ( $\mathrm{s}=$ strong, $\mathrm{m}=$ medium, $\mathrm{w}=$ weak). Enantiomeric excesses were determined by HPLC using a Daicel Chiralpak AD-H column and eluting with a $n$-hexane $/ i-\mathrm{PrOH}$ solution. Flash chromatography was performed over silica gel (230-400 mesh), purchased from Qingdao Haiyang Chemical Co., Ltd. Commercial reagents were used as received. $\beta, \gamma$-Unsaturated $\alpha$-keto-esters $\mathbf{2 a} \mathbf{- 2 k},{ }^{12} \mathbf{2 I}{ }^{13}$ and catalysts $\mathbf{4 - 9}{ }^{14}$ were
prepared according to the reported procedures.

## Typical procedure for organocatalytic asymmetric conjugate addition

A solution of 4-hydroxycoumarin $\mathbf{1 a}(0.1 \mathrm{mmol}),(E)$-methyl 2-oxo-4-phenylbut-3-enoate 2a ( 0.1 mmol ), Takemoto's catalyst $7(0.01 \mathrm{mmol})$ in THF $(1.0 \mathrm{~mL})$ was stirred at room temperature for 3 h . After the solvent was evaporated under vacuum, the residue was purified by flash column chromatography over silica gel to afford 3a as a white solid. Compound 3a was found to be in equilibrium with cyclic hemiketal 3a'. These two isomers were observed as separate compounds in ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra, but were not resolved by HPLC analysis.
4-(4-Hydroxy-oxo-2H-chromen-3-yl)- 4-phenyl-2-oxo-butyric acid methyl ester (3a). White solid ( $99 \%$ yield), exists in an equilibrium with cyclic hemiketal 3a', m.p. $170-172{ }^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{20}=$ -19.6 (c 1.0, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 2.46$ (d, $J=8.8 \mathrm{~Hz}, 1.34 \mathrm{H}$ ), 2.55 (dd, $J=$ $14.4,3.2 \mathrm{~Hz}, 0.33 \mathrm{H}), 2.80(\mathrm{ddd}, J=14.4,7.2,1.2 \mathrm{~Hz}, 0.33 \mathrm{H}), 3.90(\mathrm{~s}, 2.01 \mathrm{H}), 3.92$ (s, 0.99 H ), $4.20(\mathrm{t}, J=9.2 \mathrm{~Hz}, 0.67 \mathrm{H}), 4.36(\mathrm{dd}, J=7.2,3.2 \mathrm{~Hz}, 0.33 \mathrm{H}), 4.46(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 0.33 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $4.66(\mathrm{~s}, 0.67 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 7.23-7.38 (m, 7 H ), 7.51-7.58 (m, 1 H), 7.78 (dd, $J=8.0,1.6 \mathrm{~Hz}, 0.67 \mathrm{H}$ ), 7.83 (dd, $J=8.0,1.6 \mathrm{~Hz}, 0.33 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=33.7,34.5,35.6,38.1,53,8,53.9$, 95.5, 96.2, 102.9, 104.7, 115.0, 115.3, 116.5, 116.6, 122.8, 122.9, 123.8, 124.0, 126.6, 126.7, 127.1, 127.3, 128.4, 128.6, 131.8, 132.1, 141.5, 142.2, 152.8, 158.1, 158.5, 160.7, 161.6, 168.8, 168.9; IR (KBr): 3459 (s), 3181 (m), 3022 (w), 2952(w), 1756 (s), 1618(s), 1572 (s), 1491(s), $1453(\mathrm{~m}), 699(\mathrm{~s}) \mathrm{cm}^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{O}_{6} \mathrm{Na}(\mathrm{M}+\mathrm{Na})^{+}: 375.0845$, found: 375.0838. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min})$; $\mathrm{t}_{\mathrm{R}}($ major enantiomer) $=6.59 \mathrm{~min}$, $\mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=9.90 \mathrm{~min}, 76 \%$ ee.
4-(4-Hydroxy-2-oxo-2H-chromen-3-yl)-4-(4-methylphenyl)-2-oxo-butyric acid methyl ester (3b). White solid, $98 \%$ yield, exists in an equilibrium with cyclic hemiketal 3b'. m.p. 171-173 ${ }^{\circ} \mathrm{C}$, $[\alpha]_{\mathrm{D}}{ }^{20}=-25.7\left(\mathrm{c} 0.44, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=2.29(\mathrm{~s}, 1.02 \mathrm{H}), 2.31(\mathrm{~s}, 1.98$ H), $2.43(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1.32 \mathrm{H}), 2.51(\mathrm{dd}, J=14.4,3.2 \mathrm{~Hz}, 0.34 \mathrm{H}), 2.76(\mathrm{dd}, J=14.4,7.2 \mathrm{~Hz}$, 0.34 H ), 3.84 (s, 1.98 H ), 3.89 ( $\mathrm{s}, 1.02 \mathrm{H}$ ), 4.16 (t, $J=8.8 \mathrm{~Hz}, 0.66 \mathrm{H}$ ), 4.31 (dd, $J=7.2,3.2 \mathrm{~Hz}$, $0.34 \mathrm{H}), 4.61(\mathrm{~s}, 0.34 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $4.92(\mathrm{~s}, 0.66 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 7.07-7.11 (m, 4 H$)$, 7.23-7.35 (m, 2 H$)$, 7.48-7.57 (m, 1 H ), $7.77(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 0.66 \mathrm{H}), 7.83(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 0.34 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): \delta=20.1,33.2,34.1$, 35.6, 38.2, 53.8, 53.9, 95.4, 96.1, 103.0, 105.2, 115.1, 115.3, 116.6, 116.7, 122.7, 122.9, 123.7, $124.0,126.9,127.2,129.3,129.4,131.8,132.0,136.2,136.3,138.3,139.3,152.9,157.8,158.3$, 160.6, 161.6, 168.9, 169.0; IR (KBr): 3182 (m), 2920 (m), 2851 (w), 1753 (s), 1680 (s), 1619 (s), 1573 (s), 1513 (m), 1454 (m), $760(\mathrm{~m}) \mathrm{cm}^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{21} \mathrm{H}_{19} \mathrm{O}_{6}(\mathrm{M}+\mathrm{H})^{+}$: 367.1182, found: 367.1180. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min}) ; \mathrm{t}_{\mathrm{R}}($ major enantiomer $)=$ $7.01 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=10.90 \mathrm{~min}, 75 \%$ ee.
4-(4-Hydroxy-2-oxo-2H-chromen-3-yl)-4-(4-methoxyphenyl)- 2-oxo-butyric acid methyl
ester (3c). White solid, $99 \%$ yield, exists in an equilibrium with cyclic hemiketal 3c'. m.p. $147-148{ }^{\circ} \mathrm{C},[\alpha]_{D^{20}}=-35.6\left(\mathrm{c} 0.50, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=2.43(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, $1.80 \mathrm{H}), 2.50(\mathrm{dd}, J=14.4,3.2 \mathrm{~Hz}, 0.10 \mathrm{H}), 2.75(\mathrm{dd}, J=14.4,7.6 \mathrm{~Hz}, 0.10 \mathrm{H}), 3.76(\mathrm{~s}, 0.30 \mathrm{H})$, 3.77 (s, 2.70 H), $3.86(\mathrm{~s}, 2.70 \mathrm{H}), 3.90(\mathrm{~s}, 0.30 \mathrm{H}), 4.15(\mathrm{t}, J=8.8 \mathrm{~Hz}, 0.90 \mathrm{H}), 4.29(\mathrm{dd}, J=7.2$, $3.2 \mathrm{~Hz}, 0.10 \mathrm{H}), 4.58(\mathrm{~s}, 0.10 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $4.84(\mathrm{~s}, 0.90 \mathrm{H}$, OH , the position is concentration dependent), 6.81-6.86 (m, 2 H ), 7.13-7.17 (m, 2 H ), 7.24-7.35 $(\mathrm{m}, 2 \mathrm{H}), 7.49-7.53(\mathrm{~m}, 1 \mathrm{H}), 7.77(\mathrm{dd}, J=8.0,3.2 \mathrm{~Hz}, 0.90 \mathrm{H}), 7.83(\mathrm{dd}, J=8.0,3.2 \mathrm{~Hz}, 0.10$ H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=32.9,33.7,35.6,38.2,53.8,53.9,55.1,55.2,95.1,96.2$, $103.1,105.1,113.9,114.1,115.0,115.3,116.5,116.6,122.7,122.9,123.7,124.0,128.1,128.4$, $131.8,132.0,133.4,134.2,152.7,152.8,157.9,158.3,160.7,161.6,168.9$; IR (KBr): 3204 (m), 2997 (w), 2952 (w), 2840 (w), 1761 ( s), 1681 (s), 1573 (s), 1511 (m), 1455 (m), 759 (s) cm ${ }^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{21} \mathrm{H}_{19} \mathrm{O}_{7}(\mathrm{M}+\mathrm{H})^{+}: 383.1131$, found: 383.1142. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol = 70:30, $\lambda=254$ $\mathrm{nm}, 0.8 \mathrm{ml} / \mathrm{min}) ; \mathrm{t}_{\mathrm{R}}($ major enantiomer $)=8.36 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=14.06 \mathrm{~min}, 75 \%$ ee.
4-(4-Fluoro-phenyl)-4-(4-hydroxy-2-oxo-2H-chromen-3-yl)-2-oxo-butyric acid methyl ester (3d). White solid, $99 \%$ yield, exists in an equilibrium with cyclic hemiketal 3d'. m.p. 195-197 ${ }^{\circ} \mathrm{C}$, $[\alpha]_{D^{20}}=-26.2\left(\mathrm{c} 0.58, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=2.42-2.50(\mathrm{~m}, 1.65 \mathrm{H}), 2.79(\mathrm{dd}$, $J=14.4,7.6 \mathrm{~Hz}, 0.35 \mathrm{H}), 3.91(\mathrm{~s}, 1.95 \mathrm{H}), 3.93(\mathrm{~s}, 1.05 \mathrm{H}), 4.19(\mathrm{dd}, J=10.8,7.6 \mathrm{~Hz}, 0.65 \mathrm{H})$, $4.31(\mathrm{dd}, J=7.2,2.8 \mathrm{~Hz}, 0.35 \mathrm{H}), 4.55(\mathrm{~s}, 0.35 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $4.74(\mathrm{~s}, 0.65 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 6.94-7.01 (m, 2 H ), 7.19-7.21 (m, $2 \mathrm{H}), 7.31-7.37(\mathrm{~m}, 2 \mathrm{H}), 7.52-7.59(\mathrm{~m}, 1 \mathrm{H}), 7.78(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 0.65 \mathrm{H}), 7.82(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, $0.35 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta=33.0,33.9,35.4,38.1,54.0,54.1,95.4,95.9,102.8,104.8,115.0$, $115.1,115.2,115.3,115.4,115.7,116.6,116.7,122.8,122.9,123.8,124.1,128.5,128.6,128.9$, 129.0, 132.0, 132.2, 137.2, 137.9, 138.0, 152.8, 152.9, 158.0, 158.4, 160.4, 160.5, 161.5, 162.8, 162.9, 168.9, 169.0; IR (KBr): 3420 (s), 2956 (w), 2925(w), 2853 (w), 1717 (s), 1626 (s), 1575 (m), 1509 (s), 1456 (w), $761(\mathrm{~m}) \mathrm{cm}^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{O}_{6} \mathrm{~F}(\mathrm{M}+\mathrm{H})^{+}$: 371.0931, found: 371.0939. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min}) ; \mathrm{t}_{\mathrm{R}}($ major enantiomer) $=6.51 \mathrm{~min}$, $\mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=10.27 \mathrm{~min}, 75 \%$ ee.
4-(4-Chloro-phenyl)-4-(4-hydroxy-2-oxo-2H-chromen-3-yl)-2-oxo-butyric acid methyl ester (3e). White solid, $99 \%$ yield, exists in an equilibrium with cyclic hemiketal 3e'. m.p. 192-194 ${ }^{\circ} \mathrm{C}$, $[\alpha]_{D^{20}}=-37.6\left(\mathrm{c} 0.45, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=2.36-2.49(\mathrm{~m}, 1.64 \mathrm{H}), 2.79(\mathrm{dd}$, $J=14.4,7.2 \mathrm{~Hz}, 0.36 \mathrm{H}), 3.90(\mathrm{~s}, 1.92 \mathrm{H}), 3.92(\mathrm{~s}, 1.08 \mathrm{H}), 4.18(\mathrm{dd}, J=11.2,7.2 \mathrm{~Hz}, 0.64 \mathrm{H})$, $4.29(\mathrm{dd}, J=7.6,2.8 \mathrm{~Hz}, 0.36 \mathrm{H}), 4.60(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 0.36 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $4.80(\mathrm{~s}, 0.64 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 7.17-7.36 ( $\mathrm{m}, 6 \mathrm{H}$ ), $7.51-7.59(\mathrm{~m}, 1 \mathrm{H}), 7.77(\mathrm{dd}, J=8.0,1.6 \mathrm{~Hz}, 0.64 \mathrm{H}), 7.82(\mathrm{dd}, J=8.0,1.6 \mathrm{~Hz}, 0.36 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta=33.1,34.1,35.2,37.9,54.0,54.1,95.4,95.9,102.5,104.4,114.9,115.2$, 116.6, 116.7, 122.8, 122.9, 123.9, 124.1, 128.4, 128.5, 128.8, 128.9, 132.0, 132.2, 132.3, 132.4, $140.2,140.9,152.8,152.9,158.2,158.5,160.6,161.5,168.8,168.9$; IR (KBr): 3210 (m), 2958 (w), 2853 (w), 1756 (s), 1679 (s), 1619 (s), 1492 (m), 1456 (m), 767 (m) cm ${ }^{-1}$; HRMS (ESI)
calcd for $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{O}_{6} \mathrm{Cl}(\mathrm{M}+\mathrm{H})^{+}: 387.0635$, found: 387.0636. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}$, $0.8 \mathrm{ml} / \mathrm{min}) ; \mathrm{t}_{\mathrm{R}}($ major enantiomer $)=6.74 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=10.45 \mathrm{~min}, 79 \%$ ee.

## 4-(3-Chloro-phenyl)-4-(4-hydroxy-2-oxo-2H-chromen-3-yl)-2-oxo-butyric acid methyl ester

 (3f). White solid, $99 \%$ yield, exists in an equilibrium with cyclic hemiketal 3f'. m.p. $140-141{ }^{\circ} \mathrm{C}$, $[\alpha]_{\mathrm{D}}{ }^{20}=-29.7\left(\mathrm{c} 0.72, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta=2.41-2.48(\mathrm{~m}, 1.68 \mathrm{H}), 2.75(\mathrm{dd}, J=14.4$, $7.6 \mathrm{~Hz}, 0.32 \mathrm{H}), 3.81(\mathrm{~s}, 2.04 \mathrm{H}), 3.88(\mathrm{~s}, 0.96 \mathrm{H}), 4.16(\mathrm{dd}, J=10.6,7.2 \mathrm{~Hz}, 0.68 \mathrm{H}), 4.25(\mathrm{dd}, J$ $=7.2,3.2 \mathrm{~Hz}, 0.32 \mathrm{H}), 4.94(\mathrm{~s}, 0.32 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $5.30(\mathrm{~s}$, $0.68 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $7.10(\mathrm{~m}, 6 \mathrm{H}), 7.48-7.57(\mathrm{~m}, 1 \mathrm{H}), 7.77(\mathrm{dd}$, $J=8.0,1.2 \mathrm{~Hz}, 0.68 \mathrm{H}), 7.80(\mathrm{dd}, J=8.0,1.2 \mathrm{~Hz}, 0.32 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): \delta=33.5,34.3$, $35.3,37.8,537,53.9,95.5,96.0,102.1,103.8,114.8,115.1,116.4,116.6,122.8,122.9,123.9$, $124.0,125.5,125.7,126.6,126.9,127.1,127.6,129.4,129.8,132.0,132.2,133.9,134.2,143.9$, 144.3, 152.7, 158.5, 158.8, 160.8, 161.6, 168.6, 168.8; IR (KBr): 3401 (s), 2926 (w), 2853 (w), $1682(\mathrm{~s}), 1621(\mathrm{~s}), 1574(\mathrm{~m}), 1492(\mathrm{~m}), 1456(\mathrm{~m}), 759(\mathrm{~m}) \mathrm{cm}^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{O}_{6} \mathrm{Cl}(\mathrm{M}+\mathrm{H})^{+}: 387.0635$, found: 387.0638. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min}$ ); $\mathrm{t}_{\mathrm{R}}$ $($ major enantiomer $)=5.89 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=8.54 \mathrm{~min}, 75 \%$ ee.4-(4-Bromo-phenyl)-4-(4-hydroxy-2-oxo-2H-chromen-3-yl)-2-oxo-butyric acid methyl ester (3g). White solid, $98 \%$ yield, exists in an equilibrium with cyclic hemiketal $\mathbf{3 g}^{\prime}$. m.p. 198-199 ${ }^{\circ} \mathrm{C}$, $[\alpha]^{20}=-42.0\left(\mathrm{c} 0.40, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=2.36-2.49(\mathrm{~m}, 1.64 \mathrm{H}), 2.79$ (ddd, $J=14.4,7.2,1.6 \mathrm{~Hz}, 0.36 \mathrm{H}$ ), $3.90(\mathrm{~s}, 1.92 \mathrm{H}), 3.92(\mathrm{~s}, 1.08 \mathrm{H}), 4.16(\mathrm{dd}, J=11.2,7.2 \mathrm{~Hz}$, $0.64 \mathrm{H}), 4.28(\mathrm{dd}, J=7.2,2.8 \mathrm{~Hz}, 0.36 \mathrm{H}), 4.58(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 0.36 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $4.77(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 0.64 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 7.12-7.14 (m, 2 H), 7.28-7.43 (m, 4 H ), 7.51-7.60 (m, 1 H$), 7.77$ (dd, $J=8.0,1.6 \mathrm{~Hz}$, $0.64 \mathrm{H}), 7.81(\mathrm{dd}, J=8.0,1.6 \mathrm{~Hz}, 0.36 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=33.2,34.2,35.1$, $37.9,54.0,54.1,95.3,95.9,102.5,104.4,114.9,115.2,116.6,116.8,120.4,120.5,122.8,122.9$, 123.9, 124.1, 128.9, 129.3, 131.4, 131.8, 132.1, 132.3, 140.8, 141.5, 152.8, 152.9, 158.1, 158.5, $160.5,161.5,168.8,168.9$; IR (KBr): 3205 (m), 2956 (w), 1757 (s), 1678 (s), 1618 ( s$), 1571$ (s), $1491(\mathrm{~m}), 1456(\mathrm{~m}), 766(\mathrm{~s}) \mathrm{cm}^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{O}_{6} \mathrm{Br}(\mathrm{M}+\mathrm{H})^{+}$: 431.0130 , found: 431.0148. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min}$ ); $\mathrm{t}_{\mathrm{R}}$ (major enantiomer) $=7.04 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}$ $($ minor enantiomer $)=11.30 \mathrm{~min}, 76 \%$ ee.
4-Furan-2-yl-4-(4-hydroxy-2-oxo-2H-chromen-3-yl)-2-oxo-butyric acid methyl ester (3h). White solid, $95 \%$ yield, exists in an equilibrium with cyclic hemiketal $\mathbf{3 h}$ '. m.p. 147-149 ${ }^{\circ} \mathrm{C}$, $[\alpha]^{20}=19.6\left(\mathrm{c} 0.66, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=2.53(\mathrm{dd}, J=14.0,6.8 \mathrm{~Hz}, 0.60$ H), 2.59-2.69 (m, 1 H ), 2.78 (dd, $J=14.4,6.8 \mathrm{~Hz}, 0.40 \mathrm{H}), 3.82(\mathrm{~s}, 1.80 \mathrm{H}), 3.92(\mathrm{~s}, 1.20 \mathrm{H})$, $4.53(\mathrm{dd}, J=10.4,6.8 \mathrm{~Hz}, 0.60 \mathrm{H}), 4.60(\mathrm{dd}, J=6.8,2.4 \mathrm{~Hz}, 0.40 \mathrm{H}), 4.69(\mathrm{~s}, 0.40 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $5.00(\mathrm{~s}, 0.60 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 6.89-6.91 (m, 2 H ), 7.13-7.17 (m, 1 H ), 7.27-7.35 (m, 2 H ), 7.50-7.77 (m, 1 H ), 7.78 (dd, $J=8.0,1.6 \mathrm{~Hz}, 0.60 \mathrm{H}$ ), $7.82(\mathrm{dd}, J=8.0,1.6 \mathrm{~Hz}, 0.40 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$
$=29.0,30.1,35.5,38.5,53.8,54.0,95.6,95.9,102.9,104.4,114.9,115.1,116.5,116.7,122.9$, 123.0, 123.5, 123.8, 124.0, 124.1, 124.8, 125.1, 126.6, 132.1, 132.3, 145.1, 152.7, 152.8, 157.7, 157.9, 160.6, 161.6, 168.6; IR (KBr): 3353 (s), 3110 (w), 3075 (w), 2952 (w), 1739 (s), 1682 (s), 1625 (s), 1608 (m), 1491 (m), 1455 (m), 769 (m), 719 (m) cm ${ }^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{O}_{6} \mathrm{~S}(\mathrm{M}+\mathrm{H})^{+}: 359.0589$, found: 359.0587. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min}$ ); $\mathrm{t}_{\mathrm{R}}$ $($ major enantiomer $)=9.99 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=14.56 \mathrm{~min}, 77 \% \mathrm{ee}$.
4-(4-Hydroxy-oxo-2H-chromen-3-yl)-2-oxo-4-phenylbutyric acid ethyl ester (3i). White solid, $96 \%$ yield, exists in an equilibrium with cyclic hemiketal $3 i^{\prime}$. m.p. $167-168{ }^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{20}=-20.9$ (c $0.33, \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.35(\mathrm{dt}, J=7.2,2.0 \mathrm{~Hz}, 3 \mathrm{H}$ ), $2.44(\mathrm{~d}, J=9.2$ $\mathrm{Hz}, 1.12 \mathrm{H}$ ), 5.07 (dd, $J=14.4,3.2 \mathrm{~Hz}, 0.44 \mathrm{H}$ ), 2.80 (ddd, $J=14.4,7.2,0.8 \mathrm{~Hz}, 0.44 \mathrm{H}$ ), 4.20 (t, $J=9.2 \mathrm{~Hz}, 0.56 \mathrm{H}), 4.30-4.39(\mathrm{~m}, 2.44 \mathrm{H}), 4.53(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 0.44 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 4.76 ( $\mathrm{s}, 0.56 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $7.21-7.37(\mathrm{~m}, 7 \mathrm{H}), 7.50-7.59(\mathrm{~m}, 1 \mathrm{H}), 7.77(\mathrm{dd}, J=8.0,1.6 \mathrm{~Hz}, 0.56 \mathrm{H}), 7.82(\mathrm{dd}, J=8.0,1.6$ $\mathrm{Hz}, 0.44 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=13.9,14.0,33.7,34.6,35.5,38.1,63.4,63.5,95.4$, $96.1,102.9,104.9,115.1,115.3,116.5,115.7,122.7,122.8,127.3,124.0,126.6,126.7,127.1$, $127.4,128.4,128.7,131.8,132.0,141.6,142.4,152.7,152.8,158.1,158.5,160.7,161.6,168.4$, 168.5; IR (KBr): 3454 (m), 3084 (w), 3026 (w), 2986 (w), 2926 (w), 1751 (s), 1713 (s), 1681 (s), $1492(\mathrm{~m}), 1454(\mathrm{~m}), 756(\mathrm{~s}), 699(\mathrm{~m}) \mathrm{cm}^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{O}_{5}(\mathrm{M}+\mathrm{H})^{+}: 303.1232$, found: 303.1233. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min}) ; \mathrm{t}_{\mathrm{R}}($ major enantiomer) $=5.90 \mathrm{~min}$, $t_{R}($ minor enantiomer $)=8.70 \mathrm{~min}, 76 \%$ ee.
4-(4-Hydroxy-oxo-2H-chromen-3-yl)-2-oxo-4-phenylbutyric acid isobutyl ester (3j). White solid, $95 \%$ yield, exists in an equilibrium with cyclic hemiketal $3{ }^{3}$. m.p. $140-141{ }^{\circ} \mathrm{C},[\alpha]^{20}=$ -24.2 (c 0.36, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.29-1.33(\mathrm{~m}, 6 \mathrm{H}), 2.35-2.46(\mathrm{~m}, 1.36$ H), $2.50(\mathrm{dd}, J=14.4,3.2 \mathrm{~Hz}, 0.32 \mathrm{H}), 2.78(\mathrm{dd}, J=14.4,7.2 \mathrm{~Hz}, 0.32 \mathrm{H}), 4.19$ (dd, $J=11.2,7.2$ $\mathrm{Hz}, 0.68 \mathrm{H}), 4.32(\mathrm{dd}, J=7.2,3.2 \mathrm{~Hz}, 0.32 \mathrm{H}), 4.74(\mathrm{~s}, 0.32 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $5.05(\mathrm{~s}, 0.68 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 5.13-5.16 ( $\mathrm{m}, 1 \mathrm{H}$ ), 7.19-7.35 (m, 7 H ), 7.49-7.57 (m, 1 H ), 7.75 (dd, $J=8.0,1.2 \mathrm{~Hz}, 0.68 \mathrm{H}$ ), $7.80(\mathrm{dd}, J=8.0,1.2$ $\mathrm{Hz}, 0.32 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=21.5,33.7,34.7,38.0,71.8,71.9,95.3,96.0$, $102.9,104.9,115.1,115.4,116.5,116.6,122.7,122.8,123.7,124.0,126.6,126.7,127.1,127.4$, 128.3, 128.7, 131.8, 132.0, 141.6, 142.6, 152.8, 158.1, 160.6, 167.9, 168.0; IR (KBr): 3316 (s), 3058 (w), 2985 (w), 2942 (w), 1753 (s), 1674 (s), 1620 (s), 1492 (m), 1451 (m), 765 (s), 700 (m) $\mathrm{cm}^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{22} \mathrm{H}_{19} \mathrm{O}_{6}(\mathrm{M}-\mathrm{H})^{+}: 379.1182$, found: 379.1180. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda$ $=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min}) ; \mathrm{t}_{\mathrm{R}}($ major enantiomer $)=5.90 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=8.70 \mathrm{~min}$, $74 \%$ ee.
4-(4-Hydroxy-oxo-2H-chromen-3-yl)-2-oxo-4-phenylbutyric acid allyl ester (3k). White solid, $99 \%$ yield, exists in an equilibrium with cyclic hemiketal 3k'. m.p. $108-109{ }^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{20}=-18.6$ ( c $0.72, \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=2.46(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1.34 \mathrm{H}), 2.55(\mathrm{dd}, J=14.4$,
$3.2 \mathrm{~Hz}, 0.33 \mathrm{H}), 2.81(\mathrm{dd}, J=14.4,6.8 \mathrm{~Hz}, 0.33 \mathrm{H}), 4.20(\mathrm{t}, J=8.8 \mathrm{~Hz}, 0.67 \mathrm{H}), 4.33-4.36(\mathrm{dd}, J$ $=7.2,3.2 \mathrm{~Hz}, 0.33 \mathrm{H}), 4.55(\mathrm{~s}, 0.33 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 4.70-4.79 $(\mathrm{m}, 2 \mathrm{H}), 4.83(\mathrm{~s}, 0.67 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $5.30-5.40(\mathrm{~m}, 2 \mathrm{H})$, $5.86-5.96(\mathrm{~m}, 1 \mathrm{H}), 7.21-7.26(\mathrm{~m}, 7 \mathrm{H}), 7.34-7.56(\mathrm{~m}, 1 \mathrm{H}), 7.77(\mathrm{dd}, J=8.0,1.2 \mathrm{~Hz}, 0.67 \mathrm{H})$, 7.82 (dd, $J=8.0,1.2 \mathrm{~Hz}, 0.33 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=33.6,34.6,35.5,38.1,67.5$, $67.6,95.5,96.1,102.9,104.9,115.1,115.3,116.5,116.7,119.9,120.0,122.7,122.8,123.8$, $124.0,126.7,126.8,127.1,127.4,128.4,128.7,130.5,131.8,132.1,141.5,142.3,152.8,152.9$, 158.0, 158.4, 160.6, 161.6, 168.1, 168.2; IR (KBr): 3349 (m), 2929 (w), 1759 (s), 1678 (s), 1619 (s), 1572 (m), $1493(\mathrm{~m}), 1454(\mathrm{~m}), 786(\mathrm{~s}), 698(\mathrm{~m}) \mathrm{cm}^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{22} \mathrm{H}_{17} \mathrm{O}_{6}(\mathrm{M}-$ $\mathrm{H})^{+}: 377.1025$, found: 377.1032. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min})$; $\mathrm{t}_{\mathrm{R}}$ (major enantiomer $)=6.42 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=10.19 \mathrm{~min}, 77 \%$ ee.
4-(4-Hydroxy-oxo-2H-chromen-3-yl)-5-methyl-2-oxo-hexanoic acid ethyl ester (31). Colorless oil, $74 \%$ yield, exists in an equilibrium with cyclic hemiketal 31'. $[\alpha]_{D^{20}}=81.8$ (c 0.44, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 0.72(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2.1 \mathrm{H}), 0.95(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 0.9 \mathrm{H})$, $1.00-1.03(\mathrm{~m}, 3 \mathrm{H}), 1.26(\mathrm{t}, J=7.2 \mathrm{~Hz}, 0.9 \mathrm{H}), 1.43(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2.1 \mathrm{H}), 2.11(\mathrm{dd}, J=7.2,13.6$ $\mathrm{Hz}, 0.7 \mathrm{H}), 2.17-2.24(\mathrm{~m}, 0.7 \mathrm{H}), 2.34(\mathrm{~d}, J=5.6 \mathrm{~Hz}, 0.7 \mathrm{H}), 2.59-2.67(\mathrm{~m}, 0.3 \mathrm{H}), 2.91-2.96(\mathrm{~m}$, $0.3 \mathrm{H}), 3.07-3.13(\mathrm{~m}, 0.3 \mathrm{H}), 3.28-3.36(\mathrm{~m}, 0.3 \mathrm{H}), 4.30(\mathrm{q}, J=7.2 \mathrm{~Hz}, 0.6 \mathrm{H}), 4.44(\mathrm{qd}, J=0.4$, $7.2 \mathrm{~Hz}, 1.4 \mathrm{H}), 4.76(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 0.7 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $5.10(\mathrm{~s}$, $0.3 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $7.21-7.31(\mathrm{~m}, 2 \mathrm{H}), 7.47-7.53(\mathrm{~m}, 1 \mathrm{H}), 7.69$ (dd, $J=1.6,8.0 \mathrm{~Hz}, 0.7 \mathrm{H}$ ), 7.74 (dd, $J=1.6,8.0 \mathrm{~Hz}, 0.3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=$ $13.9,14.0,15.8,19.1,20.3,21.0,25.2,27.0,29.1,29.4,32.6,34.9,63.4,63.5,95.8,97.0,105.5$, 105.7, 115.2, 115.5, 116.2, 116.4, 122.6, 123.6, 123.8, 131.5, 131.6, 152.4, 152.5, 157.7, 157.9, 161.3, 162.4, 169.0, 169.2; HRMS (ESI) calcd for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{O}_{6}(\mathrm{M}-\mathrm{H})^{+}$: 331.1182, found: 331.1188. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min})$; $\mathrm{t}_{\mathrm{R}}($ major enantiomer) $=5.31 \mathrm{~min}$, $\mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=6.57 \mathrm{~min}, 80 \%$ ee.
4-(6-Methyl-4-hydroxy-2-oxo-2H-chromen-3-yl)-2-oxo-4-phenyl-butyric acid methyl ester (10b). White solid, $99 \%$ yield, exists in an equilibrium with cyclic hemiketal 10b'. m.p. $182-183{ }^{\circ} \mathrm{C},[\alpha]_{D^{20}}=-30.4\left(\mathrm{c} 0.70, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=2.41(\mathrm{~s}, 2.58 \mathrm{H})$, 2.43 (s, 0.42 H ), $2.44(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1.72 \mathrm{H}), 2.52(\mathrm{dd}, J=14.4,3.6 \mathrm{~Hz}, 0.14 \mathrm{H}), 2.78$ (ddd, $J=$ $14.4,7.2,1.2 \mathrm{~Hz}, 0.14 \mathrm{H}$ ), 3.85 ( $\mathrm{s}, 2.68 \mathrm{H}$ ), 3.91 ( $\mathrm{s}, 0.42 \mathrm{H}$ ), 4.18 (t, $J=8.8 \mathrm{~Hz}, 0.86 \mathrm{H}$ ), 4.32 (dd, $J=7.2,3.2 \mathrm{~Hz}, 0.14 \mathrm{H}), 4.53(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 0.14 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $4.82(\mathrm{~s}, 0.86 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $7.17-7.37(\mathrm{~m}, 7 \mathrm{H})$, $7.55(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 0.86 \mathrm{H}), 7.60(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 0.14 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=20.9$, $33.7,34.6,35.6,38.1,53.8,53.9,95.4,96.1,102.7,104.7$, 114.7, 114.9, 116.3, 116.4, 122.4, $122.5,126.7,127.1,127.3,128.4,128.6,132.9,133.1,133.5,133.8,141.6,142.4,150.9,151.0$, 158.0, 158.4, 160.9, 161.8, 169.0; IR (KBr): 3298 (s), 3065 (w), 2999 (w), 2951 (w), 2924 (w), 1761 (s), 1672 (s), 1623 (s), 1494 (m), 1449 (m), 780 (m), 699 (m) cm ${ }^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{21} \mathrm{H}_{17} \mathrm{O}_{6}(\mathrm{M}-\mathrm{H})^{+}: 365.1025$, found: 365.1031. The enantiometric excess was determined by

HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min}$ ); $\mathrm{t}_{\mathrm{R}}($ major enantiomer $)=6.37 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=13.62 \mathrm{~min}, 79 \%$ ee.
4-(4-Hydroxy-6-methyl-2-oxo-2H-pyran-3-yl)-2-oxo-4-phenyl-butyric acid methyl ester (10c). White solid, $99 \%$ yield, exists in an equilibrium with cyclic hemiketal $10{ }^{\prime}$ '. m.p. $50-52^{\circ} \mathrm{C}$, $[\alpha]^{20}=32.5\left(\mathrm{c} 0.60, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=2.19(\mathrm{~s}, 2.10 \mathrm{H}), 2.22(\mathrm{~s}, 0.90 \mathrm{H})$, 2.31 (d, $J=8.4 \mathrm{~Hz}, 1.40 \mathrm{H}$ ), 2.38 (dd, $J=14.4,4.4 \mathrm{~Hz}, 0.30 \mathrm{H}$ ), $2.64(\mathrm{dd}, J=14.4,7.2 \mathrm{~Hz}, 0.30$ H), $3.66(\mathrm{~s}, 2.10 \mathrm{H}), 3.81(\mathrm{~s}, 0.90 \mathrm{H}), 4.01(\mathrm{t}, J=8.4 \mathrm{~Hz}, 0.70 \mathrm{H}), 4.11(\mathrm{dd}, J=7.2,4.4 \mathrm{~Hz}, 0.30$ $\mathrm{H}), 4.76(\mathrm{~s}, 0.30 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $5.23(\mathrm{~s}, 0.70 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $5.83(\mathrm{~s}, 0.70 \mathrm{H}), 5.87(\mathrm{~s}, 0.30 \mathrm{H}), 7.14-7.28(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta=19.6,19.7,33.1,33.6,36.0,37.9,53.4,53.6,95.3,96.0,99.8,99.9,100.0$, 101.1, 126.4, 126.5, 127.0, 127.2, 128.2, 128.4, 141.6, 141.9, 161.3, 161.4, 162.8, 163.1, 163.3, 163.5, 168.6, 168.8; IR (KBr): 3391 (s), 3028 (w), 2955 (w), 2925 (w), 1690 (s), 1579 (s), 1494 (m), 1447 (m), 761 (m), $701(\mathrm{~m}) \mathrm{cm}^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{O}_{6}(\mathrm{M}-\mathrm{H})^{+}$: 315.0869, found: 315.0862. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254 \mathrm{~nm}, 0.8 \mathrm{ml} / \mathrm{min}) ; \mathrm{t}_{\mathrm{R}}($ major enantiomer $)=6.44 \mathrm{~min}$, $\mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=8.51 \mathrm{~min}, 74 \%$ ee.
4-(4-Hydroxy-1-methyl-2-oxo-1,2-dihydroquinolin-3-yl)-2-oxo-4-phenylbutyric acid methyl ester (10d). White solid, $99 \%$ yield, exists in an equilibrium with cyclic hemiketal 10d'. m.p. $208-210^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}} 20=26.3\left(\mathrm{c} 0.38, \mathrm{CH}_{3} \mathrm{OH}\right) .{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz},\left(\mathrm{CD}_{3}\right) \mathrm{CO}\right): \delta=2.41(\mathrm{dd}, J=$ $14.0,5.6 \mathrm{~Hz}, 0.30 \mathrm{H}), 2.52(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1.40 \mathrm{H}), 2.76(\mathrm{dd}, J=14.0,7.6 \mathrm{~Hz}, 0.30 \mathrm{H}), 3.51(\mathrm{~s}$, $2.10 \mathrm{H}), 3.57(\mathrm{~s}, 2.10 \mathrm{H}), 3.58(\mathrm{~s}, 0.90 \mathrm{H}), 3.80(\mathrm{~s}, 0.90 \mathrm{H}), 4.26(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{~s}, 0.30$ $\mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $6.93(\mathrm{~s}, 0.70 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 7.10-7.30 (m, 6 H ), 7.50-7.52 (m, 1 H ), 7.62-7.67 (m, 1 H ), 7.98-8.00 (m, 1 H ); (100 MHz, DMSO- $d_{6}$ ): $\delta=28.9,34.7,35.4,38.3,38.5,38.9,52.0,52.7,95.6,96.9$, $107.3,108.1,114.5,114.8,121.5,122.4,125.4,125.6,127.1,127.3,1277.8,130.9,138.7,138.8$, $142.9,144.2,154.8,160.8,168.5,168.9$; HRMS (ESI) calcd for $\mathrm{C}_{21} \mathrm{H}_{18} \mathrm{NO}_{5}(\mathrm{M}-\mathrm{H})^{+}: 364.1185$, found: 364.1190. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane: 2 -propanol $=70: 30, \lambda=220 \mathrm{~nm}, 0.7 \mathrm{ml} / \mathrm{min}$ ); $\mathrm{t}_{\mathrm{R}}($ major enantiomer $)=8.38 \mathrm{~min}$, $\mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=14.3 \mathrm{~min}, 74 \%$ ee.
4-(2-Hydroxy-6-oxocyclohex-1-enyl)-2-oxo-4-phenyl-butyric acid methyl ester (10e). White solid, $99 \%$ yield, exists in an equilibrium with cyclic hemiketal 10e'. m.p. $144-145{ }^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{20}=$ $12.8\left(\mathrm{c} 0.58, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.98-2.12(\mathrm{~m}, 2 \mathrm{H}), 2.23-2.59(\mathrm{~m}, 6 \mathrm{H})$, 3.78 (s, 2.34 H), 3.85 (s, 0.66 H ), 3.90 (dt, $J=8.8,1.6 \mathrm{~Hz}, 0.78 \mathrm{H}$ ), 4.10-4.13 (m, 0.22 H), 4.16 (s, $0.22 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), $4.48(\mathrm{~s}, 0.78 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 7.14-7.18 (m, 3 H ), 7.24-7.28 (m, 2 H ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 20.2$, 20.7, 28.7, 28.9, 31.5, 33.2, 35.7, 36.9, 37.0, 38.31, 53.5, 53.6, 94.7, 95.6, 113.1, 115.5, $126.1,126.3,126.9,127.2,128.3,128.4,142.7,144.0,168.5,169.4,196.2,196.8$; IR (KBr): 3066 (m), 2952 (m), 1759 ( s), 1637 (s), 1609 ( s$), 1496$ (m), 1454 (m), 760 (m), 705 (m) cm ${ }^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{O}_{5}(\mathrm{M}+\mathrm{H})^{+}: 303.1232$, found: 303.1233. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane:2-propanol $=70: 30, \lambda=254$
$\mathrm{nm}, 0.8 \mathrm{ml} / \mathrm{min}) ; \mathrm{t}_{\mathrm{R}}($ major enantiomer $)=6.07 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=7.43 \mathrm{~min}, 79 \%$ ee.

4-(2-Hydroxy-4,4-dimethyl-6-oxocyclohex-1-enyl)-2-oxo-4-phenyl-butyric acid methyl ester (10f). White solid, $97 \%$ yield, exists in an equilibrium with cyclic hemiketal 10f'. m.p. $161-162{ }^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{20}=-15.0\left(\mathrm{c} 0.28, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=1.08(\mathrm{~s}, 1.95 \mathrm{H})$, $1.11(\mathrm{~s}, 1.05 \mathrm{H}), 1.17(\mathrm{~s}, 1.95 \mathrm{H}), 1.19(\mathrm{~s}, 1.05 \mathrm{H}), 2.01-2.58(\mathrm{~m}, 6 \mathrm{H}), 3.72(\mathrm{~s}, 1.95 \mathrm{H}), 3.84(\mathrm{~s}$, $1.05 \mathrm{H}), 3.86-3.91(\mathrm{~m}, 0.65 \mathrm{H}), 4.07-4.08(\mathrm{~m}, 0.35 \mathrm{H}), 4.32(\mathrm{~s}, 0.35 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 4.68 ( $\mathrm{s}, 0.65 \mathrm{H}, \mathrm{OH}$, the position is concentration dependent), 7.13-7.17 (m, 3 H ), 7.23-7.27 (m, 2 H ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=27.7$, 28.2, 28.8, 29.3, $29.7,31.7,31.9,32.1,33.3,36.0,38.3,42.3,42.4,50.8,53.5,53.6,94.9,95.8,112.0,113.9,126.1$, 126.3, 12.0, 127.3, 128.3, 128.4, 142.9, 144.0, 166.9, 167.5, 169.3, 169.4, 196.3, 196.8; IR (KBr): 3129 (s), 2953 (w), 2932 (w), 2887 (w), 1756 (s), 1596 (s), 1494 (s), 1451 (m), 762 (m), 696 (m) $\mathrm{cm}^{-1}$; HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{O}_{5}(\mathrm{M}-\mathrm{H})^{+}: 329.1389$, found: 329.1396. The enantiometric excess was determined by HPLC with a Chiralpak AD-H column (hexane: 2-propanol =70:30, $\lambda$ $=254 \mathrm{~nm}, 0.7 \mathrm{ml} / \mathrm{min}) ; \mathrm{t}_{\mathrm{R}}($ major enantiomer $)=6.22 \mathrm{~min}, \mathrm{t}_{\mathrm{R}}($ minor enantiomer $)=8.69 \mathrm{~min}, 85 \%$ ee.

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## References

1. For the reviews of catalytic asymmetric conjugate addition, see: (a) Yamaguchi, M.; Jacobsen, E. N.; Pfalz, A.; Yamamoto, H.; Eds. Comprehensive Asymmetric Catalysis, Springer: Berlin, 1999, Ch. 31. p. 2. (b) Sibi, M. P.; Manyem, S. Tetrahedron 2000, 56, 8033. (c) Krause, N.; Hoffmann-Roeder, A. Synthesis 2001, 171.
2. For the general reviews of asymmetric organocatalysis, see: (a) Dondoni, A.; Massi, A. Angew. Chem., Int. Ed. 2008, 47, 4638. (b) Yu, X.-H.; Wang, W. Chem. Asian J. 2008, 3, 516. (c) MacMillan, D. W. C. Nature 2008, 455, 304. (c) Guillena, G.; Najera, C.; Ramon, D. J. Tetrahedron: Asymmetry 2007, 18, 2249.
3. For the recent reviews of organocatalytic asymmetric conjugate addition, see: (a) Almasi, D.; Alonso, D. A.; Najera, C. Tetrahedron: Asymmetry 2007, 18, 299. (b) Tsogoeva, S. B. Eur. J.

Org. Chem. 2007, 11, 1701. (c) Chen, Y.-C. Synlett 2008, 1919. (d) Mukherjee, S.; Yang, J.-W.; Hoffman, S.; List, B. Chem. Rev. 2007, 107, 5471. (e) Pellissier, H.; Tetrahedron 2007, 63, 9267.
4. Halland, N.; Velgaard, T.; Jørgensen, K. A. J. Org. Chem. 2003, 68, 5067.
5. Calter, M. A.; Wang, J. Org. Lett. 2009, 11, 2205.
6. Zhou, L.; Lin, L. L.; Wang, W. T.; Ji, J.; Liu, X. H.; Feng, X. M. Chem. Commun. 2010, 3601.
7. Xu, D.-Q.; Wang, Y.-F.; Zhang, W.; Luo, S.-P.; Zhong, A.-G.; Xia, A.-B.; Xu, Z.-Y. Chem. Eur. J. 2010, ASAP (DOI: 10.1002/chem.201000094).
8. For the reviews of bifunctional organocatalysts: see: (a) Connon, S. J. Chem.Commun. 2008, 2499. (b) Connon, S. J. Chem. Eur. J. 2006, 12, 5418. (c) Taylor, M. S.; Jacobsen, E. N. Angew. Chem. Int. Ed. 2006, 45, 1520.
9. (a) Zhang, X.-J.; Liu, S.-P.; Li, X.-M.; Yan, M; Chan, A. S. C. Chem. Commun. 2009, 833; (b) Xuan, Y.-N.; Nie, S.-Z.; Dong, L.-T.; Zhang, J.-M.; Yan, M. Org. Lett. 2009, 11, 1583; (c) Liu, S.-P.; Zhang,X.-J.; Lao, J.-H.; Yan, M. Arkivoc 2009, 7, 268. (d) Zhang, X.-J.; Liu, S.-P.; Lao, J.-H.; Du, G.-J.; Yan, M.; Chan, A. S. C. Tetrahedron: Asymmetry 2009, 20, 1451.
10. (a) Okino, T.; Hoashi, Y.; Takemoto, Y. J. Am. Chem. Soc. 2003, 125, 12672. (b) Okino, T.; Hoashi, Y.; Furukawa, T.; Xu, X.; Takemoto, Y. J. Am. Chem. Soc. 2005, 127, 119. (c) Hoashi, Y.; Okino, T.; Takemoto, Y. Angew. Chem., Int. Ed. 2005, 44, 4032.
11. For the biological and pharmaceutical activities of coumarin deratives, see: (a) Murray, R. D. H.; Medez, J.; Brown, S. A. The Natural Coumarins: Occurrence, Chemistry and Biochemistry, Wiley: New York, 1982. (b) The Handbook of Natural Flavonoids, Harborne, J. B.; Baxter H.; Eds. Wiley: Chichester, UK, 1999. (c) Manolov, I.; Danchev, N. D. Eur. J. Med. Chem. 1995, 30, 531. (d) Xie, L.; Takeuchi, Y.; Cosentino, L. M.; Lee, K. H. J. Med. Chem. 1999, 42, 2662. (e) Ishikawa, T. Heterocycles 2000, 53, 453. (f) Xie, L.; Takeuchi, Y.; Cosentino, L. M.; McPhail, A. T.; Lee, K. H. J. Med. Chem. 2001, 44, 664. (g) Melliou,E.; Magiatis, P.; Mitaku, S.; Skaltsounis, A. L.; Chinou, E.; Chinou, I. J. Nat. Prod. 2005, 68, 78.
12. (a) Stecher, E. D.; Ryder, H. F. J. Am. Chem. Soc. 1952, 74, 4392. (b) Audrain,H.; Thorhauge, J.; Hazell, R. G.; Jørgensen, K. A. J. Org. Chem. 2002, 65, 4487.
13. Sugimura, H.; Miura, M.; Yamada, N. Tetrahedron: Asymmetry 1997, 8, 4089.
14. (a) Kaik, M.; Gawroński, J. Tetrahedron: Asymmetry 2003, 14, 1559. (b) Li, B.-J.; Jiang, L.; Liu, M.; Chen, Y.-C.; Ding, L.-S.; Wu, Y. Synlett 2005, 603. (c) Okino, T.; Hoashi,Y.; Takemoto, Y. J. Am. Chem. Soc. 2003, 125, 12672. (d) McCooey, S. H.; Connon, S. J. Angew. Chem., Int. Ed. 2005, 44, 6367. (e) Luo, J.; Xu, L.-W.; Hay, R. A. S.; Lu, Y.-X. Org. Lett. 2009, 11, 437.

