Pyrrole studies part 47.¹¹³C NMR Spectroscopic characterisation of the products of the reaction of formylpyrroles with aniline and diaminobenzenes

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Dedicated to Professor Gurnos Jones on the occasion of his 70th birthday, and in recognition of our long standing friendship

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Abstract

Pyrrole-3,4-dicarboxaldehydes react with aniline to produce "bis-imines" having a 3H-pyrrole structure, 3b. In contrast, pyrrole-2-carboxaldehyde and pyrrole-2,5-dicarboxaldehydes produce 1H-pyrrolylmethylenimines. Benzimidazolyl derivatives, are formed exclusively from the reaction of β -formylpyrroles with 1,2-diaminobenzene, while α -formylpyrroles form 2:1 bisimines or benzimidazoles depending upon the reaction conditions. Pyrrole-2,4-dicarboxaldehydes react with aniline preferentially at the 4-CHO group. With an excess of aniline, the bis-imine is produced.

Keywords: Pyrroledicarboxaldehydes, Benzimidazolyl derivatives, ¹³C NMR

Introduction

In earlier publications we reported the reaction of α - and β -formylpyrroles with aminoalkanes^{2,3} and with α,ω -diaminoalkanes^{3,4} and showed that, in the case of the reaction of 3,4-diformylpyrroles with a,w-diaminoalkanes, either 2 + 2 or 1 + 1 cycloadducts were formed depending upon the chain length of the diaminoalkanes. Additionally, we provided ¹³C NMR evidence that the adducts existed as 3*H*-pyrroles 1b, instead of the expected 1*H*-pyrroles 1a.³ The unexpected stability of the 3*H*-isomers appears to be derived mainly from a strong intramolecular H-bond, as shown in 1b. In contrast, 2- and 3-formyl- and 2,5-diformylpyrroles react with aliphatic amines to produce 1*H*-pyrrolyl-methylenimines,^{2,4} although it has been

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reported that the imine derived from 2-formyl-3-methoxy-5-(pyrrol-2-yl)pyrrole exists as the isomeric 2*H*-pyrrole system 2,⁵ which is presumably stabilised by intramolecular H-bonding.

Reactions of 2,5-dimethylpyrrole-3-carboxaldehyde 2,5-dimethylpyrrole-3,4-dicarboxaldehyde

2,5-Dimethylpyrrole-3-carboxaldehyde reacts with aniline to give the ^{1}H -pyrrol-3-ylmethylene-imine, characterised by the ^{13}C NMR signals for the pyrrole ring at δ 104.5, 119.4, 127.9 and 132.8. The ^{1}H and ^{13}C chemical shifts for the CH=N group at δ 8.27 and 154.7, respectively, differ by ca. $^{+}0.1$ and $^{-}1.1$ ppm, respectively, compared with the corresponding signals for analogous aliphatic imines. 3 In contrast, and analogous to its reaction with aliphatic amines, 2,5-dimethylpyrrole-3,4-dicarboxaldehyde reacts with aniline to produce the "bisimine", which exists predominantly in the ^{3}H -pyrrole form 3b, as evident from the time-averaged ^{13}C NMR chemical shifts for the pyrrole ring carbon atoms at δ 119.5 and 155.8 (cf. ref. 3). The time-averaged ^{1}H and ^{13}C signals for the imino group appear at δ 8.36 and 152.9, respectively.

In contrast with its reaction with 1,2-diaminoethane,³ 2,5-dimethylpyrrole-3,4-dicarboxaldehyde reacted with 1,2-diaminobenzene to yield the 2-(4-formylpyrrol-3-yl)benzimidazole and the bis-benzimidazolyl derivative 4, with no evidence for a macrocyclic 2 \pm 2 imino adduct analogous to 1. The formation of the benzimidazolyl derivatives, obtained from b-formylpyrroles, is clearly confirmed by their ¹³C NMR signals at d 144.7 \pm 3.9.6 It is probable

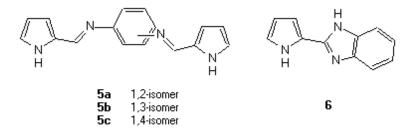
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that methylenimines are initially formed in equilibrium with the 1,2-dihydrobenzimidazolyl derivatives but, as the macrocyclic systems analogous to those formed with diaminoalkanes would be anti-aromatic, their formation is energetically unfavourable and a more favourable oxidation of the dihydrobenzimidazoles leads to the isolated aromatic products. Polymeric imine derivatives are formed in trace quantities (δ_H 8.24, δ_C 153.5).

Reactions of pyrrole-2-carboxaldehyde, pyrrole-2,5-dicarboxaldehyde and 3,4-dimethylpyrrole-2,5-dicarboxaldehyde

All NMR spectral data for the imines obtained from aminobenzenes and pyrrole-2-carboxaldehyde indicates that, as with the corresponding reactions with alkylamines, ^{2,4} the products have the 1*H*-pyrrole structure characterised by the ¹³C NMR signals for the pyrrole ring at δ 109.8 \pm 0.4, 116.7 \pm 0.6, 123.7 \pm 0.1 and 130.7 \pm 0.1. The a-methylenimino group generates ¹H and ¹³C signals at δ 8.26 \pm 0.28 and 149.7 \pm 1.3, respectively.

The reaction of the 2-carboxaldehyde with 1,2-diaminobenzene has been described previously (e.g. refs. 7 - 11), although there is some confusion with the reaction conditions required specifically to produce the thermally unstable bis-imine 5a or the benzimidazole 6; products having the same melting points have been described (without unequivocal spectral evidence) as having different structures. As with the pyrrol-3-yl isomers, the bis-imine and the benzimidazolyl derivatives are distinguished readily by their ¹H and ¹³C NMR spectra (Table 1). The addition of copper salts to the reaction of the aldehyde with 1,2-diaminobenzene does not act, as might be predicted, as a template for the formation of the bis-imine, but provides a favourable redox system for the conversion of the dihydrobenzimidazole, formed in equilibrium from the bis-imine, into 6. As expected, 1,3- and 1,4-diaminobenzenes produced the bis-imines 5b and 5c.



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 compound	observed mp (°C)	literature mp (°C)	1H NMR signal ^a	13C NMR signal ^a
5a	197 - 199	148^{b}	7.70	150.6
6	260 - 261	278 - 280, ^c 273 - 275, ^d 257 - 258, ^e	-	146.3
5b	145 - 147	137^{b}	8.38	150.5
5c	216 - 218	150, ^b 210 - 212f	8.37	149.3

Table 1. Physical data for the products of pyrrole-2-carboxaldehyde with 1,2-, 1,3- and 1,4-diaminobenzene

Pyrrole-2,5-dicarboxaldehydes are converted into the bis-imines 7a and 7b upon reaction with aniline but, as with the 3,4-dicarboxaldehydes, reaction with 1,2-diaminobenzene produced a complex mixture comprised mainly of 8 (δ_C 142.0), together with a polymeric imine and the macrocyclic 2 + 2 imino adduct (δ_H 8.22, δ_C 150.0), analogous to that formed from the reaction of pyrrole-2,5-dicarboxaldehyde with a,w-diaminoalkanes.^{4,13} None of the products could be obtained in sufficiently pure condition for full characterisation.

Reactions of 1-methylpyrrole-2,4-dicarboxaldehyde

Predictably, reaction of the dialdehyde with an excess of aniline produces the bis-imine 12 but, when reacted with one equivalent of aniline, two mono-imines are produced in a *ca.* 2:1 ratio, as indicated by ¹H NMR spectral integration of the CHO and *N*-methyl signals. Comparison of the NMR spectral data (in particular, the ¹³C NMR chemical shifts of the CH=N group) for the two mono-imines with those for analogous mono-imines 10b, 11b and 11c (Table 2) indicate that the major product is the imine 10a, while 11a is the minor product.

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 $[^]a$ N-phenyl (pyrrol-2-yl)methylenimine δ 8.28 and 150.3. b ref. 7. c ref. 8. d refs. 9, 10. e ref. 11. f ref. 12.

Table 2. ¹H and ¹³C NMR signals for imines derived from 2- and 4-pyrrolecarboxaldehydes

		4-CH=N group ^b		
compound	¹ H NMR signal	¹³ C NMR signal	¹ H NMR signal	¹³ C NMR signal
10a	-	-	8.28	153.8
10b	-	-	8.30	153.9
11a	8.28	150.4	-	-
11b	8.23	150.4	-	-
11c	8.17	150.4	-	-
12	8.22	150.7	8.22	152.8

 $[^]a$ N-phenyl (pyrrol-2-yl)methylenimine δ 8.28 and 150.3 b N-phenyl (2,5-dimethylpyrrol-3-yl)methylenimine δ 8.22 and 154.7.

Experimental Section

General Procedures. Unless indicated otherwise, ¹H spectra were measured for CDCl₃ solutions at 60, 100 or 270 MHz using JEOL PMX60SI, JEOL FX-100 JEOL JNM-EX-270 spectrometers and ¹³C NMR spectra were obtained at 67.5 or 100 MHz using JEOL JNM-EX-270 or JEOL JNM-GX-400 spectrometers. All chemical shifts are recorded relative to Me4Si. Infrared spectral measurements were obtained for Nujol mulls using a Perkin-Elmer 577 spectrometer. High resolution mass spectral data were obtained under EI conditions at 70 eV with a AEI MS902 spectrometer.

N-[(2,5-Dimethylpyrrol-3-yl)methylene]-*N*-phenylamine. Aniline (0.37 g, 4 mmol) and 2,5-dimethylpyrrole-3-carboxaldehyde (0.49 g, 4 mmol)in benzene (20 mL) were heated under reflux for 6 h in the presence of A4 molecular sieves and then kept at 15°C for 12 h.. The precipitated product was recrystallised from benzene to give the imine (0.55 g, 69%) as colourless needles, mp 185 - 187°C. Anal. Calcd. for $C_{13}H_{14}N_2$: $C_{13}H_{14}H_{14}H_{15}$: $C_{13}H_{14}H_{15}$: $C_{13}H_{14}H_{15}$: $C_{13}H_{14}H_{15}$: $C_{13}H_{15}H_{15}$: $C_{13}H_{15}$

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C, 78.6; H, 7.0; N, 13.8. δ H 2.21 (s, 5-Me), 2.39 (s, 2-Me), 6.33 (d, 1.5 Hz, pyrrolyl 4-CH), 7.00 - 7.45 (m, 5 x phenyl CH), 8.27 (s, CH=N). dC 11.3 (q, 2-Me) 12.7 (q, 5-Me), 104.5 (d, pyrrolyl 4-C), 119.4 (s, pyrrolyl 3-C), 120.9 (d, phenyl 2-C/6-C), 124.5 (d, phenyl 4-C), 127.9 (s, pyrrolyl 5-C), 128.9 (d, phenyl 3-C/5-C), 132.8 (s, pyrrolyl 2-C), 153.8 (s, phenyl 1-C), 154.7 (d, CH=N). **3-Benzimidazol-2-yl-2,5-dimethylpyrrole.** 2,5-Dimethylpyrrole-3-carboxaldehyde (1.23 g, 10 mmol) was stirred with 1,2-diaminobenzene (1.1 g, 10 mmol) in methanol (20 mL) at 25°C for 18 h. The solvent was removed under reduced pressure an the red residue was purified by centrifugally accelerated chromatography on Kiesegel using a Chromatatron® Model 7924T with dichloro-methane: ethyl acetate (3:1) as the eluent to give 3-benzimidazol-2-yl-2,5-dimethylpyrrole (0.18 g, 8.5%), mp 200 - 204°C. Anal. Calcd for $C_{13}H_{13}N_3$ C, 73.9; H. 6.2; N, 19.9. Found: C, 73.7; H, 6.0; N, 19.5. $\delta_{\rm H}$ 2.18 (s, 2-Me), 2.45 (s, 5-Me), 6.18 (d, 1.5 Hz, pyrrolyl 4-CH), 6.65 - 6.73 (m, aryl CH), 7.47 (m, aryl CH). $\delta_{\rm C}$ 12.7 (q, 2-Me), 13.1 (q, 5-Me), 105.6 (d, pyrrolyl 4-C), 111.1 (d, benzimidazolyl 4-C/7-C), 114.7 (s, pyrrolyl 3-C), 121.6 (s, pyrrolyl 5-C), 121.9 (d, benzimidazolyl 5-C/6-C), 126.8 (s, pyrrolyl 2-C), 129.1 (s, benzimidazolyl 3a-C/7a-C), 144.2 (s, benzimidazolyl 2-C).

1,3-Dimethyl-4-(phenyliminomethyl)-6-(phenylamino)-2-azafulvene (**3b).** Aniline (0.74 g, 8 mol) and 2,5-dimethylpyrrole-3,4-dicarboxaldehyde (0.6 g, 4 mmol) and in benzene (25 mL) were heated under reflux for 2 h in the presence of A4 molecular sieves. The reaction mixture was cooled to room temperature, filtered, and evaporated under reduced pressure. The crude product was recrystallised from CCl4:petroleum ether (3:1) to yield 3b (0.86 g, 71%) as red crystals, mp 125°C. Anal. Calcd. for $C_{20}H_{19}N_3$: C, 79.7; H, 6.35; N, 13.9. Found: C, 79.5; H, 6.5; N, 13.8. δ_H 2.52 (s, 3-Me/4-Me), 6.97 - 7.50 (m, 10 x phenyl CH), 8.36 (br s, CH=N/=CH-NH). δ_C 11.1 (q, 3-Me/4-Me), 119.5 (d, pyrrolyl 2-C/5-C), 120.8 (d, phenyl 2-C/6-C), 125.5 (d, phenyl 3-C/5-C), 129.7 (d, phenyl 4-C), 148.7 (s, phenyl 1-C), 152.9 (d, CH=N/CH-NH), 155.8 (s, pyrrolyl 3-C/4-C).

3-Benzimidazol-2-yl-2,5-dimethylpyrrole-4-carboxaldehyde and **3,4-bis-(benzimidazol-2-yl)-2,5-dimethylpyrrole** (**4**). 2,5-Dimethylpyrrole-3,4-dicarboxaldehyde(1.51 g, 10 mmol) was stirred with 1,2-diaminobenzene (1.08 g, 10 mmol) in MeOH (20 mL) at 25°C for 24 h. The solvent was removed under reduced pressure an the red residue was purified by centrifugally accelerated chromatography on Kiesegel using a Chromatatron® Model 7924T with methanol as the eluent to give 3-benzimidazol-2-yl-2,5-dimethylpyrrole-4-carboxaldehyde (0.08 g, 3%), mp 244 - 247°C. Anal. Calcd. for $C_{14}H_{13}N_3O$: C, 70.3 H, 5.5; N, 17.6. Found: C, 70.1; H, 5.4; N, 17.5. δ_H 2.54 (s, 2-Me), 2.67 (s, 5-Me), 7.40 (m, aryl CH), 7.56 (m, aryl CH), 9.86 (s, CHO). δ_C 11.0 (q, 2-Me), 12.9 (q, 5-Me), 108.5 (s, pyrrolyl 3-C), 117.8 (d, benzimidazolyl 4-C/7-C), 121.2 (d, benzimidazolyl 5-C/6-C), 130.0 (s, benzimidazolyl 3a-C/7a-C), 131.9 (s, pyrrolyl 2-C), 142.7 (s pyrrolyl 5-C), 148.8 (s, benzimidazolyl 2-C), 185.7 (d, CHO) and 4 (0.98 g, 30%) mp 270°C (decomp). Anal. Calcd. for $C_{20}H_{17}N_5$: C, 73.4; H, 5.2; N, 21.4, m/z 327.1484. Found: C, 70.1; H, 5.4; N, 17.5, 327.1489. δ_H 2.75 (s, 2-Me/5-Me), 7.67 (dd, 3.3 Hz, 6.4 Hz), 7.99 (dd, 3.3 Hz, 6.4

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Hz). δ_C 12.0 (q, 2-Me,/5-Me), 110.0 (d, benzimidazolyl 4-C/7-C), 128.2 (s 2 x pyrrolyl 2-C/5-C), 128.3 (d, 2 x benzimidazolyl 5-C/6-C), 128.8 (s, 2 x benzimidazolyl 3a-C/7a-C), 128.8 (d, 2 x pyrrolyl 3-C/4-C), 141.1 (s, 2 x benzimidazolyl 2-C).

N-**Pyrrol-2-ylmethylene-***N*-**phenylamine.** Pyrrole-2-carboxaldehyde (1.90 g, 20 mmol) was converted into the methylenimine (2.6 g, 77%), mp 93 - 94.5°C (lit., pp 93 - 94°C), using the procedure described in the literature. 2 δ_H (DMSO- 2 d6) 6.22 (t, 3.5 Hz, pyrrolyl 4-CH), 6.68 (dd, 3.5, 2.0 Hz, pyrrolyl 3-CH), 6.95 - 7.10 (m, pyrrolyl 5-CH), 7.11 - 7.41 (m, 5 x phenyl CH), 8.27 (s, CH=N). δ_C (DMSO- 2 d6) 109.5 (d, pyrrolyl 4-C), 116.1 (d, pyrrolyl 3-C), 120.6 (d, phenyl 2-C/6-C), 123.6 (d, pyrrolyl 5-C), 124.8 (d phenyl 4-C), 129.0 (d, phenyl 3-C/5-C), 130.6 (s, pyrrolyl 2-C), 150.3 (d, CH=N), 152.1 (s, phenyl 1-C).

Reaction of pyrrole-2-carboxaldehyde with 1,2-diaminobenzene

- (a) 1,2-Diaminobenzene (1.08 g, 10 mmol) and pyrrole-2-carboxaldehyde (1.90 g, 20 mmol) in MeOH (45 mL) were stirred at 15°C for 96 h. The precipitated brown solid was collected, recrystallised from EtOH, and further purified by preparative TLC from silica (Rf 0.54) to give 2-(pyrrol-2-yl)benzimidazole 6 (0.07 g, 3.8%), mp 260 261°C (lit. $^{8-11}$, mp 257 258°C; 274 275°C; 278 280.5°C). Anal. Calcd. for $C_{11}H_9N_3$: C, 72.1; H, 4.95; N, 22.9. Found: C, 71.9; H, 5.1; N, 22.7. δ_H 6.35 6.45 (m, pyrrolyl 4-H), 7.07 7.20 (m, pyrrolyl 3-H), 7.40 7.52 (m, pyrrolyl 5-H), 7.60 7.72 (m, benzimidazolyl 4-H/5-H/6-H/7H). δ_C 110.1 (d, pyrrolyl 4-C), 110.4 (d, pyrrolyl 3-C), 114.3 (d, benzimidazolyl 4-C/7-C), 121.4 (s, pyrrolyl 2-C), 122.5 (d, benzimidazolyl 5-C/6-C), 123.1 (d, pyrrolyl 5-C), 137.6 (s, benzimidazolyl 3a-C/7a-C), 146.3 (s, benzimidazolyl 2-C).
- (b) 1,2-Diaminobenzene (2.16 g, 20 mmol) and pyrrole-2-carboxaldehyde (1.90 g, 20 mmol) in *iso*-PrOH (40 mL) were added to (MeCO₂)₂Cu.H₂O (8.0 g, 40 mmol) in H₂O (100 mL). The reaction mixture was heated at 80 90°C for 2 h and then cooled to 0°C. The precipitated Cu(II) complex was collected and suspended in *iso*-PrOH through which H₂S was passed for 1 h. The precipitated CuS was removed and H₂O added to the filtrate to produce a yellow solid, which was recrystallised from EtOH to give 6 (1.88 g, 51%), mp 260 262°C.
- (c) 1,2-Diaminobenzene (1.4 g, 13 mmol) and pyrrole-2-carboxaldehyde (2.47 g, 26 mmol) in MeOH (60 mL) were stirred heated under reflux for 10 h. The reaction mixture was cooled to 15°C and the precipitated solid was purified by preparative TLC from silica (Rf 0.71) to give N,N'-bis(pyrrol-2-ylmethylene)-1,2-diaminobenzene 5a (3.1 g, 91%), mp 197 199°C (decomp.) (lit.⁷, mp 148°C). Anal. Calcd. for $C_{16}H_{14}N_4$: C, 73.3; H, 5.4; N, 21.4. Found: C, 72.9; H, 5.3; N, 21.4. δ_H 6.01 (dd, 3.5, 2.2 Hz, 2 x pyrrolyl 4-H), 6.28 6.40 (m, 2 x pyrrolyl 3-H), 6.41 (dd, 3.5, 1.5 Hz, 2 x pyrrolyl 5-H), 7.32 7.40 (m, 4 x aryl CH), 7.70 (s, CH=N). δ_C 109.6 (d, pyrrolyl 4-C), 117.2 (d, pyrrolyl 3-C), 118.9 (d, aryl 3-C/6-C), 123.8 (d, pyrrolyl 5-C), 126.6 (d, aryl 4-C/5-C), 130.9 (s, pyrrolyl 2-C), 145.7 (s, aryl 1-C/2-C), 150.6 (d, CH=N).

Reaction of pyrrole-2-carboxaldehyde with 1,3- and 1,4-diaminobenzene. Pyrrole-2-carboxaldehyde (0.95 g, 10 mmol) and the appropriate diaminobenzene (0.54 g, 5 mmol) in

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MeOH (15 mL) were stirred at 15°C for 12 h. The precipitated solid was collected and recrystallised from EtOH.

N,*N*′-Bis(pyrrol-2-ylmethylene)-1,3-diaminobenzene (5b) (82%) had mp 145 - 147°C (lit.⁷, mp 137°C). Anal. Calcd for $C_{16}H_{14}N_4$: C, 73.3; H, 5.4; N, 21.4. Found: C, 73.3, H, 5.5, N, 21.4. δ_H 6.20 - 6.28 (m, 2 x pyrrolyl 4-CH), 6.69 - 6.78 (m, 2 x pyrrolyl 3-CH), 6.94 - 7.05 (m, 2 x pyrrolyl 5-CH), 7.10 - 7.66 (m, aryl 4-CH/5-CH/6-CH), 7.33 (s, 1H, aryl 2-CH), 8.38 (s, 2 x CH=N). δ_C 110.4 (d, pyrrolyl 4-C), 113.4 (d, aryl 2-C), 117.2 (d, pyrrolyl 3-C), 118.2 (d, aryl 4-C/6-C), 123.7 (d, pyrrolyl 5-C), 129.9 (d, aryl 5-C), 130.6 (s, pyrrolyl 2- C), 150.4 (d, CH=N), 152.8 (s, aryl 1-C/3-C).

N,*N*′-**Bis**(**pyrrol-2-ylmethylene**)-**1**,**4-diaminobenzene** (**5c**) (85%) had mp 216 - 218°C (lit., 7,12 mp 150°C, 210 - 212°C). Anal. Calcd for C₁₆H₁₄N₄: C, 73.3; H, 5.4; N, 21.4. Found: C, 73.1, H, 5.4, N, 21.3. δ_H (DMSO-*d*6) 6.20 - 6.28 (m, 2 x pyrrolyl 4-CH), 6.69 - 6.79 (m, 2 x pyrrolyl 3-CH), 7.00 - 7.12 (m, 2 x pyrrolyl 5-CH), 7.25 (s, 4 x aryl CH), 8.37 (s, 2 x CH=N). δ_C (DMSO-*d*6) 109.7 (d, pyrrolyl 4-C), 116.2 (d, pyrrolyl 3-C), 121.5 (d, aryl 2-C/3-C/5-C/6-C), 123.6 (d, pyrrolyl 5-C), 130.7 (s, pyrrolyl 2-C), 149.1 (s, aryl 1-C/4-C), 149.3 (d, CH=N).

Reaction of pyrrole-2,5-dicarboxaldehydes with aniline. Aniline (0.5 g, 5.4 mmol) and the appropriate pyrrole-2,5-dicarboxaldehyde (2.7 mmol) were heated under reflux in benzene (10 mL) in the presence of A4 molecular sieves for 2 h and then allowed to stand at 15°C for 24 h. The precipitated bis-imine was collected, separated from the molecular sieves, and recrystallised from EtOH.

Pyrrole-2,5-dicarboxaldehyde gave 7a. (75%) mp 172 - 173°C. Anal. Calcd for C₁₈H₁₅N₃: C, 79.1; H, 5.5; N, 15.4. Found: C, 78.9; H, 5.5; N, 15.3. dH (DMSO-*d*6) 6.89 (s, pyrrolyl 3-CH/4-CH), 7.05 - 7.49 (m, 10 x phenyl CH), 8.47 (s, CH=N). dC (DMSO-*d*6) 114.8 (d, pyrrolyl 3-C/4-C), 120.6 (d, phenyl 2-C/6-C), 125.4 (d, phenyl 4-C), 129.1 (d, phenyl 3-C/5-C), 134.2 (s, pyrrolyl 2-C/5-C), 150.5 (d, CH=N), 151.5 (s, phenyl 1-C).

3,4-Dimethylpyrrole-2,5-dicarboxaldehyde gave 7b. (80%), mp 154 -155°C. Anal. Calcd for $C_{20}H_{19}N_3$: C, 79.7; H, 6.35; N, 13.9. Found: C, 79.4; H, 6.5; N, 13.9. δ_H 2.20 (s, 3-Me/4-Me), 6.95 - 7.35 (m, 10 x phenyl CH), 8.36 (s, 2 x CH=N). d_C 8.6 (q, Me), 120.9 (d, phenyl 2-C/6-C), 125.6 (d, phenyl 4-C), 125.9 (s, pyrrolyl 3-C/4-C), 129.1 (d, phenyl 3-C/5-C), 130.3 (s, pyrrolyl 2-C/5-C), 147.2 (d, CH=N), 151.9 (s, phenyl 1-C).

Reaction of pyrrole-2,5-dicarboxaldehyde with 1,2-diaminobenzene. 1,2-Diaminobenzene (0.82 g, 7.6 mmol) and pyrrole-2,5-dicarboxaldehyde (0.46 g, 3.8 mmol) in EtOH (30 mL) were heated under reflux for 3 h and then cooled to 0°C. H_2O (50 mL) was added to precipitate an inseparable mixture of products (1.2 g) comprising of 8, as the major product [δ_H 6.60 (s, pyrrolyl 3-CH/4-CH), 6.90 - 7.04 (m, aryl CH), 6.55 - 6.73 (m, aryl CH). δ_C 115.4 (d, pyrrolyl 3-C/4-C), 116.6 (d, benzimidazolyl 4-C/7-C), 120.1 (benzimidazolyl 5-C/6-C), 122.6 (s, pyrrolyl 2-C/5-C), 136.8 (s, benzimidazolyl 3a-C/7a-C), 142.0 (s, benzimidazolyl 2-C)] and imino derivatives [*inter alia* dH 8.22 and dC 150.0 (d, CH=N) and 146.9 (d, aryl 1-C/2-C)].

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N-[(4-Acetyl-1-methylpyrrol-2-yl)methylene]-N-phenylamine (11b). Aniline (0.47 g, 5 mmol) was added to a stirred solution of 4-acetyl-1-methylpyrrole-2-carboxaldehyde (0.76 g, 5 mmol) in benzene (20 mL) and the mixture was heated under reflux for 2 h in the presence of 4A molecular sieves. The mixture was filtered, dried (MgSO₄), and evaporated. The residue was subjected to chromatography on silica, using petroleum ether:ethyl acetate (1:1) as the eluent, and then distilled to give 11b (0.98 g, 87%) as a viscous oil, which did not solidify. Anal: Calcd for $C_{14}H_{14}N_2O$: C, 74.3; H, 6.2; N, 12.4. Found: C, 73.9; H, 6.3; N, 12.5. δ_H 2.40 (s, COMe), 4.06 (s, NMe), 6.98 (d, 3 Hz, pyrrolyl 3-CH), 7.07 - 7.39 (m, 5 x phenyl-CH + pyrrolyl 5-CH), 8.24 (s, CH=N). $\delta_{\rm C}$ 27.0 (q, COMe), 37.8 (q, NMe), 118.4 (d, pyrrolyl 3-C), 120.7 (d, phenyl 2-C/6-C), 125.5 (s, pyrrolyl 4-C), 125.7 (d, phenyl 4-C), 129.2 (d, phenyl 3-C/5-C), 131.5 (s, pyrrolyl 2-C), 132.2 (d, pyrrolyl 5-C), 150.4 (d, CH=N), 152.1 (s, phenyl 1-C), 192.4 (s, C=O). N-[(4-Methoxycarbonyl-1-methylpyrrol-2-yl)methylene]-N-phenylamine (11c). Using a procedure analogous to that described for the preparation of 11b, methyl 2-formyl-1methylpyrrole-4-carboxylate gave 11c (78%), mp 44 - 47°C. Anal. Calcd for C₁₄H₁₄N₂O₂: C, 69.4; H, 5.8; N, 11.6. Found: C, 69.0 H, 5.8; N, 11.3. δ_H 3.78 (s, NMe), 4.01 (s, OMe), 7.08 (d, 3 Hz, pyrrolyl 3-CH), 7.12 - 7.45 (m, 5 x phenyl CH + pyrrolyl 5-CH), 8.23 (s, CH=N). δ_C 37.7 (g, NMe), 51.1 (q, OMe), 115.6 (s, pyrrolyl 4-C), 119.2 (d, pyrrolyl 3-C), 120.6 (d, phenyl 2-C/6-C), 125.5 (d, phenyl 4-C), 129.0 (d, phenyl 3-C/5-C), 131.0 (s, pyrrolyl 2-C), 132.7 (d, pyrrolyl 5-C),

N-[(5-Ethoxycarbonyl-1-methylpyrrol-3-yl)methylene]-N-phenylamine (10b). Using a procedure analogous to that described for the preparation of 11b, ethyl 4-formyl-1-methylpyrrole-2-carboxylate gave 10b (77%), mp 55 - 57°C. Anal. Calcd for C₁₅H₁₆N2O₂: C, 70.3; H, 6.3; N, 10.9. Found: C, 70.3; H, 6.4; N, 10.9. $\delta_{\rm H}$ 1.36 (t, 7.1 Hz, MeCH2), 3.98 (s, NMe), 4.29 (q, 7.1 Hz, MeCH2), 7.01 - 7.46 (m, 5 x phenyl + 2 x pyrrolyl CH), 8.29 (s, CH=N). $\delta_{\rm C}$ 14.4 (q, MeCH2), 37.3 (q, NMe), 60.1 (t, MeCH2), 117.1 (d, pyrrolyl 4-C), 120.8 (d, phenyl 2-C/6-C), 122.2 (s, pyrrolyl 5-C), 124.3 (s, pyrrolyl 3-C), 125.3 (d, phenyl 4-C), 129.0 (d, phenyl 3-C/5-C), 131.0 (d pyrrolyl 2-C), 152.6 (s, phenyl 1-C), 153.9 (d, CH=N), 161.0 (s, C=O).

Reaction of 1-methylpyrrole-2,4-dicarboxaldehyde with aniline

150.4 (d, CH=N), 152.0 (s, phenyl 1-C), 164.3 (s, C=O).

(a)Aniline (0.56 g, 6 mmol) in dry acetonitrile (15 mL) was added to 1-methylpyrrole-2,4-dicarboxaldehyde (0.82 g, 6 mmol) in dry acetonitrile (10 mL) in the presence of 4A molecular sieves. The mixture was heated under reflux for 10 h, filtered, dried (MgSO₄), and evaporated to give a 2:1 mixture of N-[(5-formyl-1-methylpyrrol-3-yl)methylene]-N-phenylamine 10a and N-[(4-formyl-1-methylpyrrol-2-yl)methylene]-N-phenyl-amine 11a, which could not be separated effectively by chromatography. Anal: Calcd for $C_{13}H_{12}NO$: C, 73.6, H, 5.7, N, 13.2. Found: C, 73.9; H, 5.6; N, 13.5. 10a δ_H 3.92 (s, NMe), 6.94 - 7.55 (5 x phenyl + 2 x pyrrolyl CH), 8.26 (s, CH=N), 9.57 (d, 1.5 Hz, CHO); δ_C 37.6 (q, NMe), 117.4 (d, pyrrolyl 4-C), 120.7 (d, phenyl 2-C/6-C), 122.7 (s, pyrrolyl 5-C), 125.2 (d, phenyl 4-C), 125.5 (d, pyrrolyl 2-C), 129.1 (d, phenyl 3-C/5-C), 131.5 (s, pyrrolyl 3-C), 152.6 (s, phenyl 1-C), 153.8 (d, CH=N), 180.2 (d, CHO). 11a

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 δ_{H} 4.04 (s, NMe), 6.94 - 7.55 (5 x phenyl + 2 x pyrrolyl CH), 8.26 (s, CH=N), 9.70 (s, CHO); δ_{C} 38.1 (q, NMe), 117.5 (d, pyrrolyl 3-C), 120.7 (d, phenyl 2-C/6-C), 123.6 (s, pyrrolyl 4-C), 125.5 (d, phenyl 4-C), 129.1(d, phenyl 3-C/5-C), 133.0 (s, pyrrolyl 2-C), 134.4 (d, pyrrolyl 5-C), 150.4 (d, CH=N), 152.6 (s, phenyl 1-C), 184.5 (d, CHO).

(b) Aniline (1.12 g, 12 mmol) in dry acetonitrile (15 mL) was added to 1-methylpyrrole-2,4-dicarboxaldehyde (0.82 g, 6 mmol) in dry acetonitrile (10 mL) in the presence of 4A molecular sieves. The mixture was heated under reflux for 24 h, filtered, dried (MgSO₄), and evaporated. Recrystallisation of the crude product from petroleum ether gave 1-methyl-2,4-bis(phenyl-iminomethyl)pyrrole 12 (1.4 g, 81%). Mp 68 - 71°C. Anal: Calcd for $C_{19}H_{17}N_3$: C, 79.4; H, 6.0; N, 14.6. Found: C, 79.2; H, 6.1; N, 14.7. δ_H 4.05 (s, NMe), 7.04 (s, pyrrolyl 3-CH), 7.08 - 7.44 (m, 10 x phenyl CH + pyrrolyl 5-CH), 8.24 (s, 2 x CH=N). δ_C 37.7 (q, NMe), 117.6 (d, pyrrolyl 3-C), 120.8 (d, phenyl 2-C/6-C), 123.2 (d, pyrrolyl 5-C), 125.2 (d, phenyl 4-C), 125.3 (s, pyrrolyl 2-C), 129.1 (s, pyrrolyl 4-C), 131.5 (d, pyrrolyl 5-C), 131.7 (s, pyrrolyl 2-C), 150.7 (d, 2-CH=N), 152.8 (d, 4-CH=N).

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References

- 1. Part 46. Domingo, L.R.; Jones, R.A.; Picher, M.T.; Sepúlveda, J.A. *J. Mol. Struct.* (*Theochem*) **1996**, 362, 209.
- 2. Jones, R.A. Aust. J. Chem. **1964**, 17, 894.
- 3. Taheri, S.A.N.; Jones, R.A.; Badesha, S.S.; Hania, M.M. *Tetrahedron* **1989**, 45, 7717.
- 4. Taheri, S.A.N. PhD Thesis, University of East Anglia, UK, 1983; Quintanilla-Lopez, G.; Jones, R.A.; Taheri, S.A.N.; unpublished results.
- 5. Lindquist, N.; Fenical, W. *Experientia* **1991**, *47*, 504.
- 6. Complete analysis of the 13C NMR spectra of the benzimidazolyl rings is complicated by the time averaging of the signals for the 3a-C and 7a-C, 4-C and 7-C, and the 5-C and 6-C signals as a result of rapid tautomeric exchange of the NH proton. Where exchange was rapid only three signals were observed whereas, when exchange was slow, additional signals were observed. Only the singlet signal for the 2-C atom remains unaffected by the exchange.

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- 7. Dezelic, M.; Dolibic, G. Bull. soc. chim. rép. populaire Bosnie et Herzégovine 1957, 6, 11; Chem. Abstr. 1958, 52, 10054.
- 8. De Selms, R.C. J. Org. Chem. 1962, 27, 2163.
- 9. Merck & Co. Inc., Brit. pat. 1964, 966796; Chem. Abstr. 1965, 62, 2779.
- 10. George, B.; Papadopoulos, P., J. Org. Chem. 1977, 42, 441.
- 11. El'chaninov, M.M.; Simonov, A.M.; Olienikova, L.Ya. *Khim. Geterotsikl. Soedin.* **1980**, *1*, 71; *Chem. Abstr.* **1980**, *92*, 215346.
- 12. Pfeiffer, P.; Hesse, T.; Pfitzner, H.; Scholl, W.; Thielert, H. *J. prakt. Chem.* **1937**, *149*, 217; *Chem. Abstr.* **1938**, *32*, 512.
- 13. Fenton, D.F.; Moody, R. J. Chem. Soc. Dalton Trans. 1987, 219.

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