Synthesis and antifungal activity of novel polyheterocyclic compounds containing fused 1,2,4-triazine moiety

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Abstract

3-Amino-4-(4-chlorophenyl)-7-hydrazino-8*H*-pyrazolo[4,3-*e*][1,2,4]triazolo[1`,5`-*a*]pyridine-5-carbonitrile (**4**) was synthesized from 4-(4-chlorophenyl)-1,6-diamino-2-oxo-1,2-dihydropyridine-3,5-dicarbonitrile (**1**). Reaction of **4** with α,β -bifunctional compounds gave pyrazolotriazinotriazolopyridines (**8-14**). The behavior of **4** towards condensation reactions with indole-2,3-dione in different media gave different products **15-16**. Acetylation of **16** led to different products depending on the reaction conditions. Structures of the products have been deduced from analytical and spectral data (UV, IR, ¹H NMR, ¹³C NMR and mass spectra). Some of the products were screened for antifungal activity.

Keywords: Synthesis, *o*-diamine, triazolopyridine, pyrazolotriazinotriazolopyridine, fungicidal activity

Introduction

Polyfunctional pyridines are highly reactive reagents that have been used extensively in heterocyclic synthesis¹⁻³ and that possess biological as well as pharmacological activity.⁴⁻⁶ Triazolopyridines are also interesting compounds due to their pronounced biological activity, as they can be used as antidepressants.⁷⁻⁸ Various 1,2,4-triazine derivatives are well known to possess an array of physiological activities, such as anticancer, muscle relaxant, hypnotic, anti-inflammatory, diuretic and antihypertensive activities.⁹⁻¹²

In continuation of our work in the area of fused 1,2,4-triazines¹³⁻¹⁹ and their heterocyclization via ring closing reactions with α,β -bifunctional reagents,²⁰⁻²⁵ the present work aimed at the synthesis of fused heteropolycyclic nitrogen systems containing a fused

1,2,4-triazine moiety starting from 4-(4-chlorophenyl)-1,6-diamino-2-oxo-1,2dihydropyridine-3,5-dicarbonitrile (1) and evaluation of their antifungal activity.

Results and Discussion

The hydrazine derivative 4, as starting material for polyfused heterocyclic systems, was obtained from the reaction of 4-(4-chlorophenyl)-1,6-diaminopyridine-3,5-dicarbonitrile (1) with CS₂/KOH to give thioxotriazolopyridinone 2 followed by hydrazinolysis. Also, methylation of 2 with MeI/KOH produced the 2-methylthio derivative 3 which on hydrazinolysis afforded 4^{26-27} On the other hand, compound 4 was also obtained using alternative pathways. Treatment of diamine 1 with hydrazine hydrate afforded 4-(4chlorophenyl)-3,6,7-triamino-7*H*-pyrazolo [3,4-*b*]pyridine-5-carbonitrile (5). Heating an ethanolic solution of 5 with CS₂/KOH at reflux yielded thioxopyrazolotriazolopyridine 6 which on methylation yielded the methylthio derivative 7. Hydrazinolysis of 6 and/or 7 furnished the hydrazino derivative 4 (Scheme 1). Compound 4 was confirmed by its elemental analysis and spectral data. The IR spectrum revealed the disappearance of the absorption band at 1669 cm⁻¹ assigned to the C=O group of the pyridinone ring in compound **1** and a new band was observed at 1637 cm⁻¹ corresponding to an azomethine group in addition to the hydrazino group at 3469 and 3308 cm⁻¹. Its ¹³C NMR spectrum showed one signal at δ 117.1 ppm corresponding to one cyano group. The mass spectrum of 4 showed a molecular ion peak at m/e 340 with the base peak at m/e 285, presumably due to the higher stability of the fused pyrazolopyridine system.



Scheme 1. Synthetic pathway for the preparation of target compound 4.

The target compound **4** was used for the synthesis of polyfused systems. Thus, fused triazinones **8a-c** were obtained from cyclocondensation of **4** with α -oxoacids, namely; glyoxalic, pyruvic, *p*-chlorostyrylglyoxalic acid in glacial acetic acid^[28] (Scheme 2).

The isomeric fused triazinones **9** and **10** were obtained from cyclocondensation of compound **4** with monochloroacetic acid and/or chloroacetyl chloride,²⁹ respectively (Scheme 2). ¹³C NMR spectra of compounds **9** and **10** showed characteristic signals at δ 35.7 and 35.9 ppm assigned to CH₂ carbons, respectively (Figure 1).

Perhydro fused 1,2,4-triazinotriazoles **11** and **12** were obtained from boiling compound **4** with phenacyl bromide and 1,2-dibromoethane³⁰⁻³¹ in ethanolic NaOH (5%), respectively (Scheme 2).



Scheme 2. Synthetic pathway for the preparation of compounds 8-12.



Figure 1. ¹³C NMR of compounds 9 and 10.

Reactions of 4 with α , β -dicarbonyl compounds have been investigated. Thus, treatment of 4 with diethyl oxalate³² in boiling DMF produced the 9,10-dioxo derivative 13. Also, cyclocondensation of 4 with benzoin in glacial acetic acid yielded the 9,10-diphenyl derivative 14 (Scheme 3). On the other hand, cyclocondensation of compound 4 with indole-3-amino-4-(4-chlorophenyl)-7H-indolo[2,3-2,3-dione in boiling DMF afforded *e*]pyrazolo[3``,4``:6`,5`] pyrido[1`,2`:2,3][1,2,4]triazolo[5,1-*c*][1,2,4]triazine-5-carbonitrile (15), while treatment of an equimolar ratio of 4 and indole-2,3-dione in ethanolic NaOH solution 3-amino-9-(2-aminophenyl)-4-(4-chlorophenyl)-10-oxo-12-hydro-7Hafforded pyrazolo[4,3-e][1,2,4]triazino[3`,4`-5``,1``][1,2,4]triazolo[2``,3``-a]pyridine-5-carbonitrile (16) (Scheme 3). Compound 15 was obtained authentically by cyclization of 16 in glacial acetic acid and a few drops of concentrated H₂SO₄, where the absorption band of C=O group disappeared and showed co-identical IR spectra. Acetylation of compound 16 using acetic anhydride furnished the diacetyl derivative 17. The IR spectrum revealed the presence of new absorption bands at 1720, 1655 cm⁻¹ for two C=O groups and at 3422 cm⁻¹ for NH group (Scheme 3).

Finally, the behavior of compound **16** towards acetylation reactions has been studied under different conditions. Thus, refluxing **16** with glacial acetic acid afforded the monoacetyl derivative **18**. However, when the reaction was carried out in boiling glacial acetic acid containing a few drops of acetic anhydride, the diacetyl derivative **19** was isolated (Scheme 4). Structures of mono- and diacetyl derivatives **18** and **19** were established form their elemental analysis and spectral data. IR spectrum of **18** showed an absorption band at 1764 cm⁻¹ assigned to one C=O, while **19** showed two absorption bands at 1740 and 1702 cm⁻¹ assigned to two C=O functions. ¹H NMR of **18** showed a signal at δ 1.91 ppm, characteristic for one COCH₃ group, while that of **19** showed two signals at δ 1.56 and 1.92 ppm, characteristic for two COCH₃ groups.

Fungicidal activity

Some new synthesized compounds were screened for their antifungal activities against two fungi, *Alternaria alternata* and *Aspergillus niger* using the disc diffusion method.³³⁻³⁴ The tested compounds were dissolved in DMF [which has no inhibition activity] to get 1 mg/ml solution. The antibiotic flucanazole was used as standard antifungal reference. The inhibition zones of the microbial growth surrounding the filter paper disc (2.5 mm) were measured in millimeters at the end of an incubation period at 30 °C for 3 days (Table 1).

All the tested compounds showed variable activities toward the two species in comparison to the standard flucanazole which revealed that these compounds are biologically active due to the presence of different heterocycles and functional groups.

From the results obtained, it is clear that most of the tested compounds showed moderate activity toward the tested fungi except compound **13** showed higher activity towards *Alternaria alternata* fungi which mainly due to the expected dihydroxy structure (Table 1)



Scheme 3. Synthetic pathway for the preparation of compounds 13-17.



Scheme 4. Synthetic pathway for the preparation of compounds 18-19.

Compd. No.	Diameter of inhibition zone (mm)	
	Alternaria alternata	Aspergillus niger
4	++	+
5	+	+
6	++	+
8c	++	++
9	+	++
10	+	++
13	+++	+
15	++	++
16	++	+
19	++	++
(Flucanazole)	+++	+++

 Table 1. Fundicidal activity of some of the prepared compounds 2-19.

Lower active = + (inhibition zone 1–10 mm), Moderately active = ++ (inhibition zone 11–25 mm) and High active = +++ (inhibition zone > 25 mm).

Experimental Section

General Procedures. Melting points are uncorrected and were recorded in open capillary tubes on a Stuart SMP3 melting point apparatus. Infrared spectra were recorded on FT-IR Bruker Vector 22 spectrophotometer using KBr wafer technique. UV absorption spectra (DMF) were recorded on a Jasco model (V-550) UV spectrophotometer. ¹H NMR spectra were measured on Gemini spectrometer 200 MHz and AC spectrometer 250 MHz using DMSO-*d*₆ as solvent and TMS (chemical shift in δ ppm) as an internal standard. ¹³C NMR spectra were measured on AC spectrometer 250 MHz using DMSO as solvent and TMS (chemical shift in δ ppm) as an internal standard. ¹³C NMR spectra were measured on AC spectrometer 250 MHz using DMSO as solvent and TMS (chemical shift in δ ppm) as an internal standard. Mass spectra were obtained using gas chromatography GCMS qp 1000 ex Schimadzu instrument mass spectrometer (70 eV). Elemental microanalyses were performed at the Cairo University Microanalytical Center. 4- (4-chlorophenyl)-1,6-diamino-2-oxo-1,2-dihydropyridine-3,5-dicarbonitrile (1) has been prepared according to the reported method.³⁵

7-(4-Chlorophenyl)-5-oxo-2-thioxo-1,2,3,5-tetrahydro[1,2,4]triazolo[1,5-a]pyridine-6,8-

di-carbonitrile (2). A mixture of **1** (2.85 g, 0.01 mol) and carbon disulfide (0.60 mL, 0.01 mol) in ethanolic potassium hydroxide (10%, 100 mL) was refluxed for 4 hours, after cooling the reaction mixture was poured onto ice-HCl. The solid obtained was filtered, washed several times with water and crystallized from methanol to give **2** as yellow crystals, yield 2.13 g (65%), mp > 300 °C. UV λ_{max} (log ε): 346 (4.18), 275 nm (2.405). IR (KBr, cm⁻¹): 3212, 3184 (2 NH), 2216 (2 C=N), 1674 (C=O), 1284 (C=S). ¹H NMR (δ , DMSO-*d*₆): 7.64 (d, 2H, Ar-H, J = 8.4 Hz), 7.82 (d, 2H, Ar-H, J = 8.4 Hz), 9.24 (s, 1H, NH exchangeable with D₂O), 10.36 ppm (s, 1H, NH exchangeable with D₂O). Anal. Calcd. for C₁₄H₆ClN₅OS (327.76): C, 51.26; H, 1.83; N, 21.36; S, 9.76. Found C, 51.12; H, 1.80; N, 20.95; S, 9.55.

7-(4-Chlorophenyl)-2-methylthio-5-oxo-,3,5-dihydro[1,2,4]triazolo[1,5-*a*]pyridine-6,8-dicarbonitrile (3). A mixture of 2 (1.64 g, 0.005 mol) and methyl iodide (0.31 mL, 0.005 mol) in ethanolic potassium hydroxide (10%, 100 mL) was refluxed for 4 hours, after cooling the reaction mixture was poured onto ice-HCl. The solid obtained was filtered, wash several times with water and crystallized from aqueous DMF to give **3** as yellow crystals, yield 1.14 g (67%), mp > 300 °C. IR (KBr, cm⁻¹): 3209 (NH), 2960 (CH₃), 2220 (2 C=N), 1672 (C=O), 1640 (C=N). ¹H NMR (δ , DMSO-*d*₆): 4.14 (s, 3H, CH₃), 8.31 (d, 2H, Ar-H), 8.42 (d, 2H, Ar-H), 8.61 ppm (s, 1H, NH exchangeable with D₂O).

4-(4-Chlorophenyl)-3,6,7-triamino-7H-pyrazolo[3,4-*b***]pyridine-5-carbonitrile** (5). A mixture of **1** (2.85 g, 0.01 mol) and hydrazine hydrate (5 mL) was refluxed for 6 hours, after cooling the reaction mixture was poured onto ice. The solid obtained was filtered and crystallized from aqueous DMF to give **5** as yellow crystals, yield 1.44 g (48%), mp > 300 °C. UV λ_{max} (log ε); 395 (3.00), 350 (4.5), 275 nm (4.3). IR (KBr, cm⁻¹): 3470, 3303, 3188 (3 NH₂), 2214 (C=N), 1637 (C=N). ¹H NMR (δ , DMSO-*d*₆): 4.59 (bs, 4H, 2NH₂ exchangeable

with D₂O), 5.29 (bs, 1H, NH exchangeable with D₂O), 5.76 (bs, 1H, NH exchangeable with D₂O), 7.56 (d, 2H, Ar-H, J = 8.6 Hz), 7.67 ppm (d, 2H, Ar-H, J = 8.6 Hz). MS (Int.%): 299 (100), 300 (48.05), 284 (22.77), 189 (7.11), 172 (5.49), 111 (23.09). Anal. Calcd. for $C_{13}H_{10}ClN_7$ (299.72): C, 52.05; H, 3.34; N, 32.69. Found C, 51.85; H, 3.42; N, 32.35.

3-Amino-4-(4-chlorophenyl)-7-thioxo-7,8-dihydro-6*H***-pyrazolo[4,3-***e***][1,2,4]triazolo[1,5***a***] pyridine-5-carbonitrile (6). A mixture of 5 (2.99 g, 0.01 mol) and carbon disulfide (0.60 mL, 0.01 mol) in ethanolic potassium hydroxide (10%, 100 mL) was refluxed for 4 hours, after cooling the reaction mixture was poured onto ice-HCl. The solid obtained was filtered, washed several times with water and crystallized from aqueous DMF to give 6** as yellow crystals, yield 2.49 g (73 %), mp > 300 °C. UV λ_{max} (log ε); 410 (3.1), 355 (4.7), 278 nm (4.4). IR (KBr, cm⁻¹): 3469, 3305, 3187 (2NH, NH₂), 2214 (C=N) 1633 (C=N), 1290 (C=S). Anal. Calcd. for C₁₄H₈ClN₅S (341.78): C, 49.20; H, 2.36; N, 28.69; S, 9.38. Found C, 48.97; H, 2.24; N, 28.75; S, 9.26.

3-Amino-4-(4-chlorophenyl)-7-methylthio-7,8-dihydro-6*H***-pyrazolo**[**4,3***-e*][**1,2,4**]**triazolo** [**1,5***-a*]**pyridine-5-carbonitrile (7).** A mixture of **6** (1.71 g, 0.005 mol) and methyl iodide (0.31 mL, 0.005 mol) ethanolic potassium hydroxide (10%, 50 mL) was refluxed for 6 hours, after cooling the reaction mixture was poured onto ice-HCl. The solid obtained was filtered, washed several times with water and crystallized from aqueous DMF to give **7** as yellow crystals, yield 1.22 g (69 %), mp > 300 °C. IR (KBr, cm⁻¹): 3469, 3317, 3219 (NH₂, NH), 2925 (CH₃), 2216 (C=N), 1623 (C=N) and 1459, 1422 (*def.* CH₃) cm⁻¹. ¹H NMR (δ , DMSO-*d*₆): 3.96 (s, 3H, CH₃), 5.62 (bs, 2H, NH₂ exchangeable with D₂O), 8.21 (d, 2H, Ar-H, J = 8.2 Hz), 8.42 (d, 2H, Ar-H, J = 8.2 Hz), 8.95 ppm (s, 1H, NH exchangeable with D₂O).

Hydrazinolysis of 2, 3, 6 and/or 7. Formation of 4

A mixture of **2**, **3**, **6** and/or **7** (0.005 mol) and hydrazine hydrate (5 mL) refluxed for 16 hours, after cooling the reaction mixture was poured onto ice-AcOH. The solid obtained was filtered and crystallized from DMF to give **4** as yellow crystals, mp > 300 °C. UV λ_{max} (log ε): 347(4.22), 271 nm (2.49). IR (KBr, cm⁻¹): 3469, 3308, 3200 (2NH₂, 2NH), 2209 (C=N), 1628 (C=N) cm⁻¹. ¹H NMR (δ , DMSO- d_6): 5.21 (s, 6H, 2NH and 2NH₂), 7.53 (d, 2H, Ar-H), 7.64 ppm (d, 2H, Ar-H). ¹³C NMR (δ , DMSO- d_6): 92.76 (C₅-CN), 93.47 (C_{3a}), 117.12 (C=N), 129.11, 129.57, 132.58, 134.79 (6C of aryl carbons), 146.69 (C₄), 148.64 (C₃), 154.79 (C_{5a}) and 162.34 ppm (C₇ and C_{9a}). MS (Int.%): 340 (0.63), 313 (1.27), 285 (100), 228 (1.88), 113 (5.34) and 56 (3.16). Anal. Calcd. for C₁₄H₁₀ClN₉ (339.75): C, 49.49; H, 2.96; N, 37.10. Found C, 48.65; H, 2.62; N, 37.35.

3-Amino-4-(4-chlorophenyl)-9-(un)substituted-10-oxo-12hydro-7*H*-pyrazolo[4,3-*e*][1,2,4] triazino[3`,4`-5``,1``][1,2,4]triazolo[2``,3``-*a*]pyridine-5-carbonitriles (8a-c). A mixture of 4 (1.70 g, 0.005 mol) and α -oxoacids such as glyoxalic, pyruvic and *p*-chlorostyrylglyoxalic acids (0.005 mol) in glacial acetic acid (40 mL) was refluxed for 4 hours, after cooling the reaction mixture was poured onto ice. The solid obtained was filtered and crystallized to give

8a-c. Compound **8a** crystallized from DMF as yellow crystals, yield 1.02 g (54 %), mp > 300 °C. IR (KBr, cm⁻¹): 3382, 3199 (NH₂, NH), 2224 (C≡N), 1700 (C=O), 1624 (C=N). MS (Int.%): 284 (3.85), 267 (1.13), 173 (1.35), 149 (7.3), 73 (100), 69 (20.95), 55 (13.62); Anal. Calcd. for C₁₆H₈ClN₉O (377.75): C, 50.87; H, 2.13; N, 33.37. Found C, 50.61; H, 2.03; N, 33.08. Compound **8b** crystallized from DMF as yellow crystals, yield 1.1 g (56 %), mp > 300 °C. IR (KBr) cm⁻¹: 3462, 3315, 3195 (NH₂, NH), 2215 (C≡N), 1702 (C=O), 1625 (C=N) cm⁻¹. ¹H NMR (δ, DMSO-*d*₆): 2.7 (s, 3H, CH₃), 6.09 (s, 2H, NH₂ exchangeable with D₂O), 8.32 (d, 2H, Ar-H), 8.43 (d, 2H, Ar-H), 12.71 ppm (bs, 1H, NH ← OH exchangeable with D₂O). MS (Int.%): 395 (0.26), 335 (8.68), 285 (100), 270 (4.85) and 193 (4.33) Anal. Calcd. for C₁₇H₁₀ClN₉O (391.78): C, 52.12; H, 2.57; N, 32.18. Found C, 51.88; H, 2.40; N, 32.11.; Compound **8c** crystallized from DMF as yellow crystals, yield 1.68 g (62 %), mp > 300 °C. UV λ_{max} (log ε): 349 (2.95), 278 nm (1.92). IR (KBr, cm⁻¹): 3467, 3314, 3144 (NH₂, NH), 2217 (C≡N), 1701 (C=O), 1623 (C=N). Anal. Calcd. for C₂₄H₁₃Cl₂N₉O (514.34): C, 56.09; H, 2.55; N, 24.49. Found C, 56.61; H, 2.83; N, 25.01.

3-Amino-4-(4-chlorophenyl)-10-oxo-12-hydro-7H,8H,9H-pyrazolo[4,3-*e*]**[1,2,4]triazino [3`,4`-5``,1``][1,2,4]triazolo[2``,3``-***a***]pyridine-5-carbonitrile (9).** A mixture of **4** (1.70 g, 0.005 mol) and monochloroacetic acid (0.47 g, 0.005 mol) in DMF (40 mL) was refluxed for 4 hours, after cooling the reaction mixture was poured onto ice. The solid obtained was filtered and crystallized from DMF to give **9** as yellow crystals, yield 1.44 g (76 %), mp > 300 °C. IR (KBr, cm⁻¹): 3329, 3303 (NH₂, NH), 2926 (CH₂), 2212 (C=N), 1639 (C=O), 1555 (C=N).

¹H NMR (δ, DMSO-*d*₆): 2.89 (s, 2H, CH₂), 5.19 (s, 2H, NH₂), 7.4 (d, 2H, 2Ar-H), 7.6 (d, 2H, Ar-H), 11.85 ppm (bs, 2H, 2NH). ¹³C NMR (δ, DMSO-*d*₆): 35.69 (CH₂), 92.21 (C₅-CN), 93.59 (C_{3a}), 116.86 (C=N), 129.19, 129.29, 129.49, 132.11 and 134.89 (6C of aryl carbons), 146.22 (C₄), 147.76 (C_{12a}), 147.78 (C₃ and C_{5a}), 155.58 (C_{6a}) and 161.21 ppm (C₁₀ as C=O). Anal. Calcd. for C₁₆H₁₀ClN₉O (379.77): C, 50.60; H, 2.65; N, 33.19. Found C, 50.27; H, 2.43; N, 32.82.

3-Amino-4-(4-chlorophenyl)-9-oxo-12-hydro-7H,8H,10*H***-pyrazolo[4,3-***e***][1,2,4]triazino [3**`,**4**`-**5**``,**1**``][1,2,4]triazolo[**2**``,**3**``-*a*]pyridine-5-carbonitrile (**10**). A mixture of **4** (1.70 g, 0.005 mol) and chloroacetyl chloride (0.40 mL, 0.005 mol) in DMF (40 mL) was refluxed for 4 hours, after cooling the reaction mixture was poured onto ice. The solid obtained was filtered and crystallized from DMF to give **10** as yellow crystals, yield 1.23 g (65%), mp > 300 °C. IR (KBr, cm⁻¹): 3466, 3310, 3210 (NH₂, 2NH), 2945, 2832 (CH₂), 2209 (C≡N), 1700 (C=O), 1632 (C=N), 1493, 1421 (*def.* CH₂). ¹H NMR (δ , DMSO-*d*₆): 2.75 (s, 2H, CH₂), 5.3 (s, 2H, NH₂), 7.5 (d, 2H, 2Ar-H), 7.9 (d, 2H, 2Ar-H) and 11.90 ppm (bs, 2H, 2NH). ¹³C NMR δ : 35.85 (CH₂), 82 (C₅-CN), 94 (C_{3a}), 117 (C≡N), 129.62, 129.95, 132.39 and 135.28 (6C of aryl carbons), 147 (C₄), 148 (C_{12a}), 156.12 (C₃ and C_{5a}), 161.7 (C_{6a}) and 162.66 ppm (C₁₀ as C=O). Anal. Calcd. for C₁₆H₁₀ClN₉O (379.77): C, 50.60; H, 2.65; N, 33.19. Found C, 50.42; H, 2.31; N, 32.94.

3-Amino-4-(4-chlorophenyl)-9-phenyl-12-hydro-7H,10*H***-pyrazolo[4,3-***e***][1,2,4]triazino [3`,4`-5``,1``][1,2,4] triazolo [2``,3``-***a***] pyridine-5-carbonitrile (11). A mixture of 4 (1.70 g, 0.005 mol) and phenacyl bromide (1 g, 0.005 mol) in ethanolic NaOH (5%, 100 mL) was refluxed for 4 hours, after cooling the reaction mixture was acidified with conc. HCl. The solid obtained was filtered and crystallized from DMF to give 11 as yellow crystals, yield 1.12 g (51 %), mp > 300 °C. IR (KBr, cm⁻¹): 3470, 3315, 3140 (NH₂, NH), 2942, 2831 (CH₂), 2214 (C=N), 1625 (C=N). Anal. Calcd. for C₂₂H₁₄ClN₉ (439.87): C, 60.07; H, 3.21; N, 28.66. Found C, 59.75; H, 3.13; N, 28.28.**

3-Amino-4-(4-chlorophenyl)-12-hydro-7H,8H,9H,10H-pyrazolo[4,3-

e][1,2,4]triazino[3`,4`-5``,1``][1,2,4]triazolo[2``,3``-*a*]pyridine-5-carbonitrile (12). A mixture of **4** (1.70 g, 0.005 mol) and 1,2-dibromoethane (0.43 mL, 0.005 mol) in ethanolic NaOH (5%, 50 mL) was refluxed for 4 hours, after cooling the reaction mixture was acidified with conc. HCl. The solid obtained was filtered and crystallized from DMF to give **12** as yellow crystals, yield 1.04 g (57 %), mp > 300 °C. UV λ_{max} (log ε): 350 (4.50), 275 (4.25) nm. IR (KBr, cm⁻¹): 3468, 3313, 3207 (NH₂, 2NH), 2924, 2840 (CH₂), 2215 (C=N), 1625 (C=N). ¹H NMR (δ , DMSO-*d*₆): 3.3 (s, 4H, 2CH₂), 5.93 (bs, 4H, NH₂ and 2NH exchangeable by D₂O), 8.28 (d, 2H, Ar-H) and 8.53 (d, 2H, Ar-H). Anal. Calcd. for C₁₆H₁₂ClN₉ (365.79): C, 52.49; H, 3.28; N, 34.45. Found C, 52.61; H, 3.43; N, 34.15.

3-Amino-4-(4-chlorophenyl)-9,10-dioxo-12hydro-7H,8H-pyrazolo[4,3-*e*]**[1,2,4]triazino [3`,4`-5``,1``][1,2,4]triazolo[2``,3``-***a*]**pyridine-5-carbonitrile (13).** A mixture of **4** (1.70 g, 0.005 mol) and diethyl oxalate (0.68 mL, 0.005 mol) in DMF (40 mL) was refluxed for 8 hours, after cooling the reaction mixture was poured onto ice. The solid obtained was filtered and crystallized from DMF to give **13** as yellow crystals, yield 1.2 g (61%), mp > 300 °C. (KBr,

cm⁻¹): 3477, 3321, 3147 (NH₂, 2NH), 2217 (C=N), 1711 (C=O), 1629 (C=N). ¹H NMR (δ , DMSO- d_6): 5.24 (s, 2H, NH₂), 7.52-7.73 (m, 4H, Ar-H) and 11.88 (s, 2H, NH and OH of 1,2,4-triazine). ¹³C NMR (δ , DMSO- d_6): 92.15 (C₅-CN), 93.52 (C_{3a}), 116.90 (C=N); 129.19, 129.49, 132.05, 134.87 (6C of aryl carbons), 146.15 (C₄), 146.18 (C_{12a}), 146.22 (C_{5a}), 147.75 (C₃), 155.62 (C₉ and C_{6a}), 161.19 (C₁₀ as C=O).

3-Amino-4-(4-chlorophenyl)-9,10-diphenyl-12-hydro-7H,10H-pyrazolo[4,3-e][1,2,4]

triazino[3`,4`-5``,1``][1,2,4]triazolo[2``,3``-*a*]pyridine-5-carbonitrile (14). A mixture of 4 (1.70 g, 0.005 mol) and benzoin (1.06 g, 0.005 mol) in DMF (40 mL) was refluxed for 8 hours, after cooling the reaction mixture was poured onto ice. The solid obtained was filtered and crystallized from DMF to give 14 as yellow crystals, yield 1.68 g (65 %), mp > 300 °C. IR (KBr, cm⁻¹): 3470, 3315, 3142 (NH₂, NH), 2922, 2850 (CH₂) 2215 (C=N), 1626 (C=N). ¹H NMR (δ , DMSO-*d*₆): 5.5 (s, 2H, NH₂), 7.6-7.9 (m, 15H, Ar-H and CH of 1,2,4-triazin-5-yl) and 11.9 ppm (s, 1H, NH). ¹³C NMR (δ , DMSO-*d*₆): 92 (C₅-CN), 94 (C_{3a}), 117.98 (C=N and C₁₀), 130.14-136.41 (18C of aryl carbons), 147 (C₄ and C_{12a}), 148 (C₃), 156 (C_{5a}), 162 (C_{6a}), 195.69 (C₉). MS (Int.%): 461 (2.99), 285 (100), 270 (7.83), 178 (2.35), 174 (6.17), 160

(4.61). Anal. Calcd. for $C_{28}H_{18}ClN_9$ (515.97): C, 65.12; H, 3.49; N, 24.42. Found C, 64.61; H, 3.43; N, 24.35.

3-Amino-4-(4-chlorophenyl)-15-hydro-7*H***-pyrazolo[4,3-***e***]indolo[2,3-5`,6`][1,2,4]triazino [3`,4`-5``,1``][1,2,4]triazolo[2``,3``-***a***]pyridine-5-carbonitrile (15). A mixture of 4 (1.70 g, 0.005 mol) and isatine (0.74 g, 0.005 mol) in DMF (40 mL) was refluxed for 8 hours, the reaction mixture was cooled and poured onto ice. The solid obtained was filtered and crystallized from DMF to give 15 as yellow crystals, yield 1.06 g (47 %), mp > 300 °C. IR (KBr, cm⁻¹): 3468, 3316, 3179 (NH₂, NH), 2218 (C=N), 1624 (C=N). ¹H NMR (\delta, DMSO-***d***₆): 5.3 (s, 2H, NH₂), 7.5-7.9 (m, 8H, Ar-H) and 11.9 ppm (s, 1H, NH). ¹³C NMR \delta: 92 (C₅-CN), 93 (C_{3a}), 117 (C=N), 129.36-135.29 (10C of aryl carbons), 146 (C₄), 147 (C_{15a}), 152 (C₃), 156 (C_{5a}), 157 (C_{6a}), 161.67 (C_{13a}), 163 ppm (C_{8a}). Anal. Calcd. for C₂₂H₁₁ClN₁₀ (515.97): C, 58.61; H, 2.46; N, 31.07. Found C, 58.82; H, 2.30; N, 30.74.**

3-Amino-9-(2-aminophenyl)-4-(4-chlorophenyl)-10-oxo-12-hydro-7H-pyrazolo[4,3-

e][1,2,4] triazino[3`,4`-5``,1``][1,2,4]triazolo [2``,3``-*a*]pyridine-5-carbonitrile (16). A mixture of 4 (1.70 g, 0.005 mol) and isatine (0.74 g, 0.005 mol) in ethanolic NaOH (5%, 100 mL) was refluxed for 4 hours, the reaction mixture was cooled and acidified with diluted AcOH. The solid so formed was filtered and crystallized from DMF to give 16 as pale brown crystals, yield

55 g (66 %), mp > 300 °C. UV λ_{max} (log ε): 435 (2.6), 352 (4.15), 275 nm (4.2). (KBr, cm⁻¹): 3467, 3309, 3207 (2NH₂, NH), 2210 (C=N), 1726 (C=O), 1628 (C=N). ¹H NMR (δ , DMSOd₆): 3.47 (bs, 2H, NH₂), 5.40 (s, 2H, 2NH), 7.27-7.94 (m, 8H, Ar-H), 12.06 ppm (s, 1H, NH \implies OH of 1,2,4-triazinone). MS (Int.%): 469 (0.99), 322 (1.84), 285 (100), 270 (10.68), 211 (4.24), 146 (12.51), 113 (52.33), 93 (16.40), 67 (28.14). Anal. Calcd. for C₂₂H₁₃ClN₁₀O (468.87): C, 56.36; H, 2.80; N, 29.86. Found C, 56.11; H,3.66; N, 29.84

Acetylation of 15- formation of 17. A mixture of 15 (0.45 g, 0.001 mol) and acetic anydride (10 mL) was refluxed for 2 hours. The solid obtained while hot was filtered, washed with ether and crystallized from AcOH to give 17 as yellow crystals, yield 0.3 g (57 %), mp > 300 °C. IR (KBr, cm⁻¹): 3422 (NH), 2922, 2852 (CH₃), 2217 (C \equiv N), 1720, 1655 (2 C=O), 1620 (C=N). Anal. Calcd. for C₂₆H₁₅ClN₁₀O₂ (534.93): C, 58.38; H, 2.80; N, 26.18. Found C, 57.81; H, 2.56; N, 25.85.

Formation of N-acetyl derivative 18. A mixture of **16** (0.94 g, 0.002 mol) and glacial acetic acid (40 mL) was refluxed for 4 hours, the reaction mixture was cooled and poured onto ice. The solid obtained was filtered and crystallized from AcOH to give **18** as yellow crystals, yield 0.64 g (63 %), m.p. > 300 °C. UV λ_{max} (log ε): 384 (3.4), 349 (4.6), 278 nm (4.8). IR (KBr, cm⁻¹): 3465, 3407, 3314, 3143 (2NH₂, 2NH), 2219 (C=N), 1764 and 1671 (2 C=O), 1620 (C=N), 1493, 1394 (*def.* CH₃). ¹H NMR (δ , DMSO-*d*₆): 1.91 (s, 3H, CH₃), 5.32 (bs, 2H, NH₂), 7.48-7.94 (m, 8H, Ar-H), 9.63 (s, 1H, -NHCO-), 11.88 (bs, 1H, NH $\stackrel{\frown}{\longrightarrow}$ OH). Anal.

Calcd. for $C_{24}H_{15}ClN_{10}O_2$ (510.90): C,56.37; H,2.94; N, 27.40. Found C, 55.87; H, 2.75; N, 26.98

Formation of bi-acetyl derivative 19. A mixture of **16** (0.94 g, 0.002 mol) in glacial acetic acid (50 mL) and acetic anhydride (1 mL) was refluxed for 8 hours, the reaction mixture was cooled and poured onto ice. The solid obtained was filtered and crystallized from acetic acid to give **19** as yellow crystals, yield 0.61 g (58%), mp > 300 °C. IR (KBr, cm⁻¹): 3502, 3397, 33192, (3NH), 2924 (CH₃), 2234 (C≡N), 1740, 1702, 1658 (3 C=O), 1596 (C=N), 1495, 1455 (*def.* CH₃). ¹H NMR (δ, DMSO-*d*₆): 1.56 (s, 3H, CH₃), 1.92 (s, 3H, CH₃), 7.35-7.64 (m, 8H, Ar-H), 9.65

¹H NMR (δ , DMSO-*d*₆): 1.56 (s, 3H, CH₃), 1.92 (s, 3H, CH₃), 7.35-7.64 (m, 8H, Ar-H), 9.65 (bs, 1H, NH) and 13.09 (bs, 2H, 2NH). Anal. Calcd. for C₂₆H₁₇ClN₁₀O₃ (522.94): C, 59.66; H, 3.25; N, 26.77. Found C, 59.42; H, 3.36; N, 26.74.

References

- 1. Elnagdi M. H.; Ghozlan S. A.; Abdel-Razik F. M.; Maghraby A. S. Chem. Synop **1991**, *5*, 116.
- 2. Attaby F. A.; Eldin S. M.; Abdel-Razik F. M. Phosphorus, Sulfur Silicon Relat.Elem. 1995, 21, 106.
- 3. Asadov Kh. A.; Burangulova R. N.; Guseninov F. H.; Gilmanov R.Z.; Phaljachov I. Ph. *Chem. Heterocycl.Compounds* **2003**, *39*, 392.
- 4. Miletin M.; Hartl J.; Dolezal M.; Odlerova Z.; Kralova K.; Machacek M. *Molecules* **2000**, *5*, 208.
- 5. Abdel-Rahman A.; Bakhite E. A.; Al-Laifi E. A. J. Chin. Chem. Soc. 2002, 49, 223.
- 6. Rao C. S.; Venkaleswarlu V.; Achaiah G. Bioorg Med. Chem. Lett. 2006, 16, 2134.
- 7. Todd A.H. Brit. 1 1970, 203, 149; Chem. Abstr. 1970, 73, 120508b.
- 8. Younghale G.A. U.S. 1980, 288, 440; Chem. Abstr. 1982, 96, 6596c
- 9. Li F.; Feng Y.; Meng Q.; Li W.; Wang Q.; Tao F. ARKIVOC 2007, (i), 40.
- 10. Abdel-Rahman R.M., Morsy J.M.; El-Edfawy S.; Ameneand H. A. *Pharmazie* **1999**, *54*, 347.
- 11. Abdel-Rahman R.M.; Morsy J. M.; Hanafy F.; Abdel-Salam H.A. Pharmazie 1999, 54, 3
- 12. El-Gendy Z.; Abdel-Rahman R.M. Indian J. Heterocycl. Chem. 1995, 4, 295.
- 13. Abdel-Rahman, R. M. Trends Heterocycl. Chem. (India) 2002, 8, 187.
- 14. Abdel-Rahman, R. M. Phosphorus, Sulfur, Silicon Relat. Elem. 2000, 166, 315.
- 15. Abdel-Rahman, R. M. Trends Heterocycl. Chem.(India) 1999, 6, 126.
- 16. Abdel-Rahman, R. M. Pharmazie 2001, 56, 275.
- 17. Abdel-Rahman, R. M. Pharmazie 2001, 56, 195.
- 18. Abdel-Rahman, R. M. *Pharmazie* **2001**, 56, 18.
- 19. Abdel-Rahman, R. M. Pharmazie 1999, 54, 791.

- 20. Abdel-Rahman, R. M.; El-Gendy, Z.; Mahmoud, M. B. Indian J. Chem. 1990, 29B, 352.
- 21. Abdel-Rahman, R. M.; Islam, M. I. E.; El-Gendy, Z. J. Indian Chem. Soc. 1991, 68, 621.
- 22. Abdel-Rahman, R. M.; El-Gendy, Z.; Fawzy, M. M. Asian J. Chem. 1992, 4(2), 364.
- 23. Abdel-Rahman, R. M.; Fawzy, M. M.; El-Gendy, Z. Asian J. Chem. 1992, 4(3), 534.
- 24. Abdel-Rahman, R. M.; Seada, M.; El-Gendy, Z.; Islam, I. E.; Mahmoud, M. B. *Farmaco* **1993**, *48*(3), 407.
- 25. Abdel-Rahman, R.M.; Abdel-Halim, A.M.; Ibrahim, S. S.; Mohamed E. A. J. Chem. Soc. (*Pakistan*) **1987**, *9*(4), 523.
- 26. Shawali, A. S.; Gomha, S. M. Tetrahedron 2000, 58, 8559.
- 27. Ismail, M. M. Chem. Pap. 2001, 55(4), 241.
- 28. Eladawy, M. A. Sulfur Lett. 1990, 11, 1, Chem. Abstr. 1990, 118, 115260d.
- 29. Zaher, H. A.; Abdel-Rahman, R. M.; Abdel-Halim, A. M. Indian J. Chem. 1987, 26B, 110.
- 30. Fathy, N. M.; Aly, A. S.; Hassan, N. A.; Abu-Zied, K. M.; Abdel-Fatah, M. *Egypt J. Chem.* **1997**, *40*(2), 117.
- 31. Abdel-Halim, A. M.; El-Gendy, Z.; Abdel-Rahman, R. M. Pharmazie 1995, 50, 726.
- 32. Sztanke, K., Fidecka, S.; Kedzierska, E.; Karczmarzyk, Z.; Pihlaja K.; Malosuk, *Eur. J. Med. Chem.* **2005**, *40*, 127.
- 33. Gould, J. C., Bowie, J. M. Edinb. Med. J. 1952, 59, 198.
- 34. Singh, A.; Latita, R.; Dhakarey, R.; Saxena, G. J. Indian. Chem. Soc. 1996, 73, 339.
- 35. Al-Najjar, A. A. A.; Amer, S. A.; Riad, M.; Elghamy, I.; Elnagdi, M. H. J. Chem. Res.(S) **1996**, 296.