

Catalytic asymmetric aziridination of imines

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Dedicated to Professor Madeleine M. Joullie on her 80th birthday

Abstract

Introduction of bulky arene substituents into the 3- and 3'-positions of binaphthol boronates led to a significant improvement of chiral induction in the aziridination of benzylidene benzhydramines.

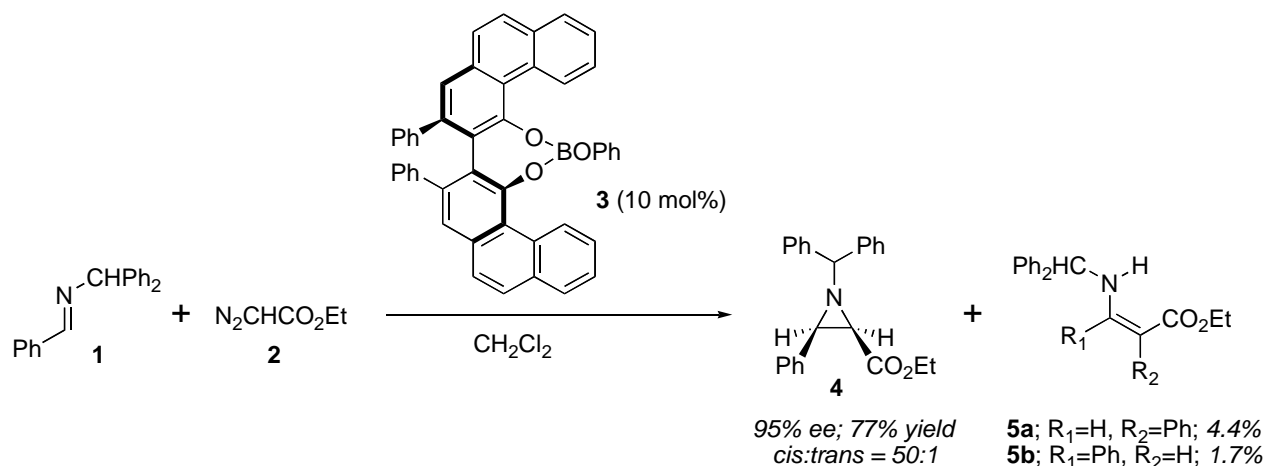
Keywords: Aziridines, imine additions, binaphthol ligands, boron Lewis acids, ethyl diazoacetate

Introduction

Aziridines are important intermediates in organic synthesis. S_N2 ring opening reactions, for example, result in substituted amines, diamines, amino alcohols, and α- or β-amino acids.¹ Aziridines are also important heterocyclic substructures present in a number of interesting biologically active natural products,² as well as therapeutic agents.³

Several methods are available for the preparation of enantiomerically enriched aziridines. However, most involve multi-step procedures, employing either optically active starting materials or stoichiometric quantities of a chiral auxiliary.^{1,4} Direct approaches to aziridines involve the cycloaddition of a nitrene fragment to an olefin, or a carbene to an imine. The latter approach has been studied by a number of groups. Most noteworthy are efforts by Evans,⁵ Jacobsen,⁶ and Scott⁷ towards the development of an efficient catalytic asymmetric version of a nitrene transfer from *N*-(*p*-toluenesulfonyl)imino]phenyliodinane (PhI=NTs) with a variety of chiral copper Lewis acids. Katsuki and Nishikori also showed that manganese salens were excellent catalytic promoters of the enantioselective aziridination of a series of styrene derivatives in very high yields and enantioselectivities.⁸ Aggarwal reported the catalytic asymmetric aziridination of imines with *in situ* generated diazo compounds as carbene sources in the presence of chiral sulfides.⁹ The contributions of Wulff and co-workers have been among the most successful to date. The aziridination of a range of benzhydryl imines with ethyl

diazoacetate, catalyzed by chiral boronates **3**, gave the optically active heterocycles **4** in good yields and excellent enantio- and diastereoselectivities (Scheme 1).¹⁰ The chiral boronates were derived from the vaulted biaryls VANOL and VAPOL and either $\text{BH}_3 \cdot \text{TfH}$ complex or, more efficiently, triphenylborate. In contrast, catalysts derived from linear biaryls such as BINOL and BANOL gave very poor to moderate induction.^{10,11}



Scheme 1. Catalytic asymmetric imine aziridination with the vaulted biaryl complex **3**.¹⁰

Results and Discussion

In the course of our total synthesis of (+)-diepoxin σ ,¹² we observed good to excellent chiral induction in the asymmetric Diels-Alder reaction between juglone and cyclopentadiene mediated by a boron-binaphthol complex.¹³ This and related reactions were further studied with a series of 3,3'-disubstituted binaphthol ligands, such as **6** and **7** (Figure 1). While binaphthol itself gave poor induction in the chiral aziridine formation, in agreement with the Wulff results,^{10,11} we hypothesized that increasing the steric bulk at the 3,3'-positions would exert improved facial control in carbene additions to imines.¹⁴ Accordingly, we examined whether the 3,3'-arylated binaphthols **6** - **8** or the steroidal binaphthols **9** - **11**^{15,16} would demonstrate increased enantioselectivities in imine aziridinations.

Chiral boronate catalysts derived from this set of six ligands **6** - **11** were formed *in situ* and used to screen the aziridination of *N*-(4-bromobenzylidene)benzhydramine **12** with ethyl diazoacetate (Scheme 2). The results from the screening of the catalysts are presented in Table 1. Generally, the reactions showed excellent diastereoselection for the *cis*-isomer of the aziridine **13**.^{10a} In addition, there was no evidence of side products such as the enamino ester **5** in the ¹H NMR spectra of the crude reaction mixtures. The catalysts derived from steroidal-type binaphthol ligands **9** - **11**, while providing aziridine **13** in good yields, gave levels of asymmetric induction similar to those observed with the parent binaphthol (entries 1 - 4).^{10,11} However, as expected, increasing the steric effect by introducing bulky substituents in the 3- and 3'-positions

of the binaphthol scaffold resulted in an increase in the enantioselectivity of the addition. Thus, binaphthalene-substituted ligand **8** provided aziridine **13** in 58% *ee* and 63% yield (Table 1, entry 5). Ligand **6** provided product in 69% *ee* (Table 1, entry 6). The highest levels of asymmetric induction were observed with biaryl ligand **7**, which had also yielded the best induction in our asymmetric Diels-Alder process.¹² Initially, reactions were conducted using methylene chloride (DCM) as solvent, resulting in moderate yields and enantiomeric excess (Table 1, entries 7 and 8). In an effort to improve the yield and enantiomeric excess of the aziridine, we evaluated the effect of variations of solvent, temperature and reaction time. A mixture of methylene chloride and toluene (1:1) led to a significant improvement in both the yield and the enantioselectivity of the process (Table 1, entry 9). The use of (trifluoromethyl)benzene in place of toluene produced the same level of asymmetric induction, but a marked reduction in product yield (Table 1, entry 10). In a mixture of toluene and diethyl ether, aziridine **13** was formed in 80% *ee* and 76% yield after 5 h (Table 1, entry 11). Pure toluene, as well as lower reaction temperatures (0 °C, 3 h, followed by r.t., 15 h) led to a reduction in the %*ee* (Table 1, entries 12 and 13, respectively). Reactions catalyzed by the chiral boronate derived from ligand **7** in toluene at room temperature, in contrast, gave good yields, and preserved high levels of enantioselectivity (78% *ee*) for aziridine **13** (Table 1, entry 14).

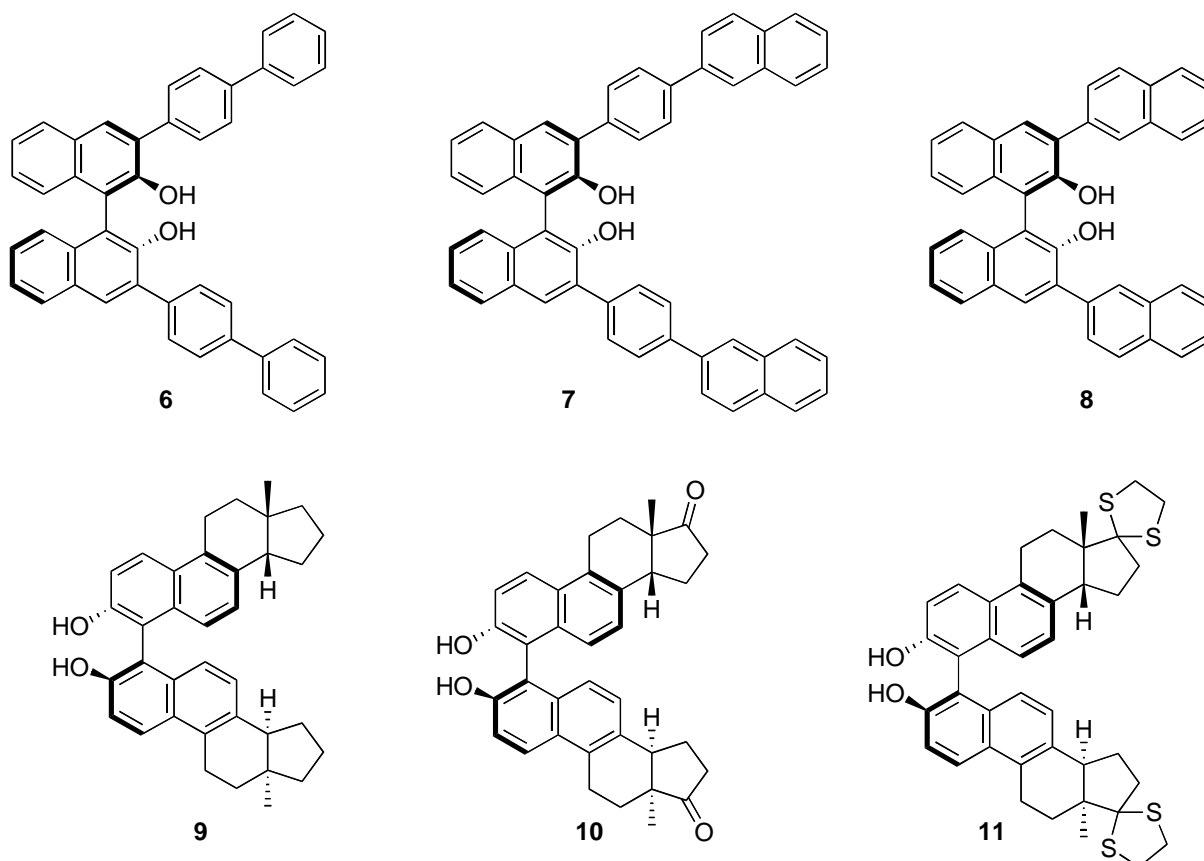
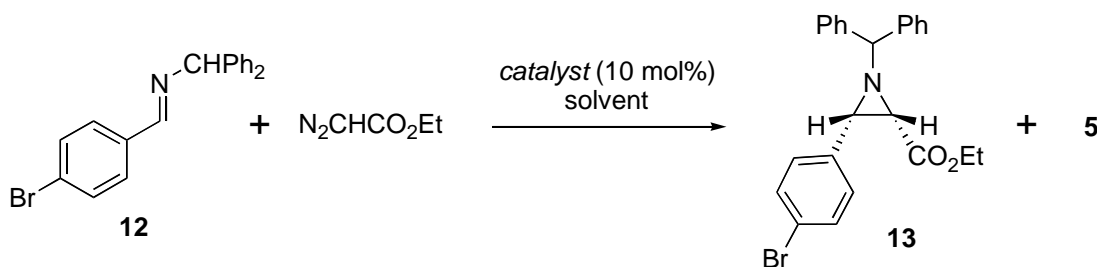


Figure 1. Chiral binaphthol ligands.



Scheme 2. Catalytic asymmetric imine aziridination with substituted binaphthyl ligands.

Table 1. Catalytic asymmetric imine aziridination with substituted binaphthyl ligands¹⁷

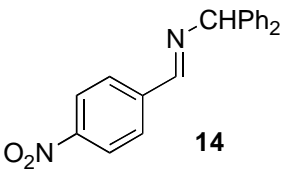
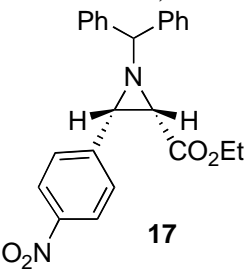
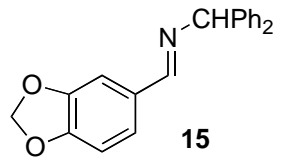
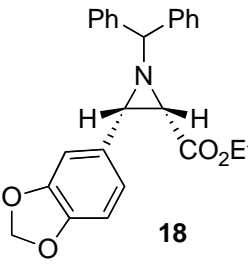
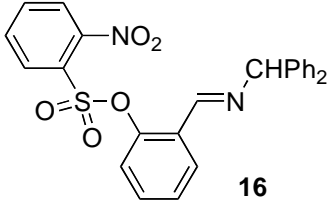
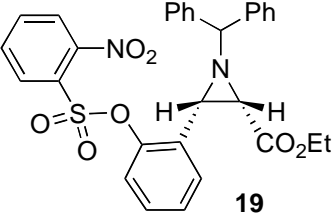
Entry	Ligand	Solvent	13 [Yield %] ^b	<i>dr</i> [<i>cis:trans</i>] ^c	<i>ee</i> [%] ^d
1	9	Toluene	66	99:1	15
2	10	Toluene	69	99:1	28
3	11	Toluene	69	99:1	17 ^e
4	11	Toluene	50	99:1	10 ^f
5	8	Toluene	63	99:1	58
6	6	Toluene	68	99:1	69
7	7	DCM	65	99:1	58 ^g
8	7	DCM	67	99:1	64 ^h
9	7	Toluene/DCM	80	99:1	72 ⁱ
10	7	CF ₃ C ₆ H ₅ /DCM	59	99:1	70 ^j
11	7	Toluene/Et ₂ O	76	99:1	80 ^k
12	7	Toluene	61	99:1	62 ^l
13	7	Toluene	81	99:1	63 ^m
14	7	Toluene	76	99:1	78 ⁿ

^aUnless stated otherwise, 0.051 mmol of chiral ligand, 23 °C, 1.1 equiv of ethyl diazoacetate vs imine **12**, and 18 h reaction time were used; imine concentration was maintained at 0.5 M for entries 1-8 and 12-16, and at 0.22 M for entries 9-11. ^bIsolated yields after chromatography on SiO₂. ^cDetermined by ¹H NMR analysis of the crude reaction mixture. ^dDetermined by HPLC analysis (Chiralcel OD); the major enantiomer was assigned in accordance to ref.^{10b} ^eReaction time of 2 h. ^fReaction time of 18 h. ^gReaction time of 8 h. ^hReaction time of 18 h. ^{i,j,k}Reaction performed in a 1:1 mixture of solvents. ^lReaction time of 5 h at 0 °C. ^mReaction temperature of 0 °C for 3 h, then room temperature for 15 h. ⁿThis reaction was repeated several times, in order to ensure reproducibility, with identical *dr* and *ee* ranging from 74-78%.

For a further exploration of the utility of the catalyst derived from ligand **7**, benzhydryl imines **1** and **14-16** derived from benzaldehyde, 4-nitrobenzaldehyde, piperonal and 2-nitrobenzenesulfonic acid 2-formylphenyl ester, respectively, were used as substrates (Table 2).

Aziridines **13**, *ent*-**4**, **17**,¹¹ **18**,¹⁸ and **19**¹⁹ were obtained in moderate yields and enantiomeric purities (Table 2, entries 2-5). In contrast, the diastereoselectivity of these insertions was excellent with the exception of benzhydryl imine **16**, which resulted in a surprisingly low 60:40 *cis:trans* isomeric ratio.

Table 2. Catalytic asymmetric aziridination of benzhydryl imines in the presence of ligand **7**

Entry	Imine	Aziridine, Yield [%]	<i>dr</i> [<i>cis:trans</i>]	<i>ee</i> [%]
1	12	13 , 76	99:1	78
2	1	<i>ent</i> - 4 , 55 ^a	99:1	54
3			99:1	55
4			99:1	55
5			60:40	55

^aImine concentration was 0.22 M in toluene, and yield based on recovered imine was 72%.

^bReaction was performed in a toluene/Et₂O solution (1:1), with the imine (0.18 M) added by syringe pump over 3 h to the reaction mixture at -40 °C, then stirred at room temperature for 15 h.

In conclusion, introduction of bulky arene substituents, in particular the 2-phenylnaphthalene moiety, into the 3- and 3'-positions of the binaphthol scaffold leads to a significant improvement of the level of chiral induction in the aziridination of simple benzhydryl imines.

Acknowledgment

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17. Typical procedure: To a flame-dried Schlenk flask cooled under argon was added a solution of ligand (0.051 mmol) and triphenylborate (0.15 mmol) in methylene chloride (2.0 mL). The reaction mixture was heated at 55 °C for 1 h under an atmosphere of argon. Vacuum was then applied for 30 min while heating was continued. The dried catalyst was cooled to room temperature under argon and dissolved in toluene (430 μ L). The solution was transferred via syringe to a flame-dried flask under argon at room temperature. A solution of imine (0.51 mmol) in toluene (430 μ L) was added by syringe to the catalyst solution, and the reaction mixture was stirred for 10 min, treated with ethyl diazoacetate (59 μ L), and stirred at room temperature for 18 h. The mixture was then diluted with methylene chloride and concentrated *in vacuo* to a lime-green solid. Chromatography on SiO₂ (hexanes/CH₂Cl₂ 2:1.5) gave a white solid. (2*S*,3*S*)-Ethyl 1-benzhydryl-3-(4-bromophenyl)aziridine-2-carboxylate (**13**) was obtained in 76% and in 78% *ee* in a *cis:trans* ratio of 99:1: Chiral HPLC (Chiralcel OD); t_r = 6.29 min (major isomer) and t_r = 10.63 min (minor isomer); ¹H NMR δ 7.58 (d, 2 H, *J* = 7.1 Hz), 7.46 (d, 2 H, *J* = 7.0 Hz), 7.19 - 7.40 (m, 10 H), 3.95 (s, 1 H), 3.96 (q, 2 H, *J* = 7.1 Hz), 3.15 (d, 1 H, *J* = 6.8 Hz), 2.69 (d, 1 H, *J* = 6.8 Hz), 1.03 (t, 3 H, *J* = 7.1 Hz).
18. Chiral HPLC (Chiralcel OD); t_r = 7.75 min (major isomer) and t_r = 15.11 (minor isomer); Mp 119-121 °C (Hexanes/CH₂Cl₂); $[\alpha]_D$ -13.3 (*c* 2.1, CH₂Cl₂); ¹H NMR δ 7.63 (d, 2 H, *J* = 7.6 Hz), 7.51 (d, 2 H, *J* = 7.6 Hz), 7.21-7.39 (m, 5 H), 6.98 (s, 1 H), 6.89 (d, 1 H, *J* = 8.0 Hz), 6.72 (d, 1 H, *J* = 8.0 Hz), 5.90 (s, 2 H), 4.03 (q, 2 H, *J* = 7.1 Hz), 3.96 (s, 1 H), 3.18 (d, 1 H, *J* = 6.8 Hz), 2.66 (d, 1 H, *J* = 6.8 Hz), 1.10 (t, 3 H, *J* = 7.1 Hz); ¹³C NMR δ 167.7, 147.1, 146.8, 142.4, 142.3, 128.9, 128.4, 127.5, 127.4, 127.2, 121.1, 108.3, 107.6, 100.8, 77.6, 60.5, 47.9, 46.4, 14.0; MS (EI) (*rel intensity*) 401 (M⁺, 1.8), 234 (100), 161 (98), 152 (28), 76 (33); HRMS (EI) calcd for C₂₅H₂₃NO₄ 401.1627, found 401.1641.
19. Chiral HPLC ((Chiralcel OD; hexanes/2-propanol = 7.5:2.5) t_r = 10.6 (major isomer), t_r = 12.61 (minor isomer); Mp 55-57 °C; $[\alpha]_D$ -7.9 (*c* 2.4, CH₂Cl₂); ¹H NMR δ 7.98 (dd, 1 H, *J* = 1.1, 7.9 Hz), 7.63-7.83 (m, 3 H), 7.55 (d, 2H, *J* = 7.1 Hz), 7.44 (d, 2 H, *J* = 6.9 Hz), 7.15-

7.38 (m, 8 H), 7.02 (dd, 1 H, $J = 1.1, 8.0$ Hz), 3.97 (s, 1 H), 3.96 (q, 2 H, $J = 7.0$ Hz), 3.49 (d, 1 H, $J = 6.7$ Hz), 2.70 (d, 1 H, $J = 6.7$ Hz), 1.02 (t, 3 H, $J = 7.1$ Hz); ^{13}C NMR δ 167.5, 148.6, 148.0, 142.3, 142.2, 135.3, 132.0, 131.8, 131.0, 129.1, 128.9, 128.5, 128.3, 127.7, 127.6, 127.2, 127.1, 127.0, 124.8, 121.3, 77.6, 60.6, 46.2, 43.7, 13.9; MS (EI) (*rel intensity*) 557 ($[\text{M}-1]^+$, 0.3), 485 ($[\text{M}-\text{CO}_2\text{Et}]^+$ 1.3), 391 (93), 186 (57), 167 (100), 132 (76), 121 (91); HRMS (EI) calcd for $\text{C}_{27}\text{H}_{21}\text{N}_2\text{O}_5\text{S}_1$ (M-CO₂Et) 485.1171, found 485.1163.